



FASEM 2019

# Neutron and X-ray imaging

Nikolay Kardjilov

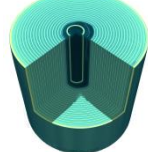
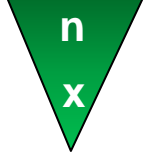
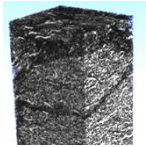
# Introduction



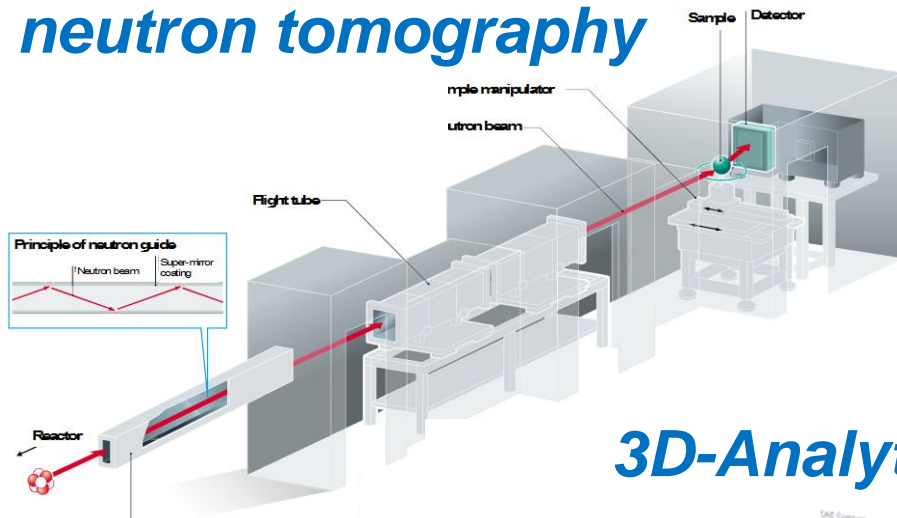
# Introduction

Institute of Applied Materials

Neutron Imaging Micro CT Synchrotron



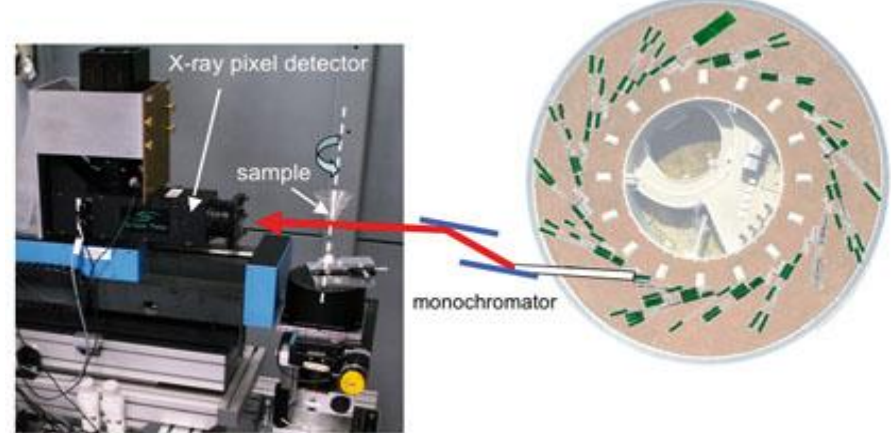
## CONRAD-2 neutron tomography



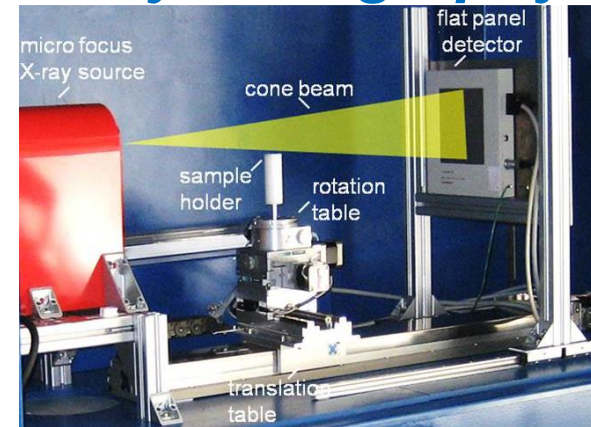
## 3D-Analytics Lab



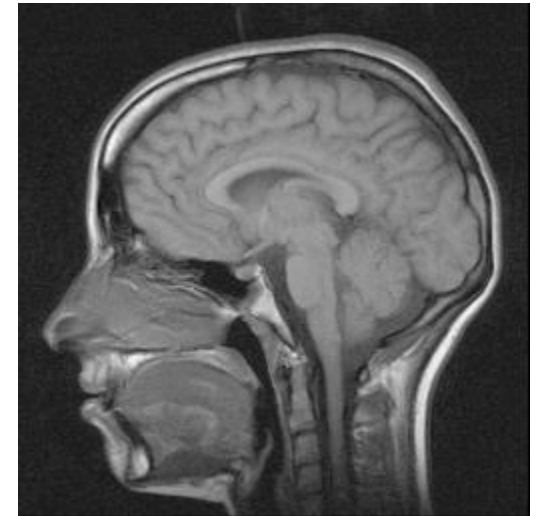
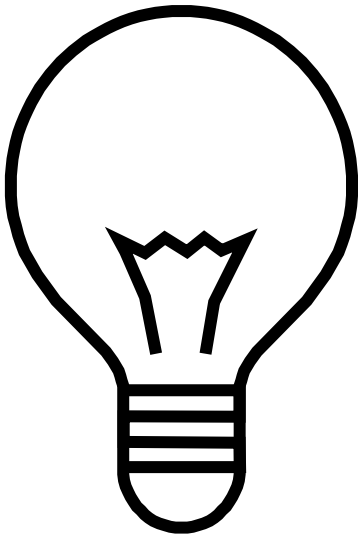
## BAM-Line @ BESSY Synchrotron tomography

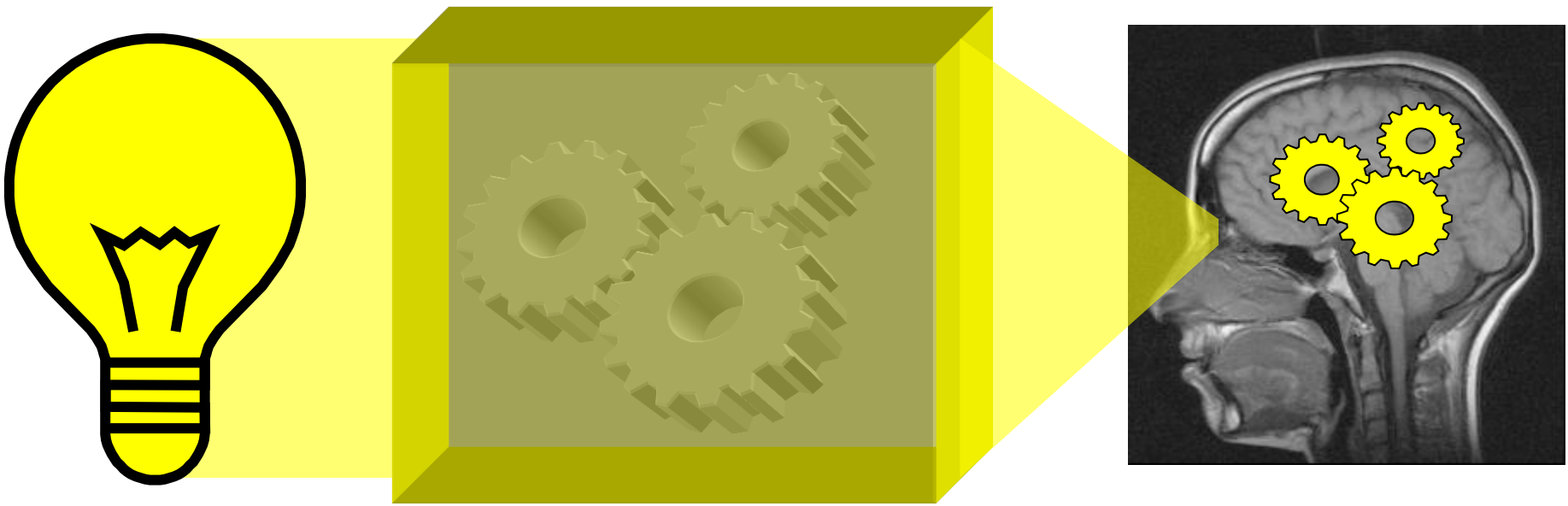


## MicroCT Lab X-ray tomography





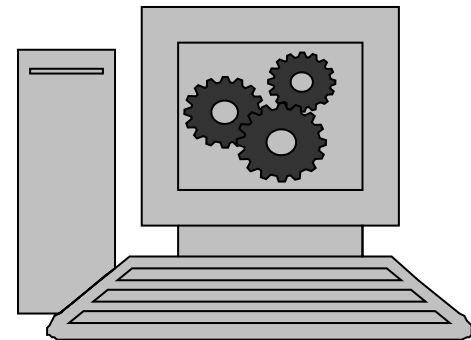
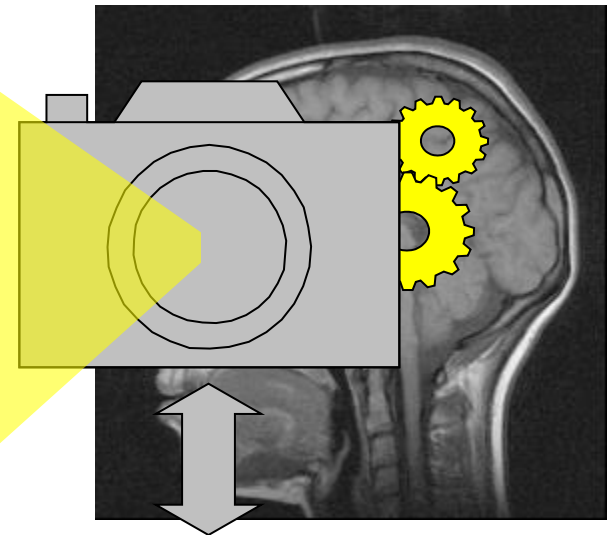
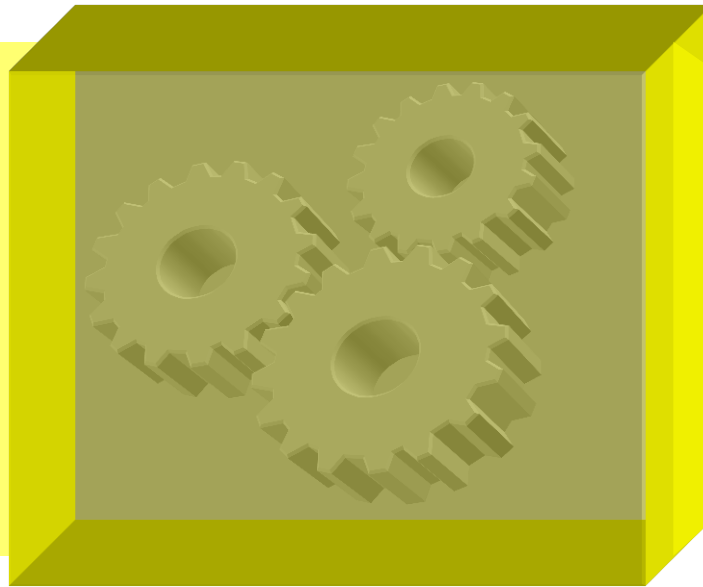
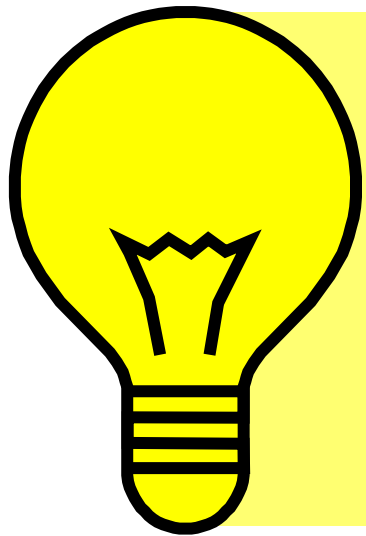




# Source

# Sample

# Detector

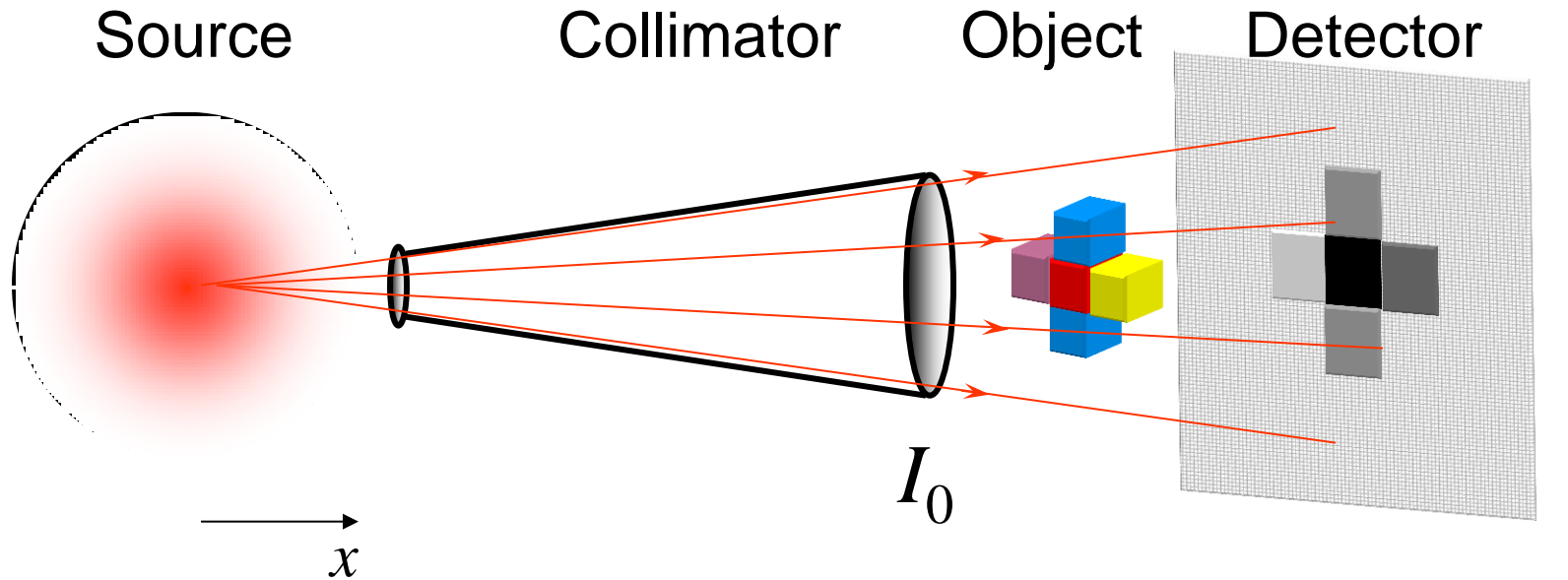


# Contrast

- **Beam optimisation**
- Detector development
- Interaction with matter
  - X-rays
  - Neutrons

# Resolution

# Beam optimisation



$$\sim I_0 e^{-\int \Sigma(x) dx}$$

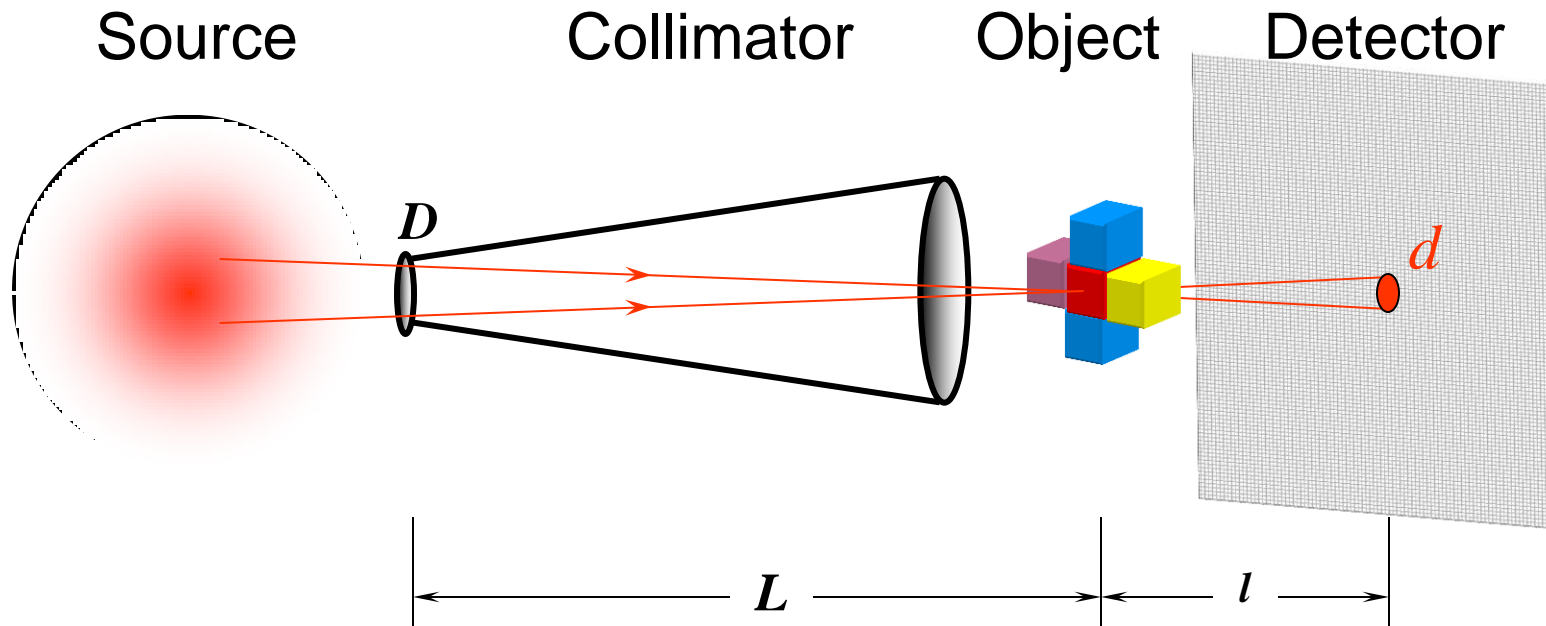
$x$  – propagation direction

$I_0$  – primary beam

$\Sigma(x)$  – attenuation coefficient



# Beam optimisation

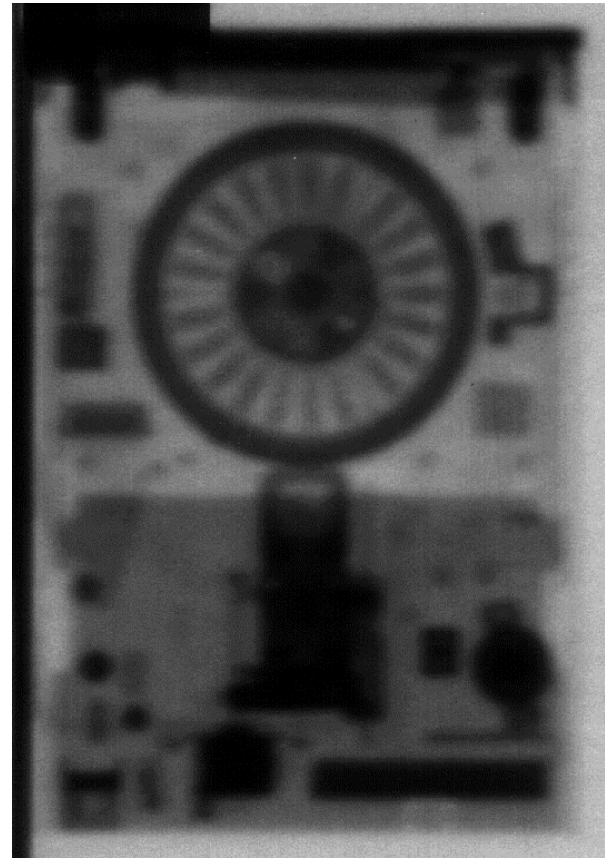
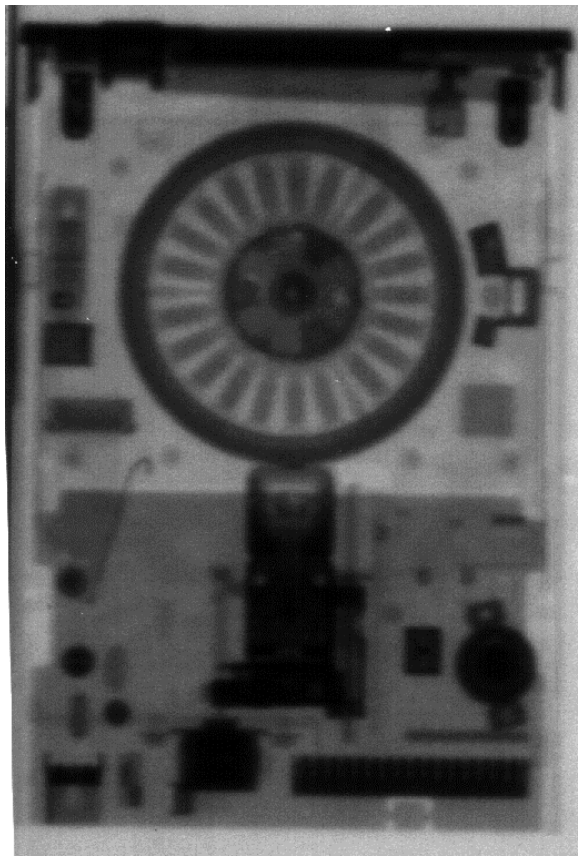
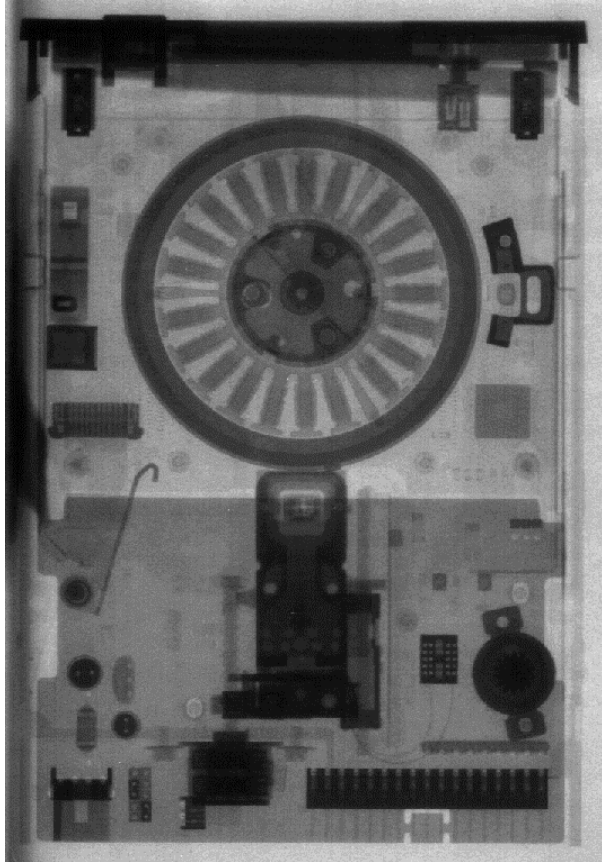


$D$  – Collimator aperture

$L$  – Distance Collimator-Object

$l$  – Distance Object-Detector

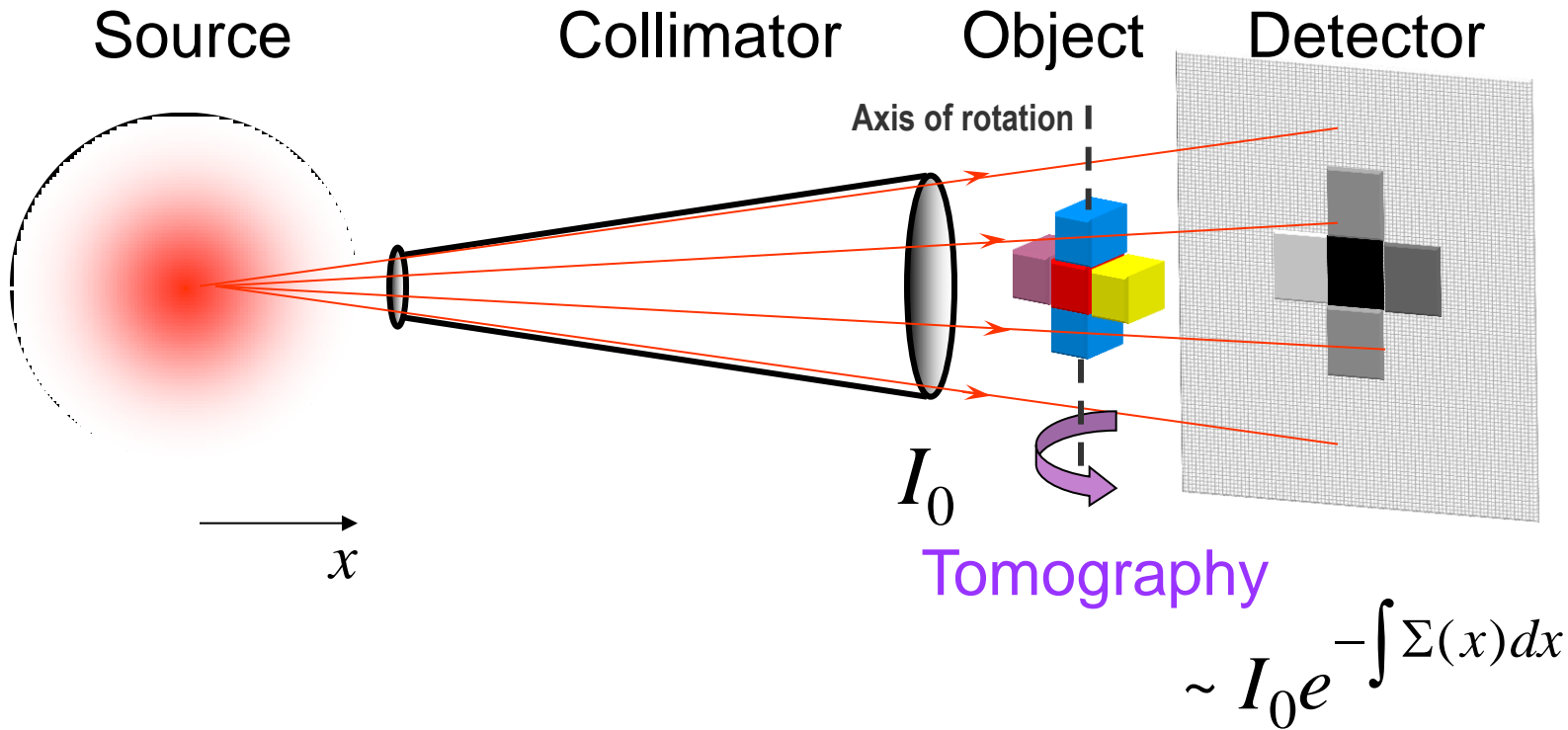
$$d = \frac{l}{L/D}$$



Radiographs of a 3,5" floppy drive in 0 cm , 10 cm and 20 cm distance from a film + Gd sandwich taken at a cold neutron guide with  $L/D=71$ .

The measurements are performed at FRM-I, TU-München by B. Schillinger

# Absorption tomography



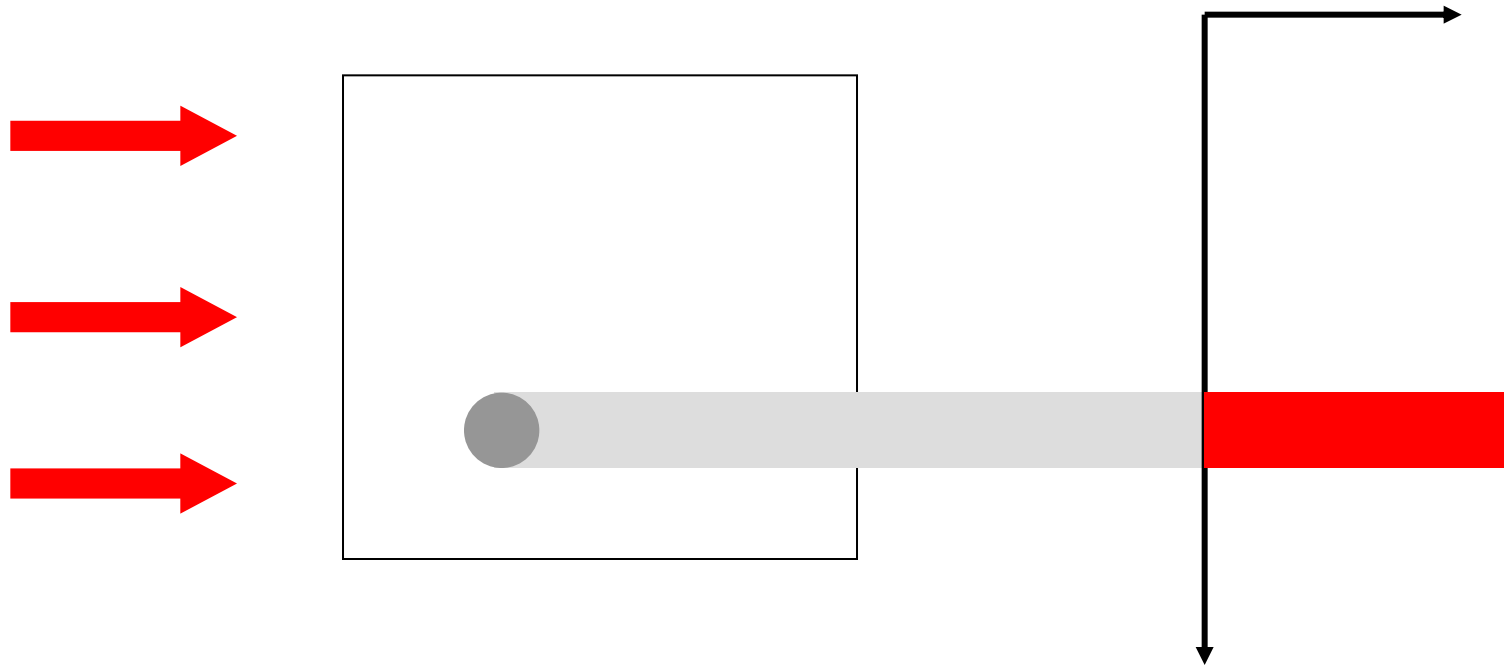
$x$  – propagation direction

$I_0$  – primary beam

$\Sigma(x)$  – attenuation coefficient

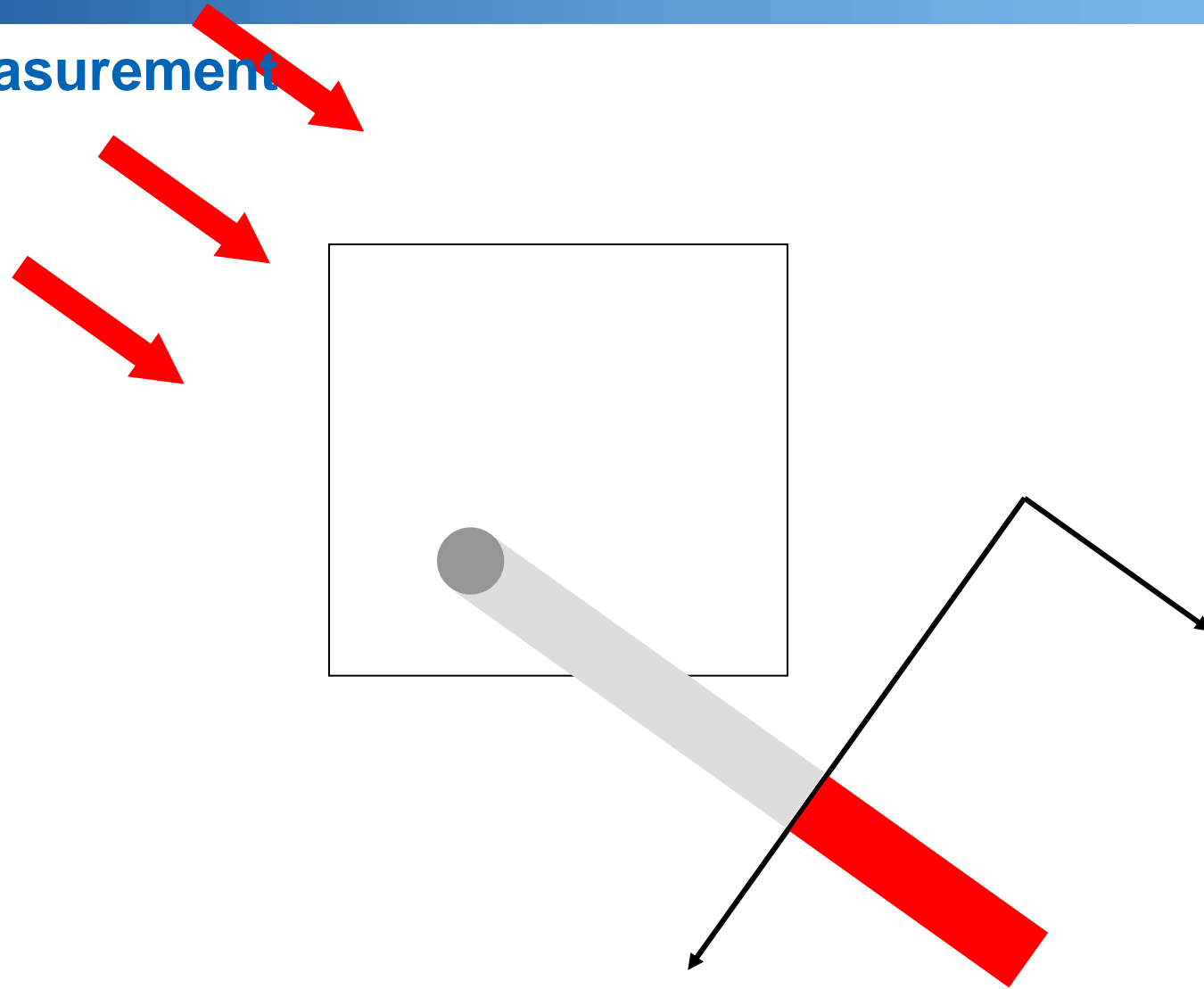
# Principle of tomographic reconstruction

## Measurement



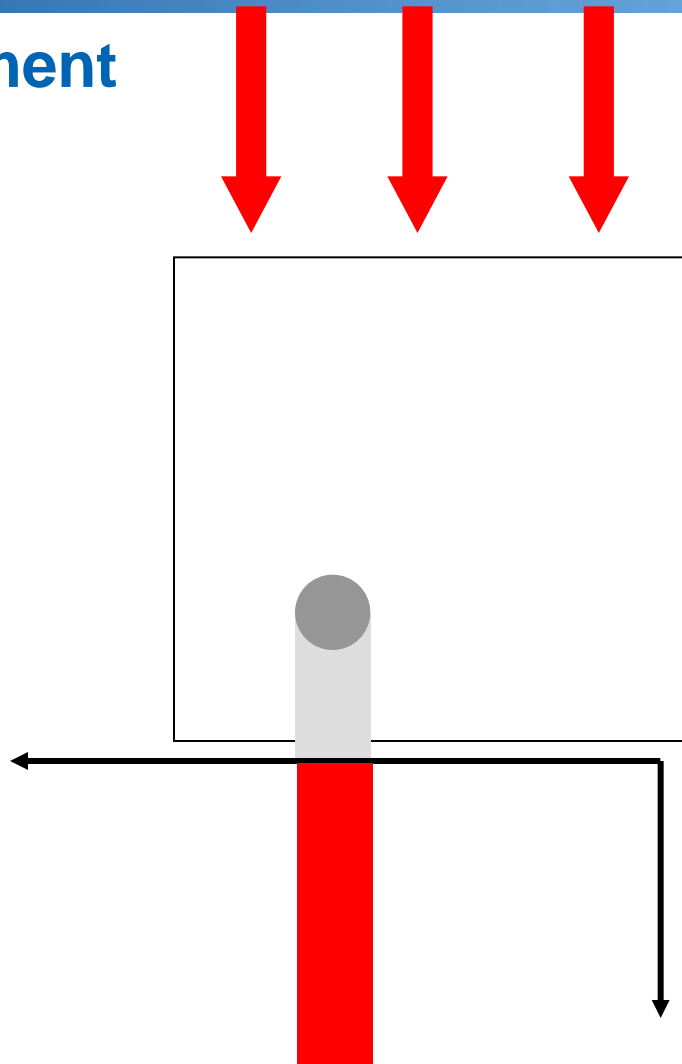
# Principle of tomographic reconstruction

Measurement



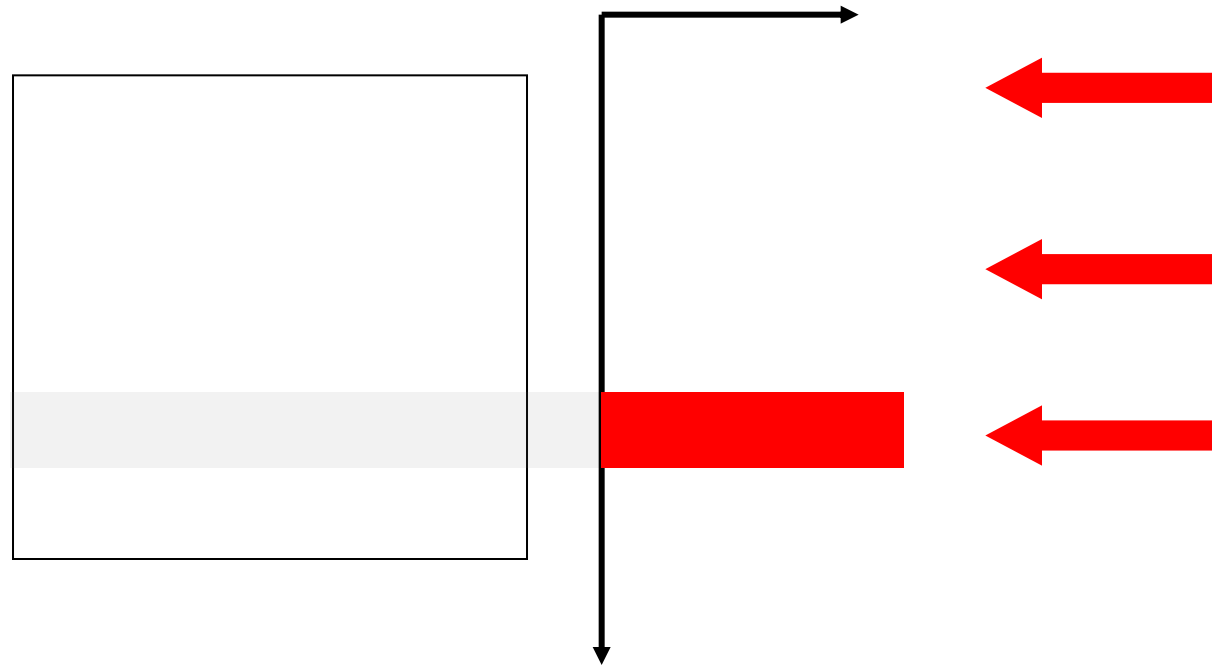
# Principle of tomographic reconstruction

Measurement



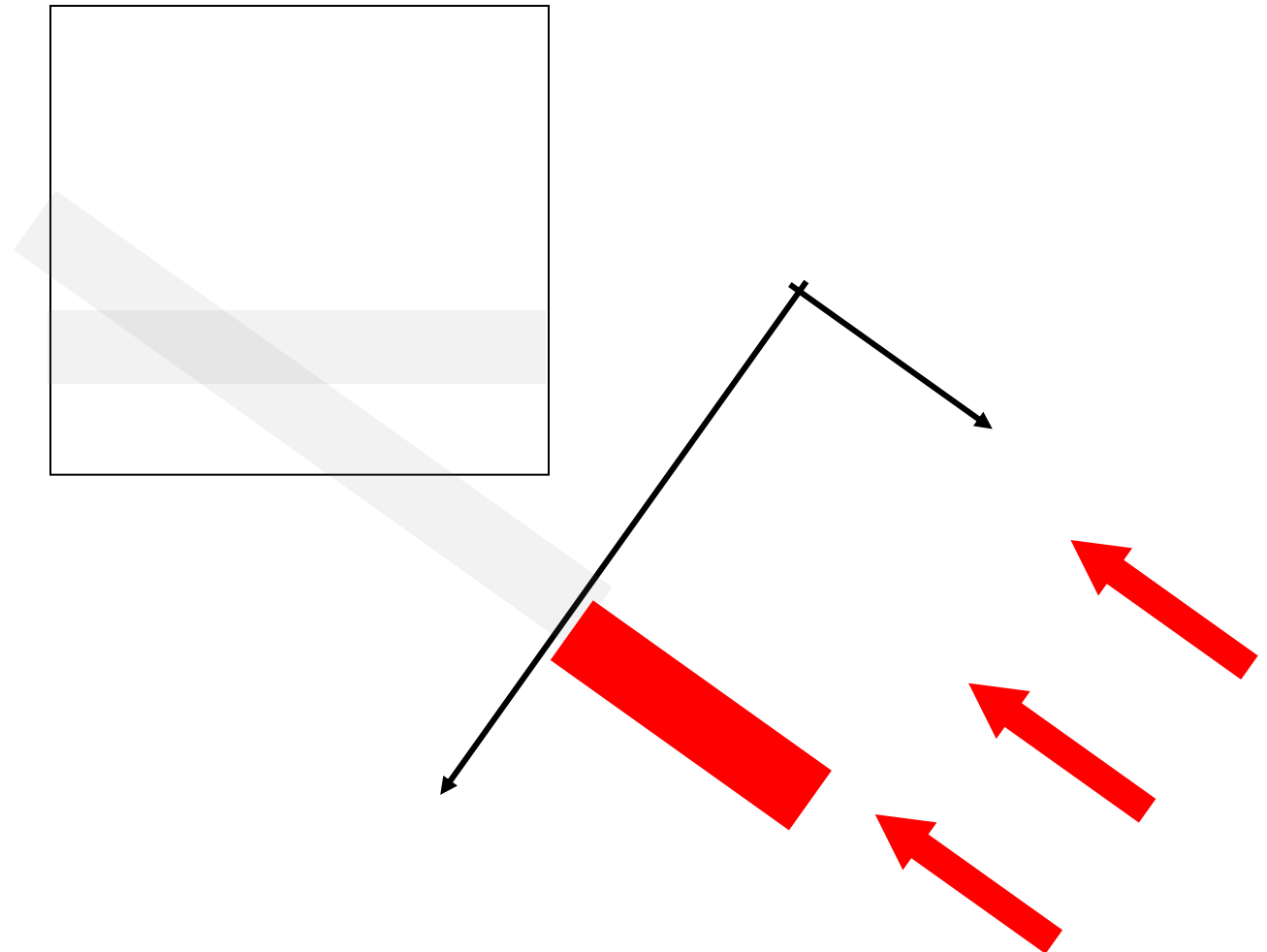


## Back projection

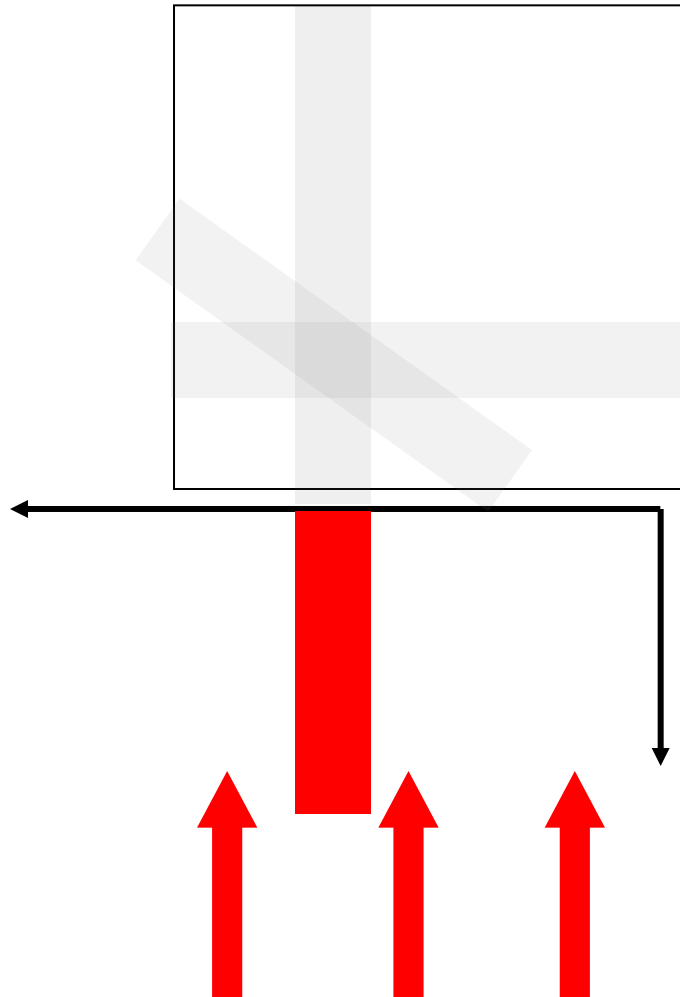


# Principle of tomographic reconstruction

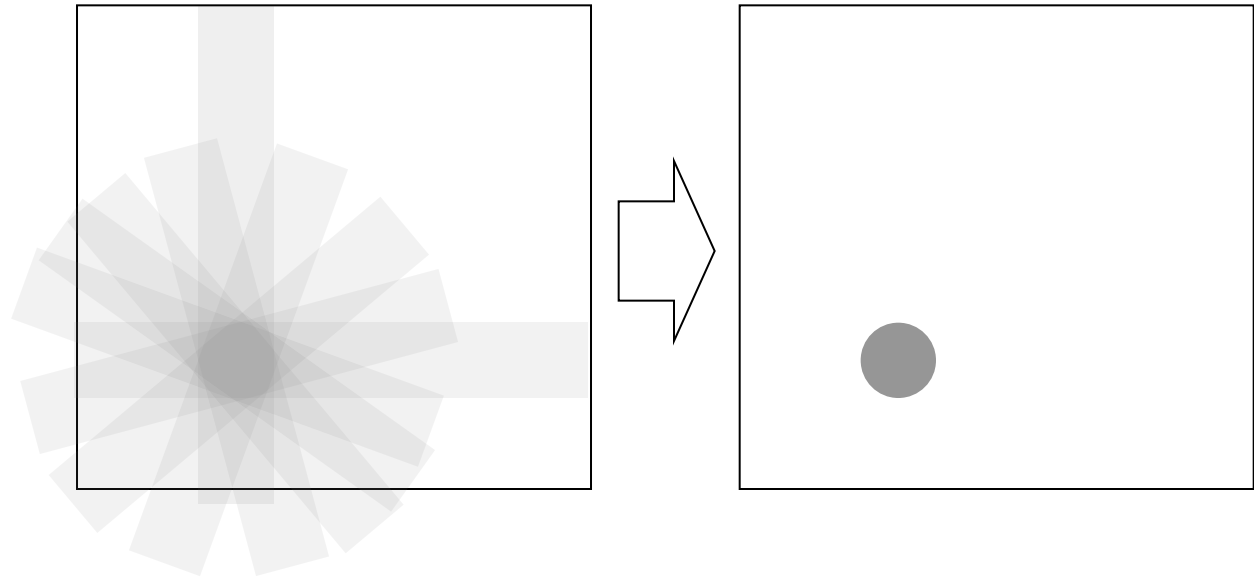
## Back projection



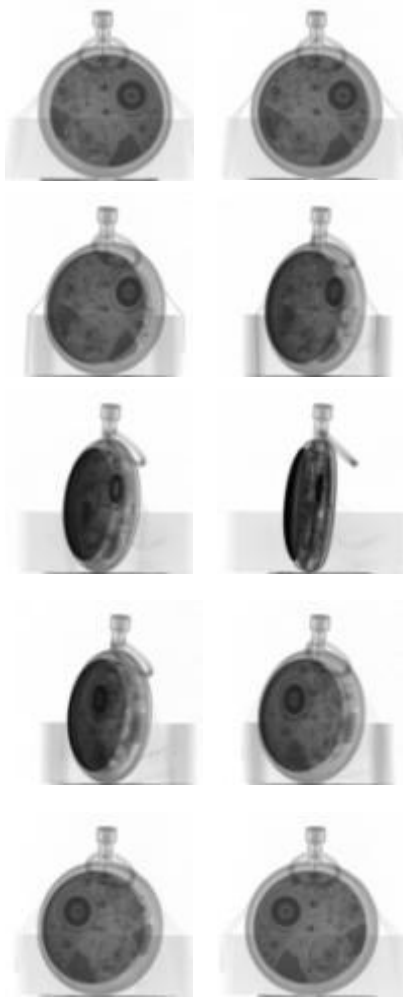
## Back projection



## Back projection



# Absorption tomography



# Resolution

- Beam optimisation
- Detector development

# Contrast

- Interaction with matter
  - X-rays (sources, interaction, detectors)
  - Neutrons



First experiments with a new kind of radiation were performed by **Konrad Röntgen** in **1895** during investigations with cathode-ray tubes.

He found the new ray could pass through most substances casting shadows of solid objects.

In conjunction with a photographic plate, a picture of interior body parts can be obtained when human tissue will be investigated.



# Neutron imaging

## Introduction



One of the first experiments late in 1895 was a film of a hand of his wife.

The bones and also finger rings deliver much higher contrast than the soft tissue.

# Introduction



**Some exotic applications of X-ray transmission in the earlier period of use.**

<http://www.orau.org/ptp/collection/shoefittingfluor/shoe.htm>

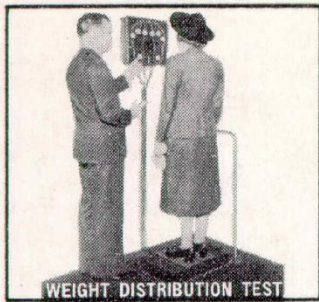
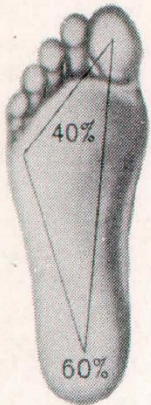


## CERTIFICATE

SHOE-FITTING TEST DATA FOR \_\_\_\_\_

1. ANKLE ROLL    GOOD     FAIR     POOR

2. WEIGHT DISTRIBUTION



| LEFT   |       | RIGHT  |
|--------|-------|--------|
| _____% | BALL  | _____% |
| _____% | OUTER | _____% |
| _____% | HEEL  | _____% |



WRONG WAY

3. X-RAY FITTING TEST



RIGHT WAY



| LEFT                     |      | RIGHT                    |
|--------------------------|------|--------------------------|
| <input type="checkbox"/> | GOOD | <input type="checkbox"/> |
| <input type="checkbox"/> | FAIR | <input type="checkbox"/> |
| <input type="checkbox"/> | POOR | <input type="checkbox"/> |



WRONG WAY

**This scientific way of approaching the problem of poorly-fitted shoes eliminates guesswork. Now you can see for yourself!**

# Neutron imaging

## Introduction

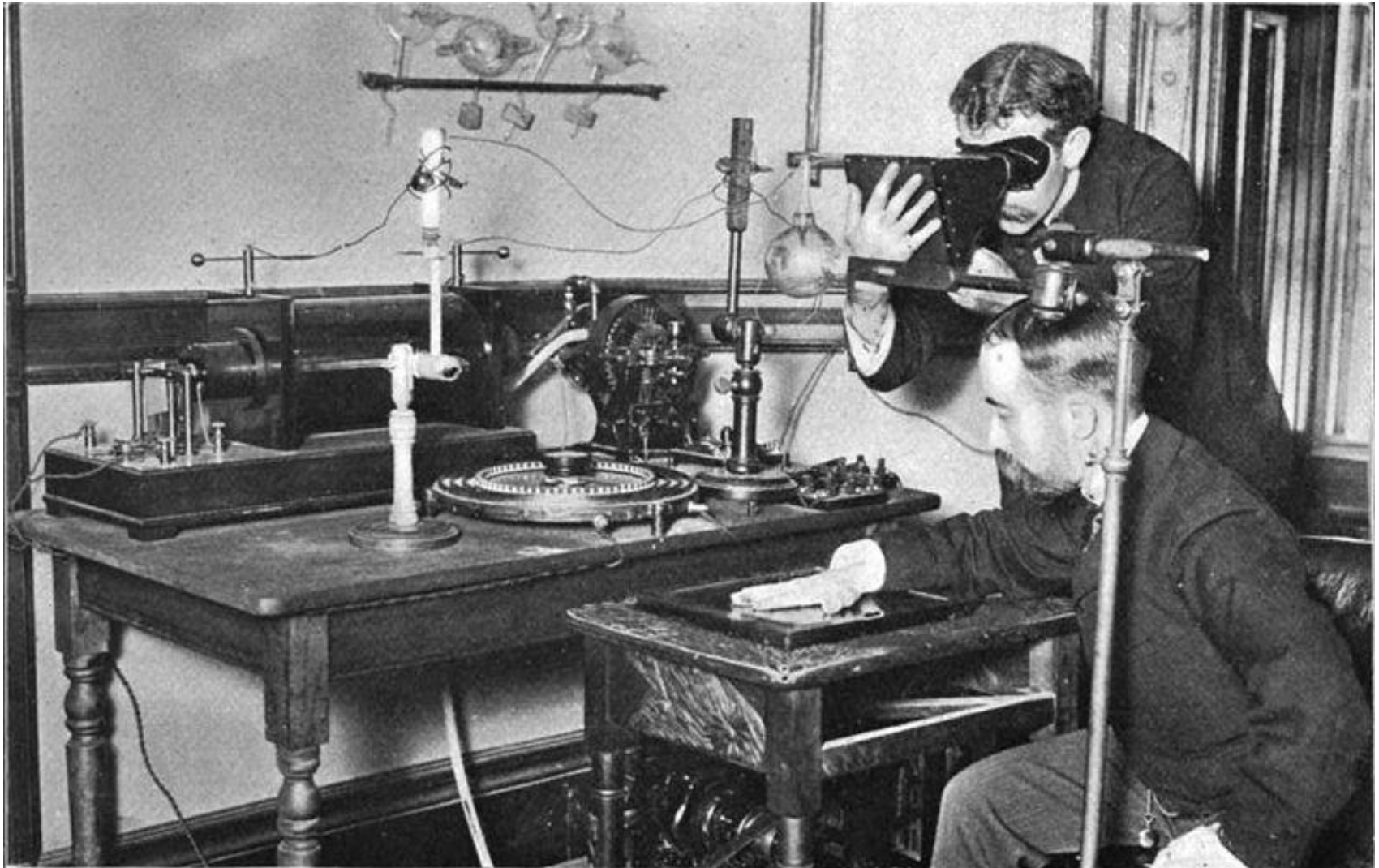
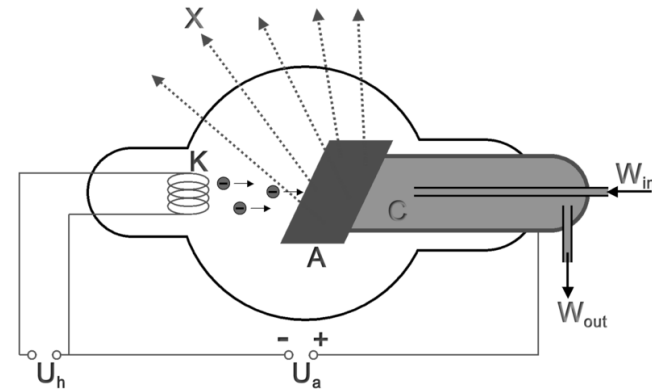
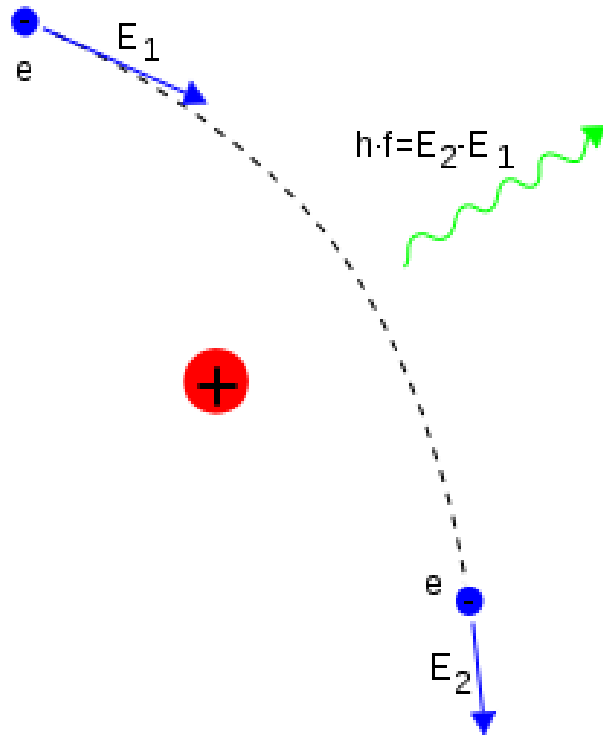


Photo of experimenters taking an X-ray with an early Crookes tube apparatus, from the late 1800s.

## Bremsstrahlung



$$E_{\text{Photon}} = \hbar\omega = E_{\text{kinetisch}} = eU$$

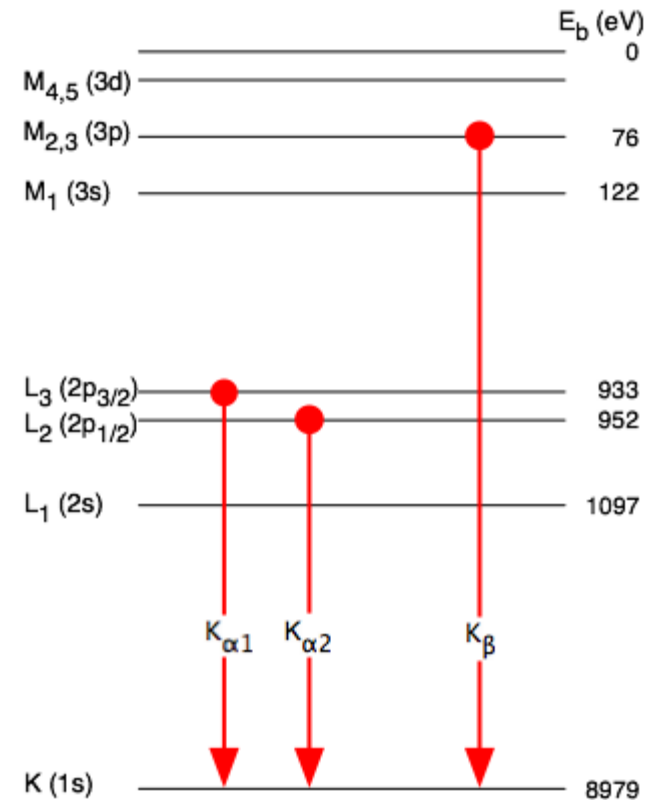
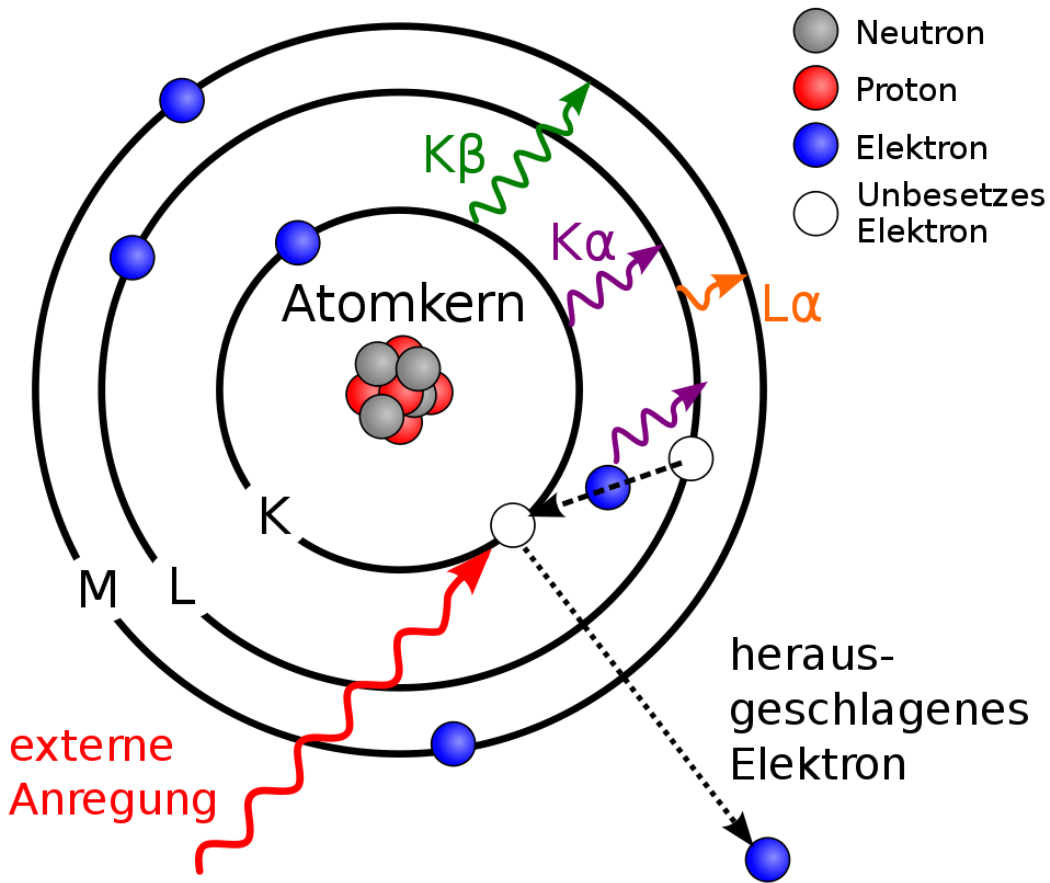
$$\lambda_{\min} = \frac{h \cdot c}{e \cdot U} \quad \lambda_{\min} = \frac{1,24 \cdot 10^{-6} \text{ V} \cdot \text{m}}{U}$$

$$J(\lambda) = K \cdot I \cdot Z \cdot \left( \frac{\lambda}{\lambda_{\min}} - 1 \right) \cdot \frac{1}{\lambda^2}$$

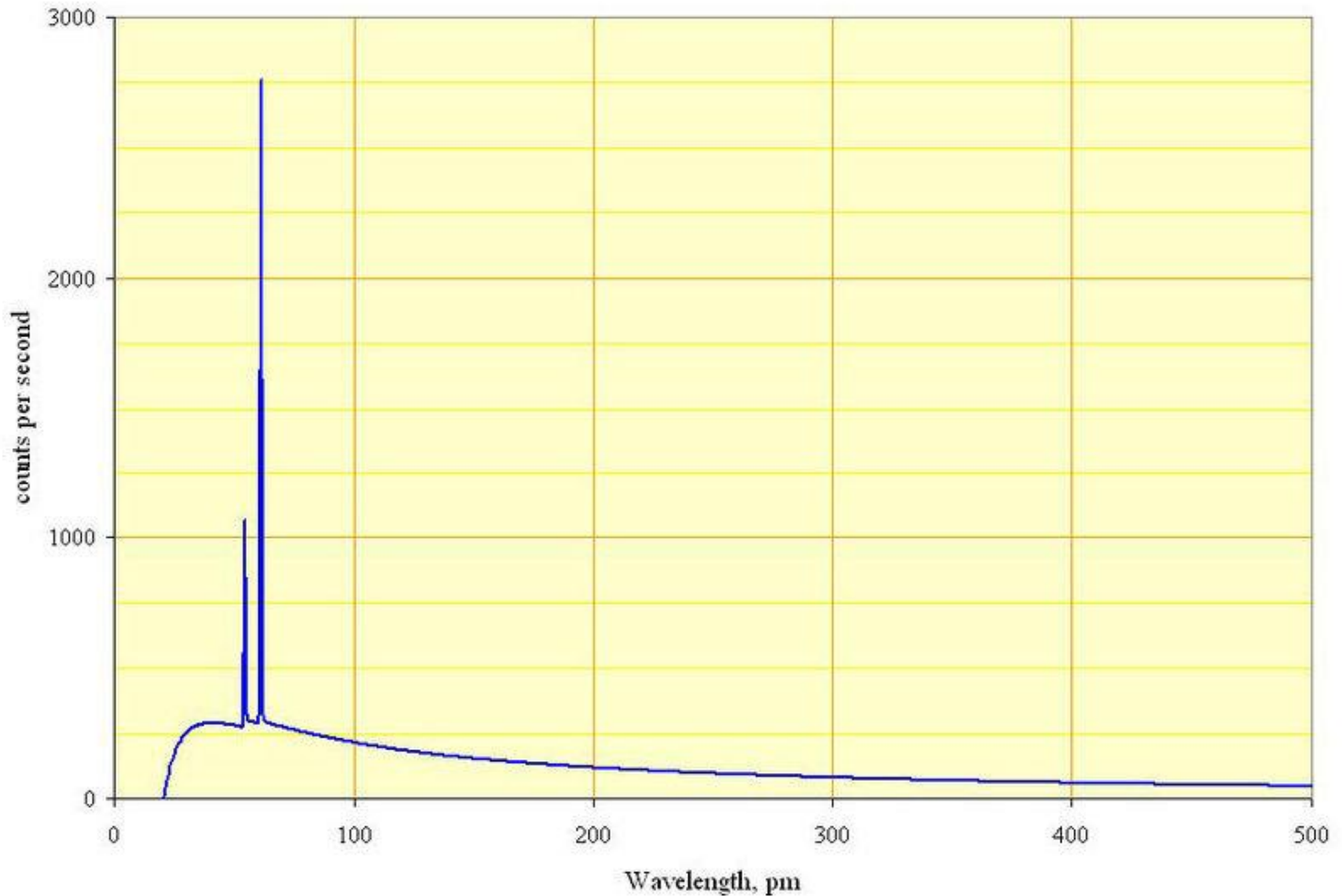
The maximum energy of the produced X-ray photon is limited by the energy of the incident electron, which is equal to the voltage on the tube, so an 80 kV tube cannot create X-rays with an energy greater than 80 keV.



## Characteristic radiation



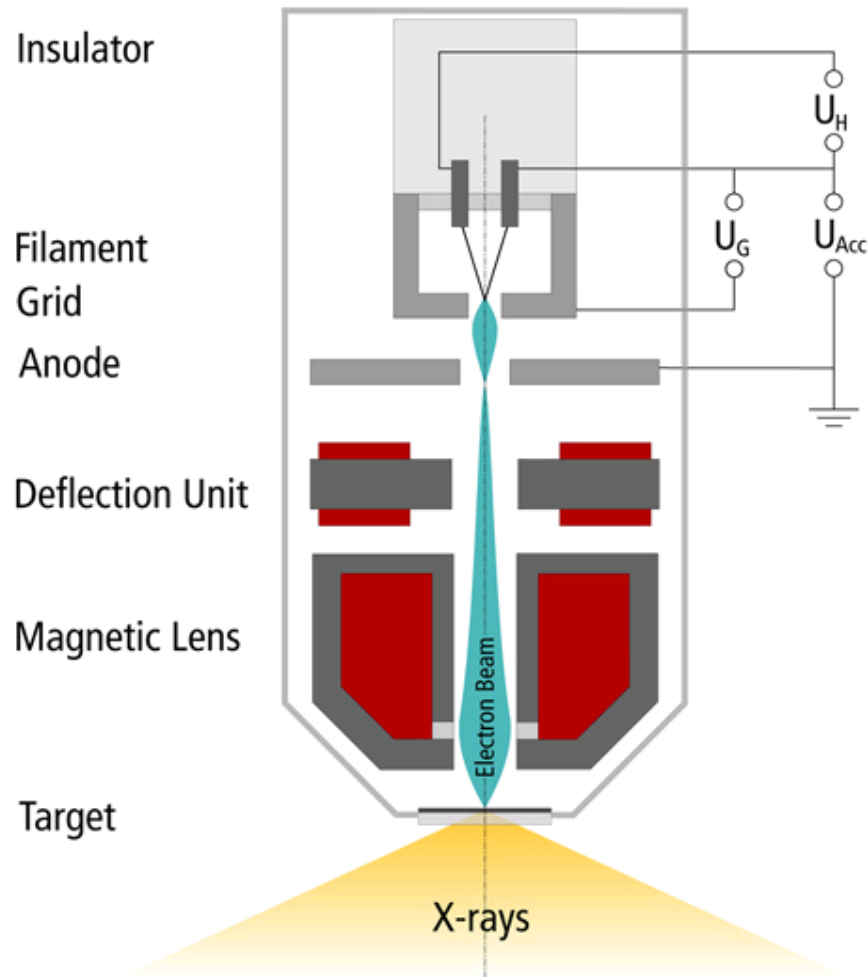
## Spectrum



Spectrum of the X-rays emitted by an X-ray tube with a rhodium target, operated at 60 kV.

<http://upload.wikimedia.org/wikipedia/commons/5/5c/TubeSpectrum.jpg>

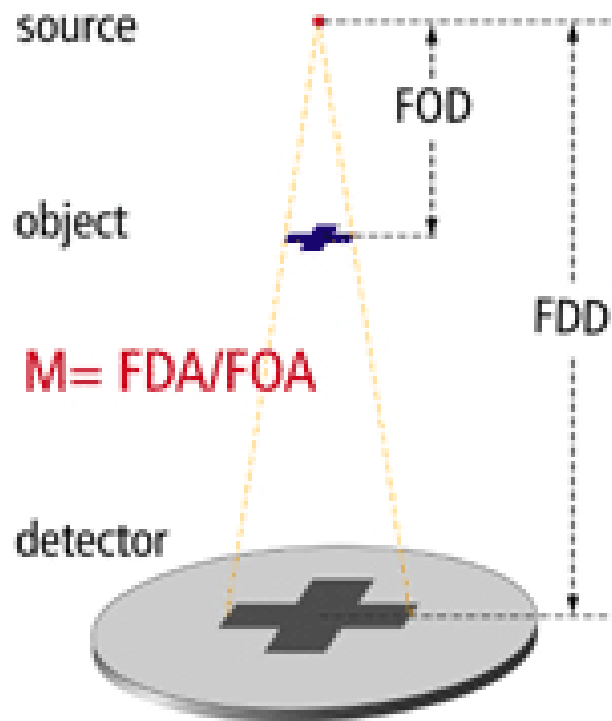
## Microfocus tubes



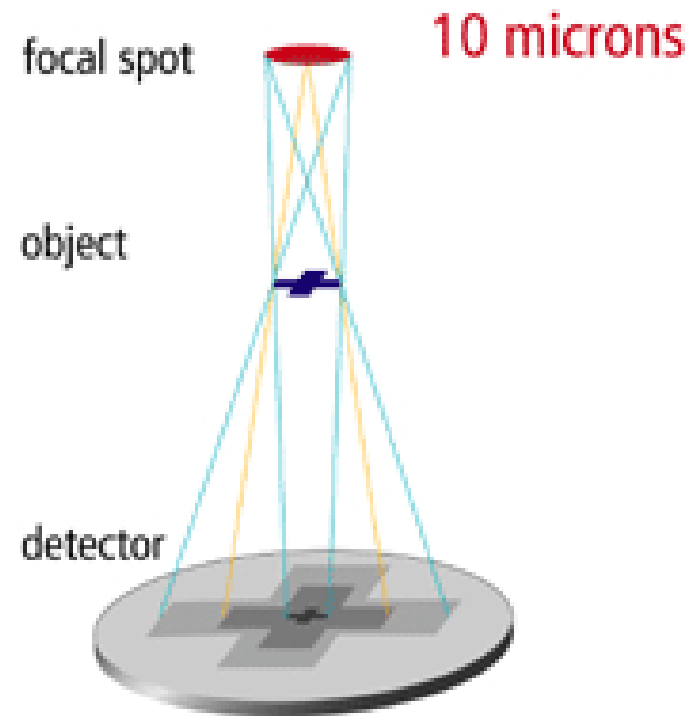
[http://www.microfocus-x-ray.com/en/company/technology/principles\\_of\\_operation/principle\\_025.html](http://www.microfocus-x-ray.com/en/company/technology/principles_of_operation/principle_025.html)

## Microfocus tubes

### Magnification



### Resolution



[http://www.microfocus-x-ray.com/en/company/technology/principles\\_of\\_operation/principle\\_025.html](http://www.microfocus-x-ray.com/en/company/technology/principles_of_operation/principle_025.html)

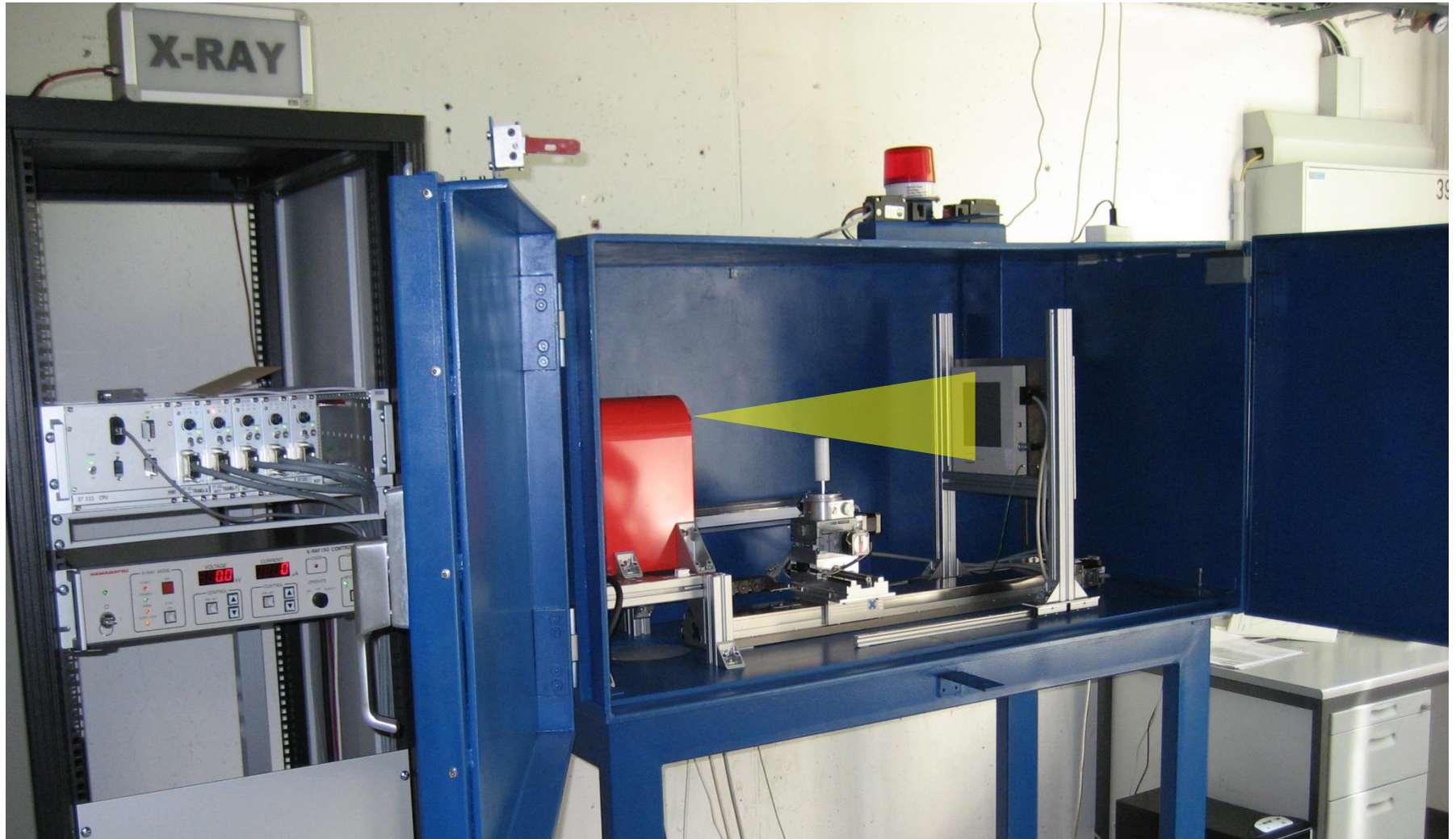
## 150 keV X-ray tube



| Sealed Type                     |                               |
|---------------------------------|-------------------------------|
| 150 kV                          |                               |
| L8121-03                        |                               |
| 40 to 150                       |                               |
| Small Spot Mode                 | 0 to 250<br>(10 W Max.)       |
| Middle Spot Mode                | 0 to 500<br>(30 W Max.)       |
| Large Spot Mode                 | 0 to 500                      |
| Small Spot Mode                 | 10                            |
| Middle Spot Mode                | 30                            |
| Large Spot Mode                 | 75                            |
| Small Spot Mode                 | 7<br>(5 $\mu\text{m}$ at 4 W) |
| Middle Spot Mode                | 20                            |
| Large Spot Mode                 | 50                            |
| 43                              |                               |
| 25 degrees (ReflectionType)     |                               |
| 17                              |                               |
| 20                              |                               |
| Continuous                      |                               |
| CE (IEC61326-1)<br>(IEC61010-1) |                               |

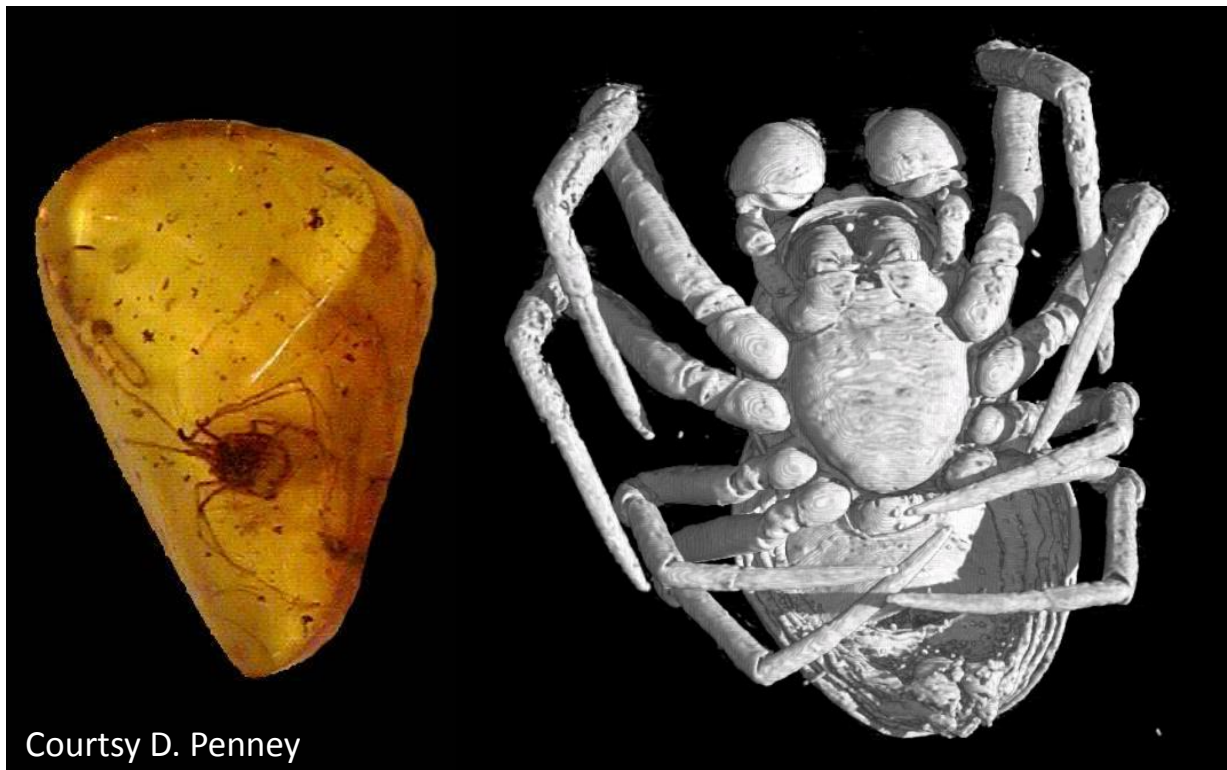
# X-ray sources

## $\mu$ -CT scanner





- Amber fossils : optically obscured or distorted



<http://www.inct.be/en/home>





Sample image: X-ray showing frontal view of both hands.

# Resolution

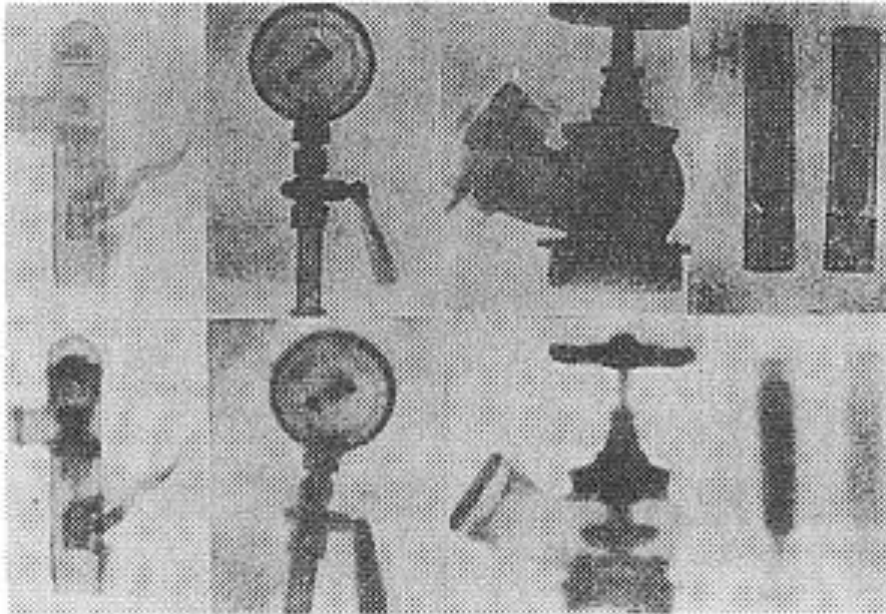
- Beam optimisation
- Detector development

# Contrast

- Interaction with matter
  - X-rays
  - **Neutrons (sources, interaction, detectors)**

# Roots of neutron radiography

Taken from C.O. Fischer's article in WCNR-4



Berlin, 1935 – 1938

*H. Kallmann & Kuhn* with Ra-Be  
and neutron generator

Berlin until Dec. 1944

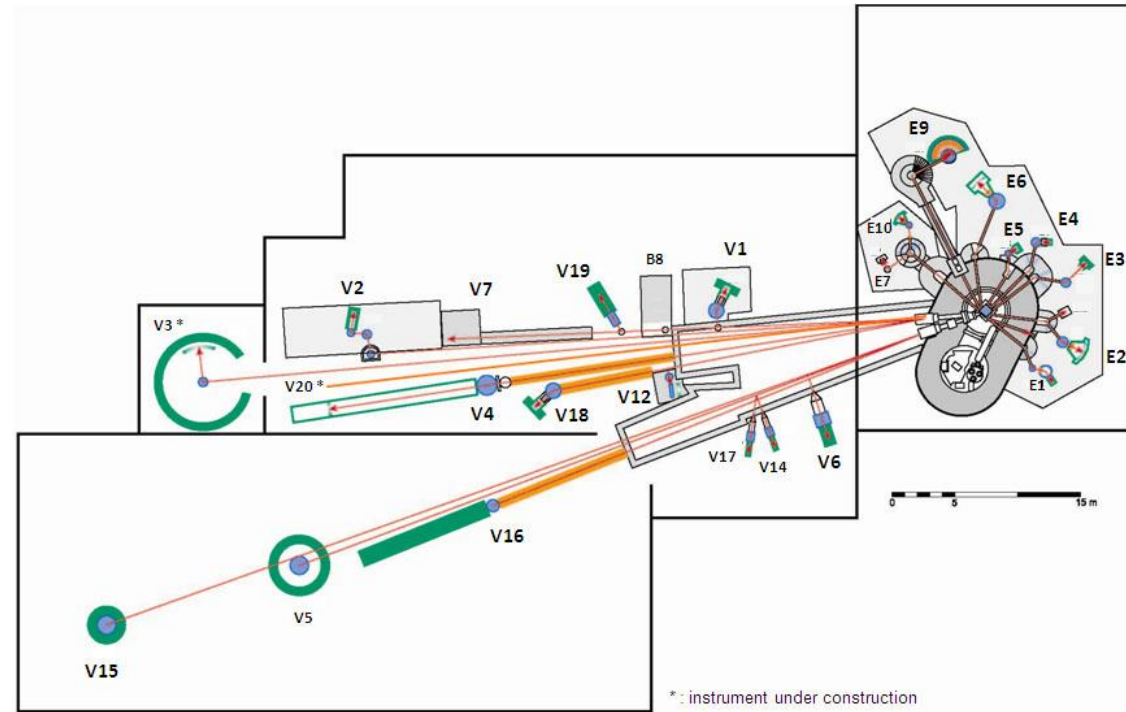
*O. Peter* with an  
accelerator neutron source

As typical: valves, manometers, injectors

*But the real programs with neutrons started after World War II at research reactors*

# Neutron sources

## BER-2



**Type:** open, light-water-moderated swimming pool reactor

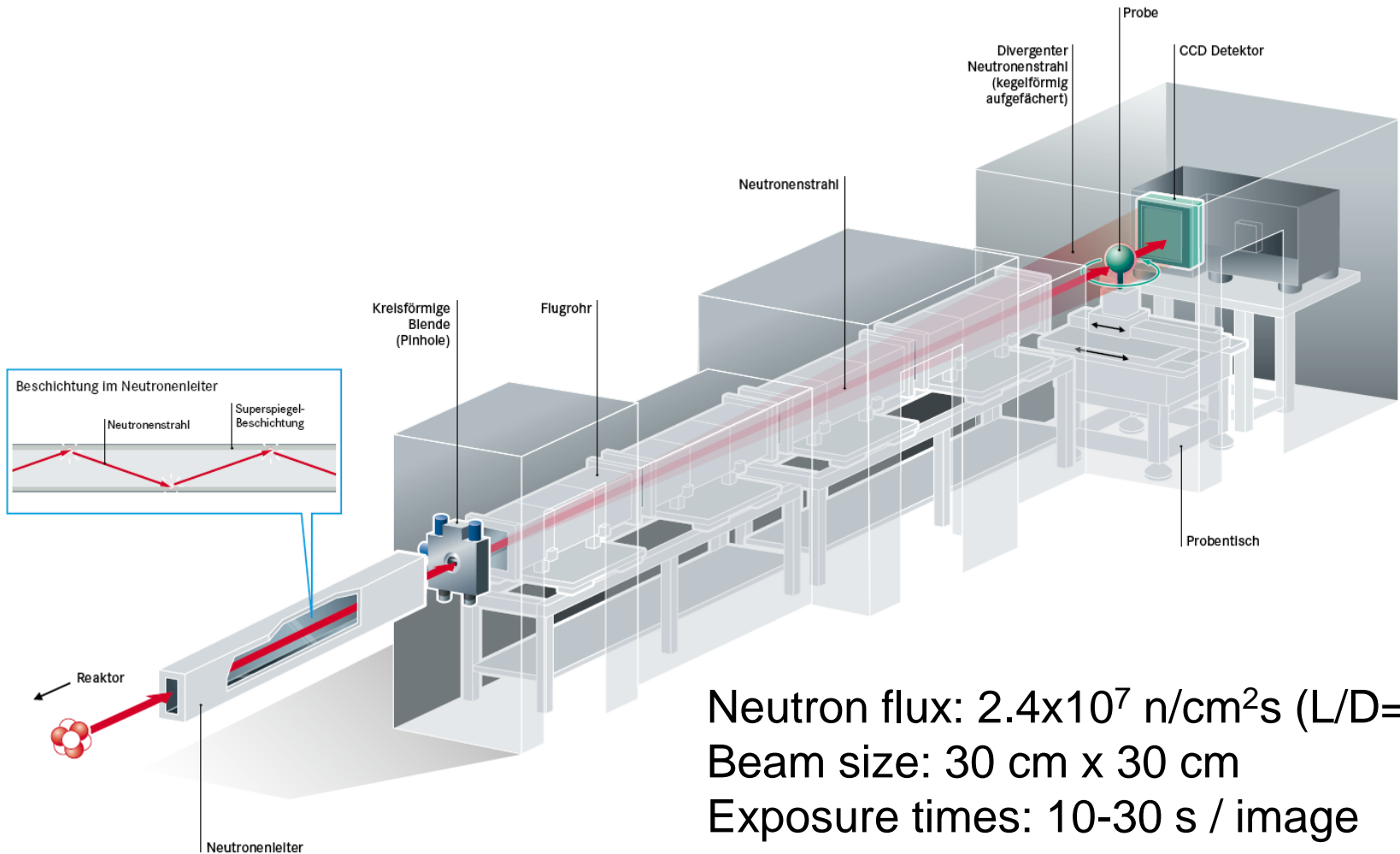
**Capacity:**

more than  $10^{14}$  neutrons per square centimeter per second in the core  
10 megawatts thermal power

The research reactor BER II is a source of neutron beams for structural and materials research.

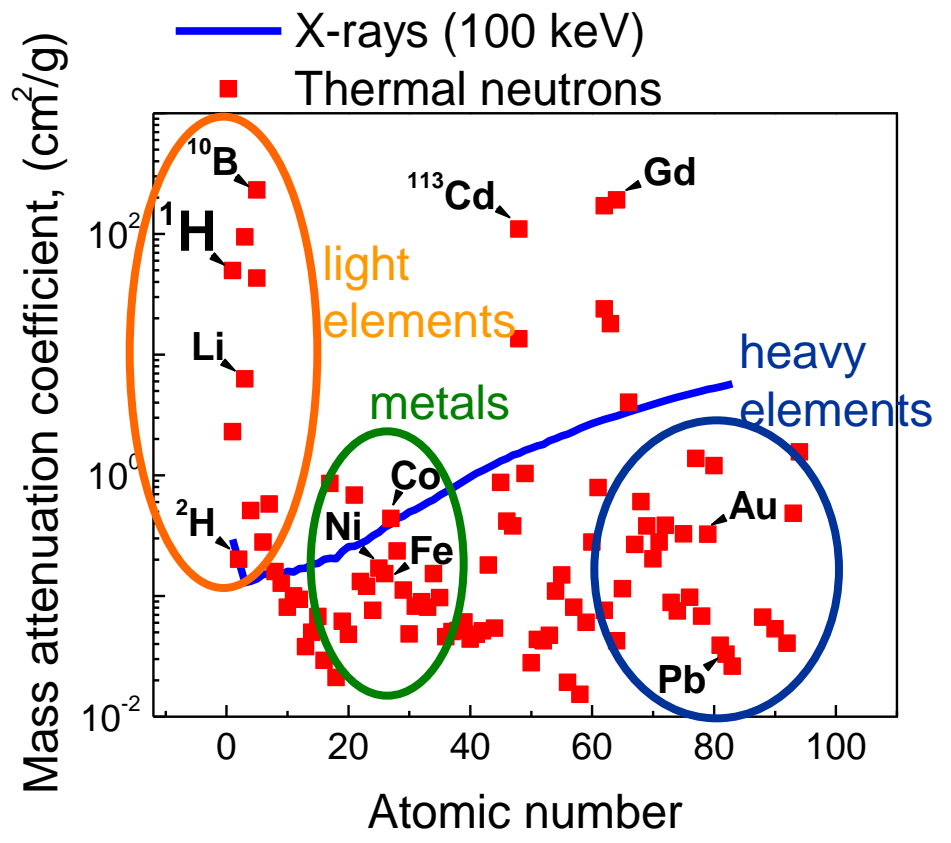
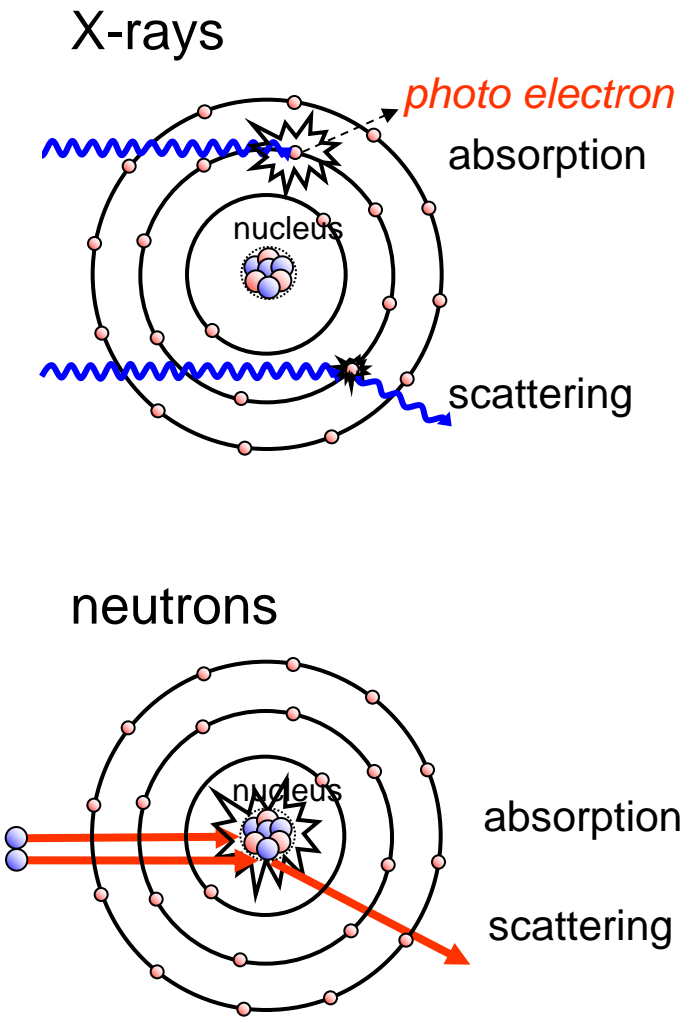
# Neutron imaging

## CONRAD-2 Instrument



Neutron flux:  $2.4 \times 10^7$  n/cm<sup>2</sup>s (L/D=350)  
Beam size: 30 cm x 30 cm  
Exposure times: 10-30 s / image

# Neutron interaction with matter



# Neutron interaction

Attenuation coefficients with X-ray [ $\text{cm}^2 \text{g}^{-1}$ ]

| 1a         | 2a          | 3b          | 4b          | 5b          | 6b         | 7b          | 8           |             |             |             |             | 1b          | 2b          | 3a          | 4a          | 5a         | 6a         | 7a         | 0          |
|------------|-------------|-------------|-------------|-------------|------------|-------------|-------------|-------------|-------------|-------------|-------------|-------------|-------------|-------------|-------------|------------|------------|------------|------------|
| H<br>0.02  |             |             |             |             |            |             |             |             |             |             |             |             |             |             |             |            |            |            | He<br>0.02 |
| Li<br>0.06 | Be<br>0.22  |             |             |             |            |             |             |             |             |             |             |             |             | B<br>0.28   | C<br>0.27   | N<br>0.11  | O<br>0.16  | F<br>0.14  | Ne<br>0.17 |
| Na<br>0.13 | Mg<br>0.24  |             |             |             |            |             |             |             |             |             |             |             |             | Al<br>0.38  | Si<br>0.33  | P<br>0.25  | S<br>0.30  | Cl<br>0.23 | Ar<br>0.20 |
| K<br>0.14  | Ca<br>0.26  | Sc<br>0.48  | Ti<br>0.73  | V<br>1.04   | Cr<br>1.29 | Mn<br>1.32  | Fe<br>1.57  | Co<br>1.78  | Ni<br>1.96  | Cu<br>1.97  | Zn<br>1.64  | Ga<br>1.42  | Ge<br>1.33  | As<br>1.50  | Se<br>1.23  | Br<br>0.90 | Kr<br>0.73 |            |            |
| Rb<br>0.47 | Sr<br>0.86  | Y<br>1.61   | Zr<br>2.47  | Nb<br>3.43  | Mo<br>4.29 | Tc<br>5.06  | Ru<br>5.71  | Rh<br>6.08  | Pd<br>6.13  | Ag<br>5.67  | Cd<br>4.84  | In<br>4.31  | Sn<br>3.98  | Sb<br>4.28  | Te<br>4.06  | I<br>3.45  | Xe<br>2.53 |            |            |
| Cs<br>1.42 | Ba<br>2.73  | La<br>5.04  | Hf<br>19.70 | Ta<br>25.47 | W<br>30.49 | Re<br>34.47 | Os<br>37.92 | Ir<br>39.01 | Pt<br>38.61 | Au<br>35.94 | Hg<br>25.88 | Tl<br>23.23 | Pb<br>22.81 | Bi<br>20.28 | Po<br>20.22 | At         | Rn<br>9.77 |            |            |
| Fr         | Ra<br>11.80 | Ac<br>24.47 | Rf          | Ha          |            |             |             |             |             |             |             |             |             |             |             |            |            |            |            |

|             |             |             |            |            |            |            |            |            |             |             |             |             |            |             |
|-------------|-------------|-------------|------------|------------|------------|------------|------------|------------|-------------|-------------|-------------|-------------|------------|-------------|
| Lanthanides | Ce<br>5.79  | Pr<br>6.23  | Nd<br>6.46 | Pm<br>7.33 | Sm<br>7.68 | Eu<br>5.66 | Gd<br>8.69 | Tb<br>9.46 | Dy<br>10.17 | Ho<br>10.91 | Er<br>11.70 | Tm<br>12.49 | Yb<br>9.32 | Lu<br>14.07 |
| *Actinides  | Th<br>28.95 | Pa<br>39.65 | U<br>49.08 | Np         | Pu         | Am         | Cm         | Bk         | Vf          | Es          | Fm          | Md          | No         | Lr<br>x-ray |

## Legend

Attenuation coefficient [ $\text{cm}^2 \text{g}^{-1}$ ] = sp.gr. \*  $\mu/\delta$

sp.gr.: Handbook of Chemistry and Physics, 56th Edition 1975-1976.

$\mu/\delta$ : J. H. Hubbell<sup>+</sup> and S. M. Seltzer Ionizing Radiation Division, Physics Laboratory National Institute of Standards and Technology Gaithersburg, MD 20899,  
<http://physics.nist.gov/PhysRefData/XrayMassCoef/tab3.html>.

**X-ray**



# Neutron interaction

Attenuation coefficients with neutrons [cm<sup>-1</sup>]

| 1a           | 2a         | 3b         | 4b         | 5b         | 6b           | 7b          | 8             |             |             |            |              | 1b         | 2b         | 3a          | 4a         | 5a         | 6a        | 7a         | 0          |
|--------------|------------|------------|------------|------------|--------------|-------------|---------------|-------------|-------------|------------|--------------|------------|------------|-------------|------------|------------|-----------|------------|------------|
| H<br>3.44    |            |            |            |            |              |             |               |             |             |            |              |            |            |             |            |            |           |            | He<br>0.02 |
| Li<br>3.30   | Be<br>0.79 |            |            |            |              |             |               |             |             |            |              |            |            | B<br>101.60 | C<br>0.56  | N<br>0.43  | O<br>0.17 | F<br>0.20  | Ne<br>0.10 |
| Na<br>0.09   | Mg<br>0.15 |            |            |            |              |             |               |             |             |            |              |            |            | Al<br>0.10  | Si<br>0.11 | P<br>0.12  | S<br>0.06 | Cl<br>1.33 | Ar<br>0.03 |
| K<br>0.06    | Ca<br>0.08 | Sc<br>2.00 | Ti<br>0.60 | V<br>0.72  | Cr<br>0.54   | Mn<br>1.21  | Fe<br>1.19    | Co<br>3.92  | Ni<br>2.05  | Cu<br>1.07 | Zn<br>0.35   | Ga<br>0.49 | Ge<br>0.47 | As<br>0.67  | Se<br>0.73 | Br<br>0.24 |           |            | Kr<br>0.61 |
| Rb<br>0.08   | Sr<br>0.14 | Y<br>0.27  | Zr<br>0.29 | Nb<br>0.40 | Mo<br>0.52   | Tc<br>1.76  | Ru<br>0.58    | Rh<br>10.88 | Pd<br>0.78  | Ag<br>4.04 | Cd<br>115.11 | In<br>7.58 | Sn<br>0.21 | Sb<br>0.30  | Te<br>0.25 | I<br>0.23  |           |            | Xe<br>0.43 |
| Cs<br>0.29   | Ba<br>0.07 | La<br>0.52 | Hf<br>4.99 | Ta<br>1.49 | W<br>1.47    | Re<br>6.85  | Os<br>2.24    | Ir<br>30.46 | Pt<br>1.46  | Au<br>6.23 | Hg<br>16.21  | Tl<br>0.47 | Pb<br>0.38 | Bi<br>0.27  | Po         | At         |           |            | Rn         |
| Fr           | Ra<br>0.34 | Ac         | Rf         | Ha         |              |             |               |             |             |            |              |            |            |             |            |            |           |            |            |
|              |            |            |            |            |              |             |               |             |             |            |              |            |            |             |            |            |           |            |            |
| *Lanthanides | Ce<br>0.14 | Pr<br>0.41 | Nd<br>1.87 | Pm<br>5.72 | Sm<br>171.47 | Eu<br>94.58 | Gd<br>1479.04 | Tb<br>0.93  | Dy<br>32.42 | Ho<br>2.25 | Er<br>5.48   | Tm<br>3.53 | Yb<br>1.40 | Lu<br>2.75  |            |            |           |            |            |
| **Actinides  | Th<br>0.59 | Pa<br>8.46 | U<br>0.82  | Np<br>9.80 | Pu<br>50.20  | Am<br>2.86  | Cm            | Bk          | Cf          | Es         | Fm           | Md         | No         | Lr<br>neut. |            |            |           |            |            |

## Legend

$$\sigma\text{-total} * \text{sp.gr.} * 0.6023$$

$$\text{Attenuation coefficient [cm}^{-1}\text{]} = \frac{\sigma\text{-total} * \text{sp.gr.} * 0.6023}{\text{at.wt.}}$$

**thermal neutrons**

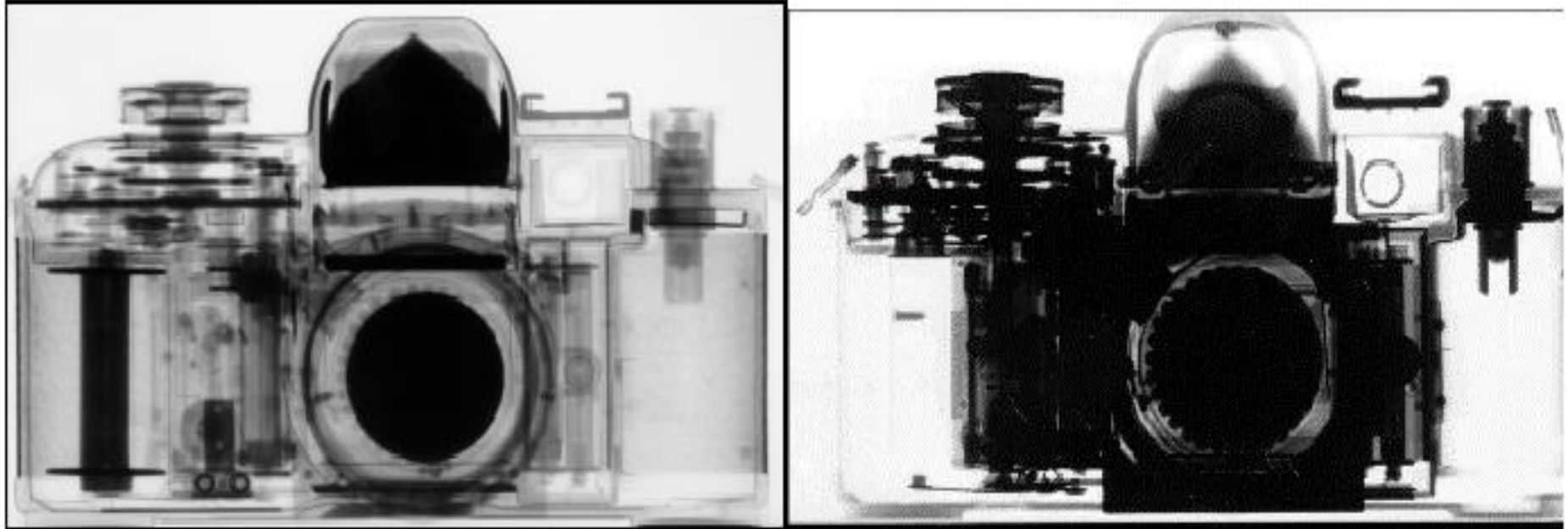
$\sigma\text{-total}$ : JEF Report 14, TABLE OF SIMPLE INTEGRAL NEUTRON CROSS SECTION DATA FROM JEF-2.2, ENDF/B-VI, JENDL-3.2, BROND-2 AND CENDL-2, AEN NEA, 1994.

and Special Feature: Neutron scattering lengths and cross sections, Varley F. Sears, AECL Research, Chalk River Laboratories Chalk River, Ontario, Canada KOJ 1JO, Neutron News, Vol. 3, 1992, <http://www.ncnr.nist.gov/resources/n-lengths/list.html>.

sp.gr.: Handbook of Chemistry and Physics, 56th Edition 1975-1976.

at.wt.: Handbook of Chemistry and Physics, 56th Edition 1975-1976.

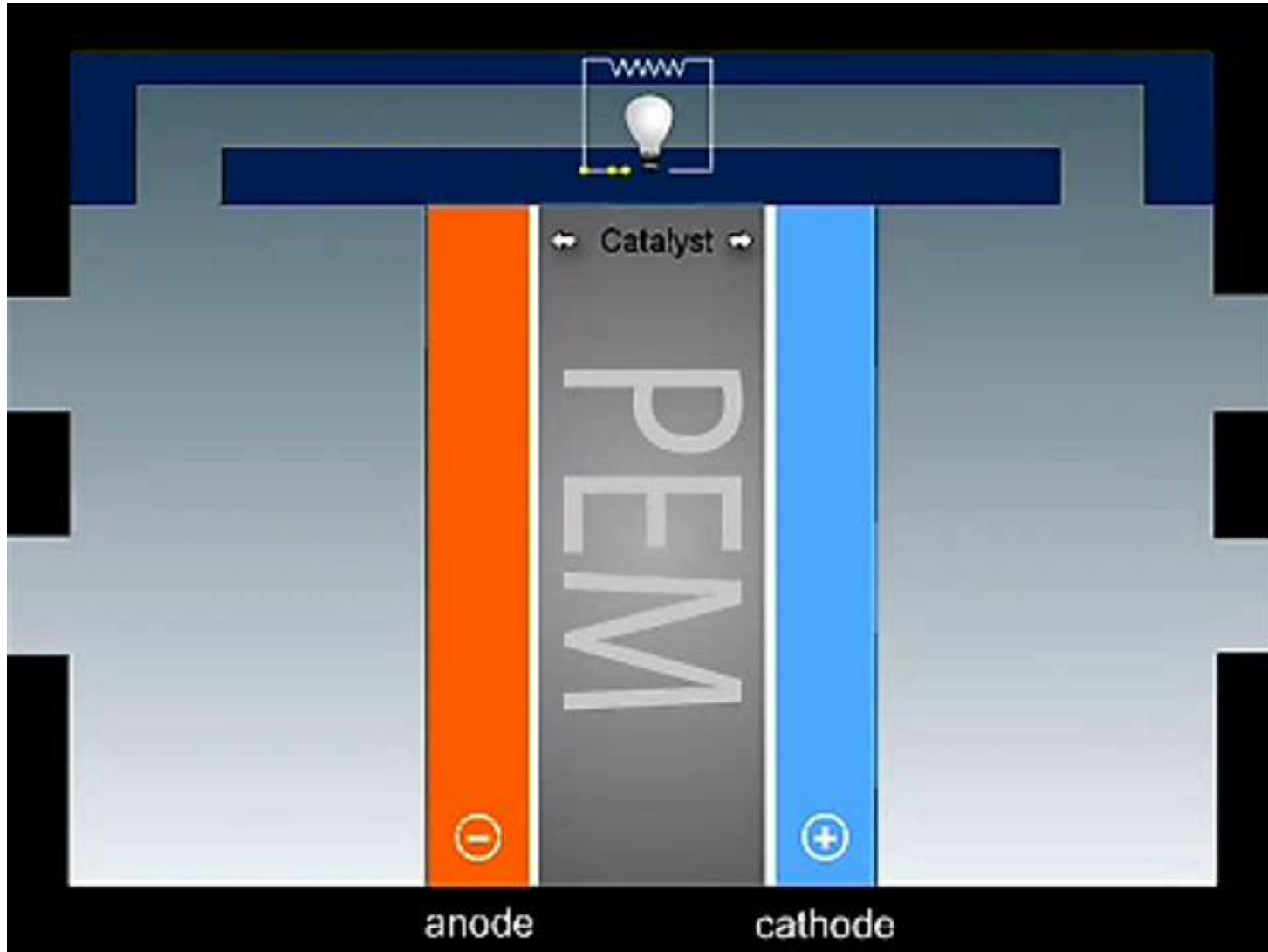
## Neutron radiography - examples



The example for a camera helps to explain differences in neutron (left) and X-ray (right) radiography. Whereas the hydrogen containing parts can be visualised with neutron even at thin layers, thicker metallic components are hard to penetrate with X-rays.

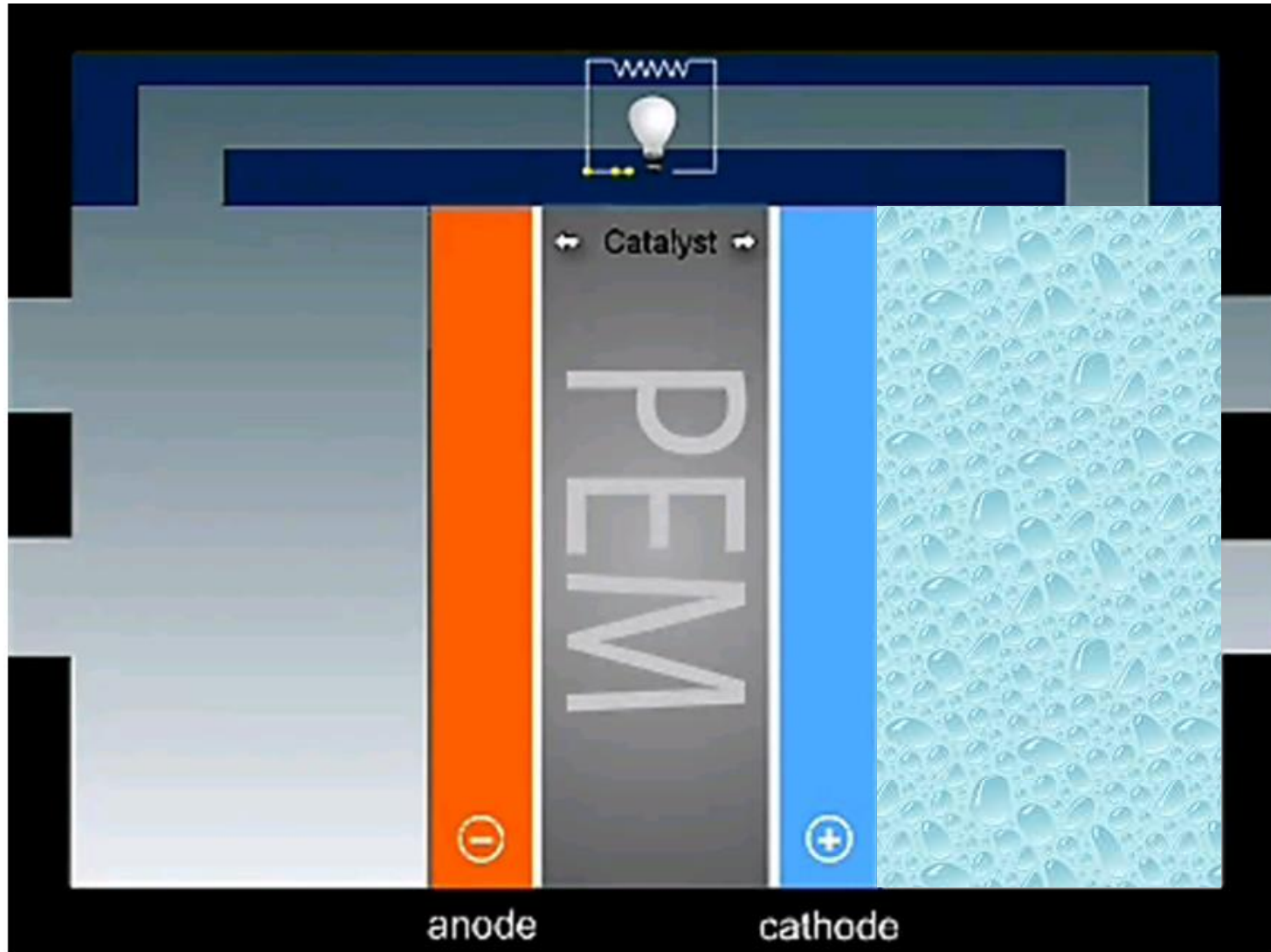
# Attenuation Contrast

## Fuel cells



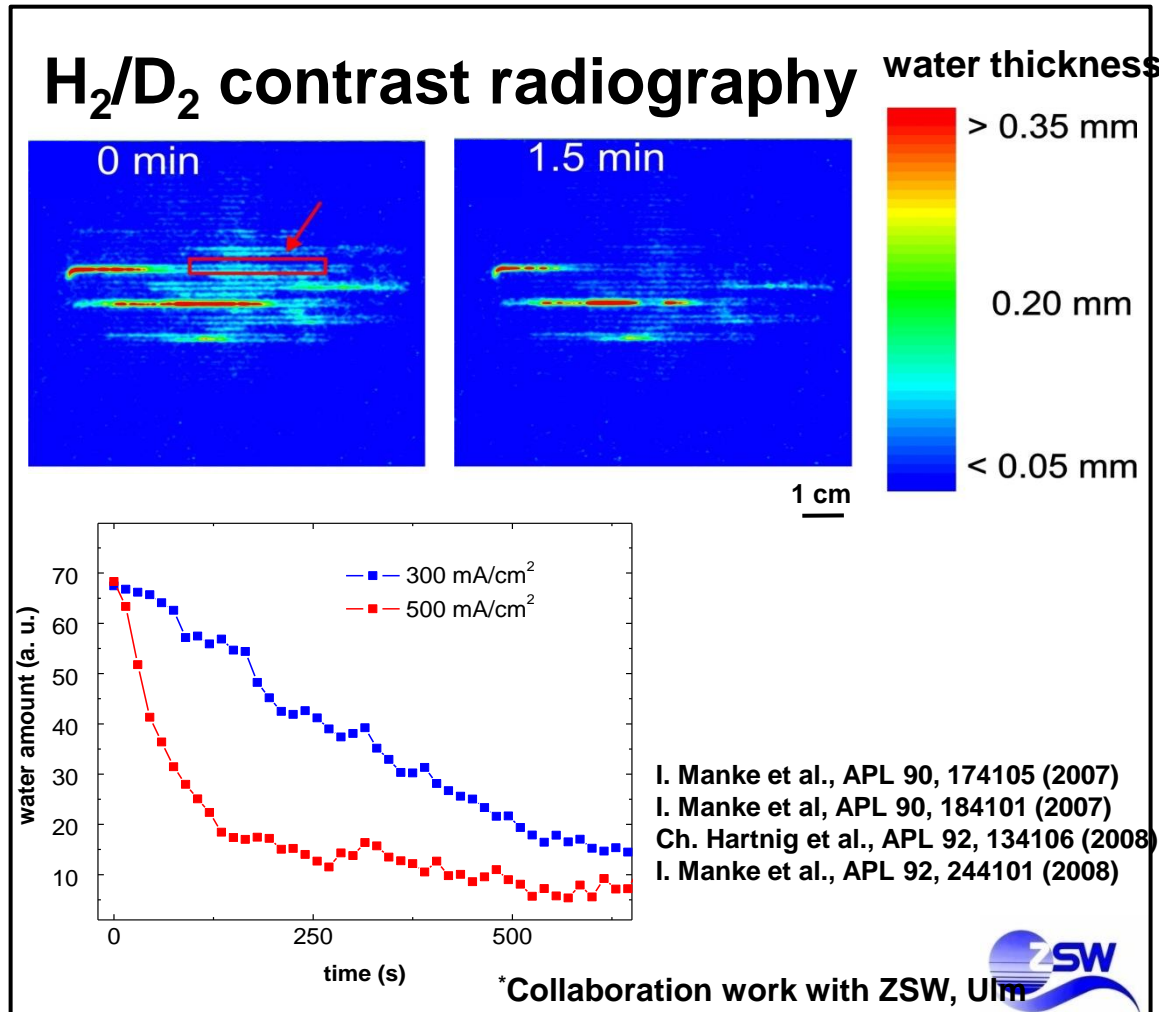
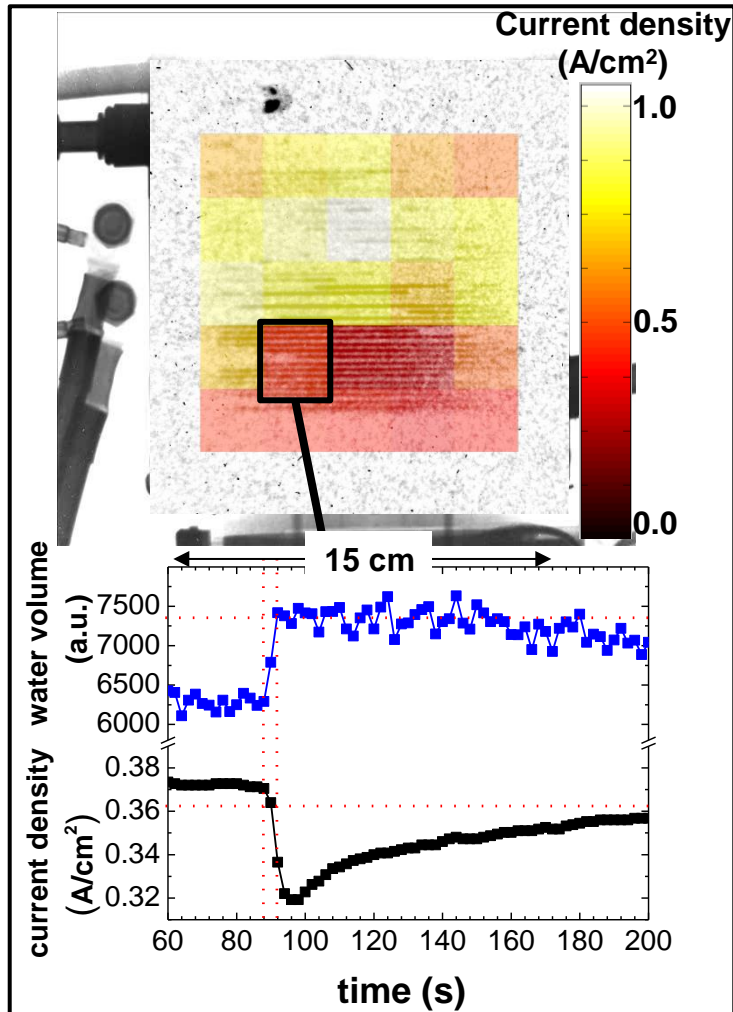
# Attenuation Contrast

## Fuel cells



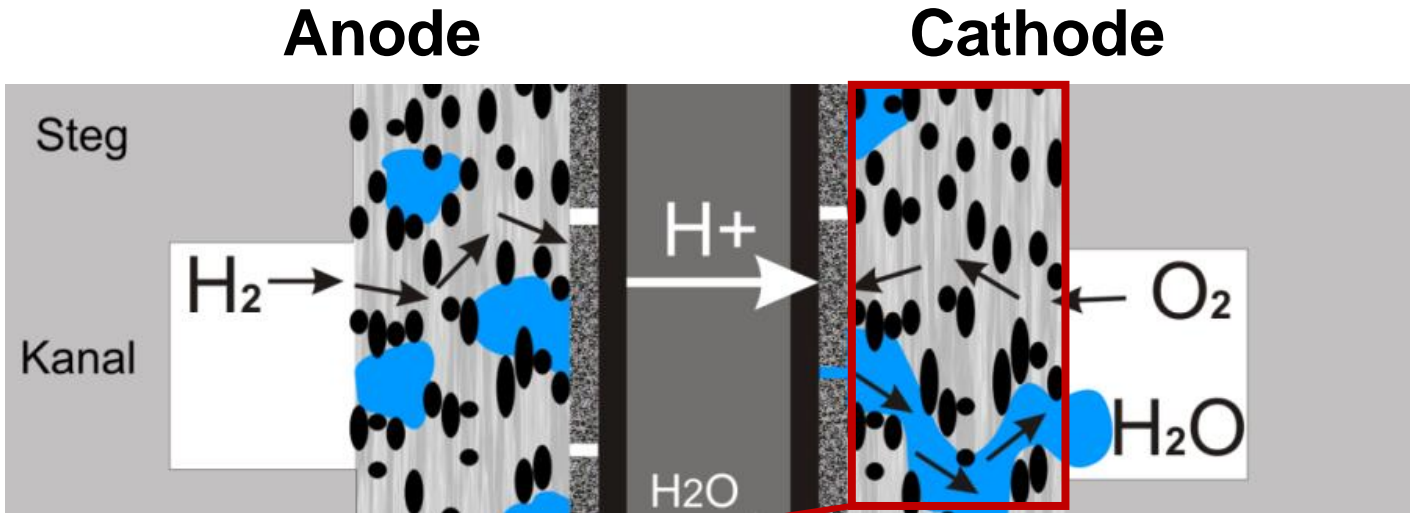
# Attenuation Contrast

## Fuel cells



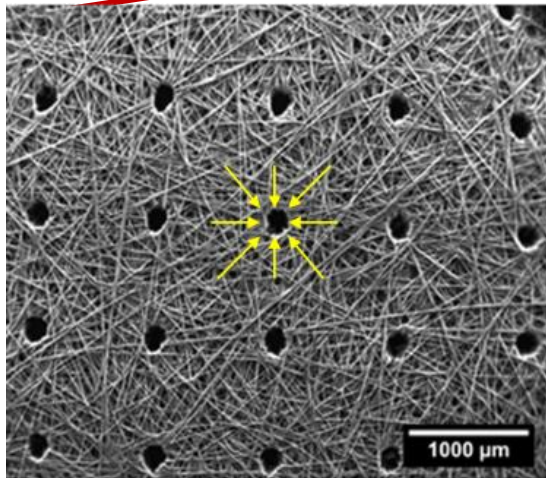
# Attenuation Contrast

## Fuel Cells

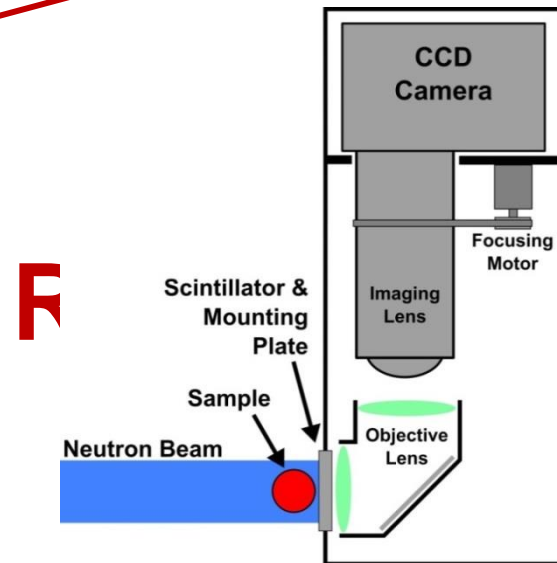


Innovative design

Hydrophobic material



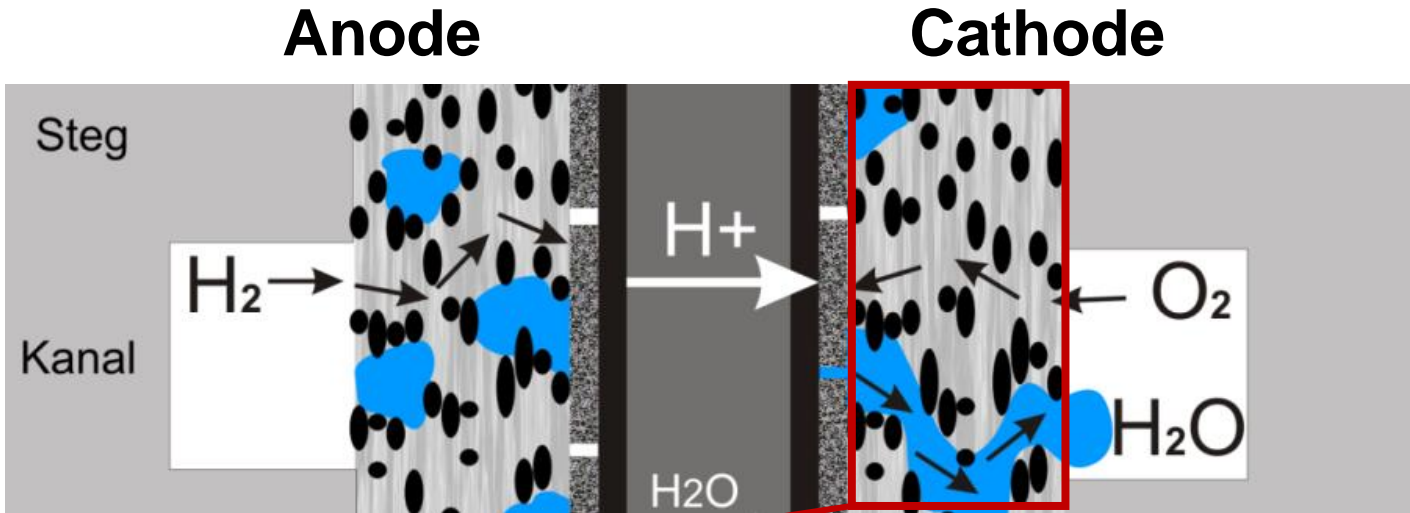
Perforation = Drainage effect





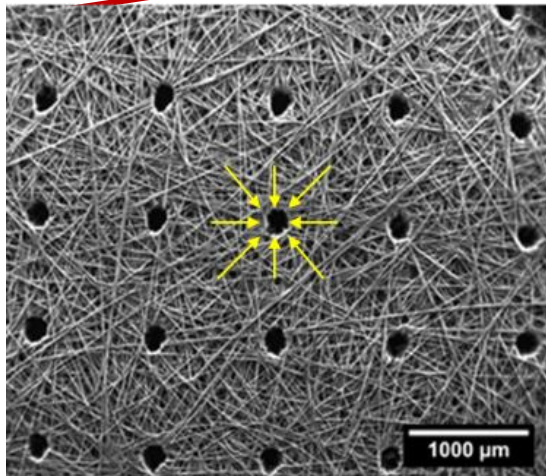
# Attenuation Contrast

## Fuel Cells

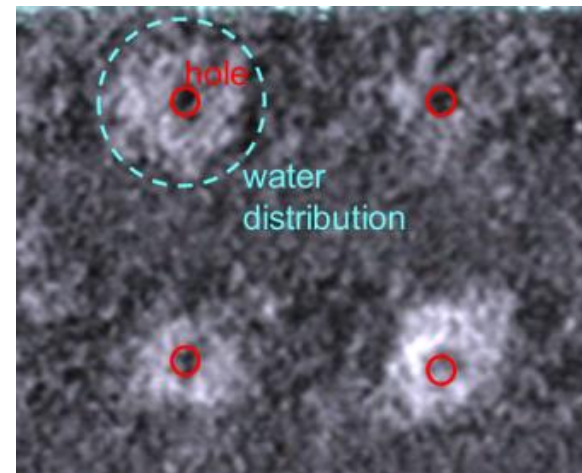


Innovative design

Hydrophobic material



Perforation = Drainage effect



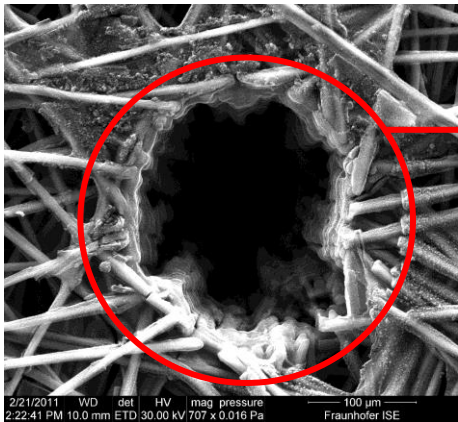
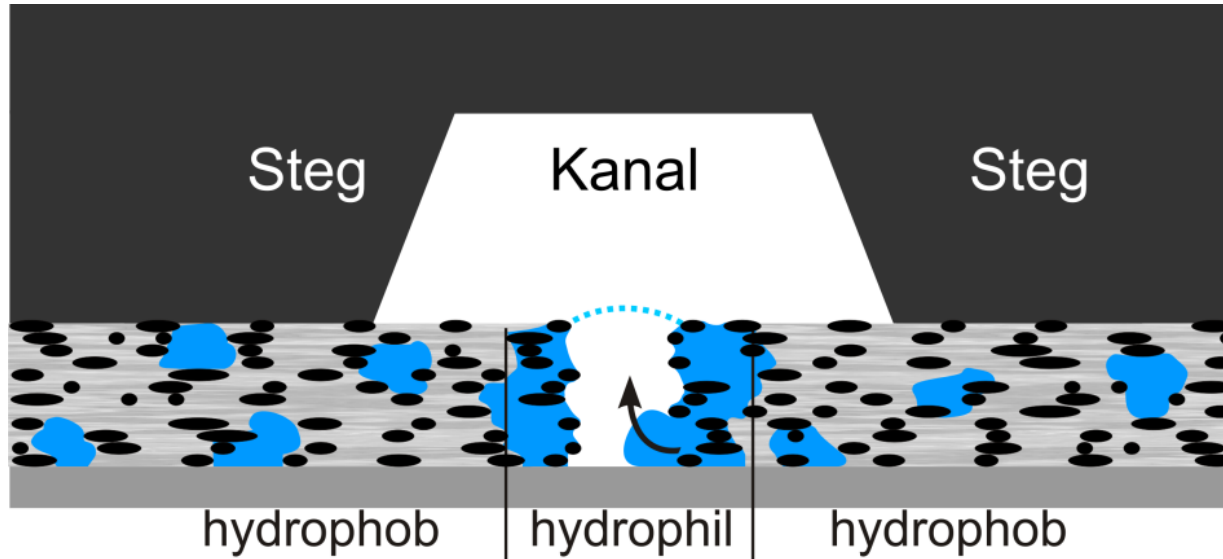
Neutron image

BER II



# Attenuation Contrast

## Fuel Cells



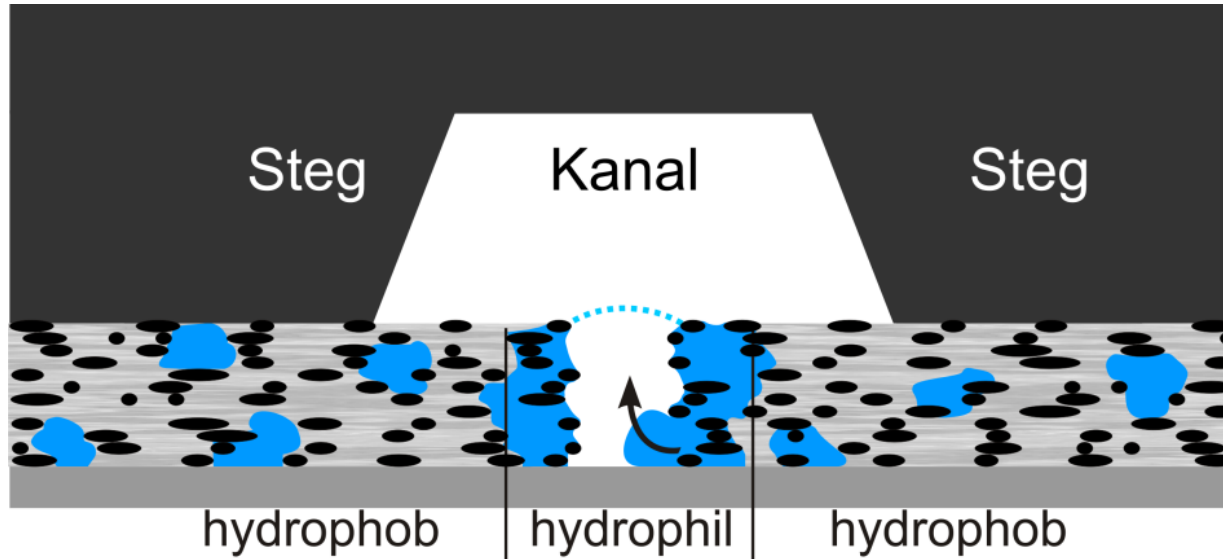
Heat affected zone



Hydrophilic areas cause water agglomerations

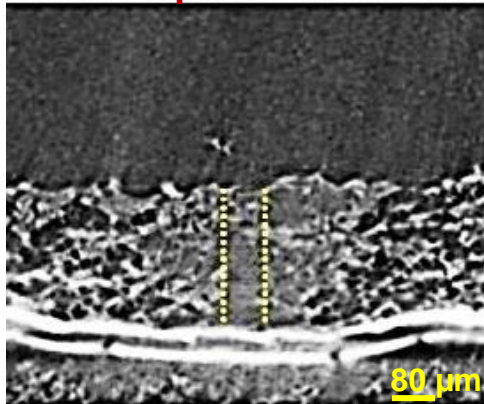
# Attenuation Contrast

## Fuel Cells

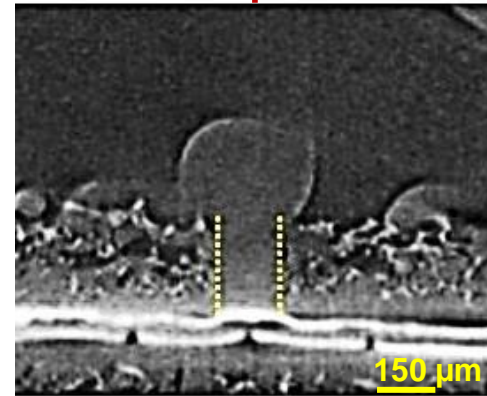


BESSY II

laser perforation



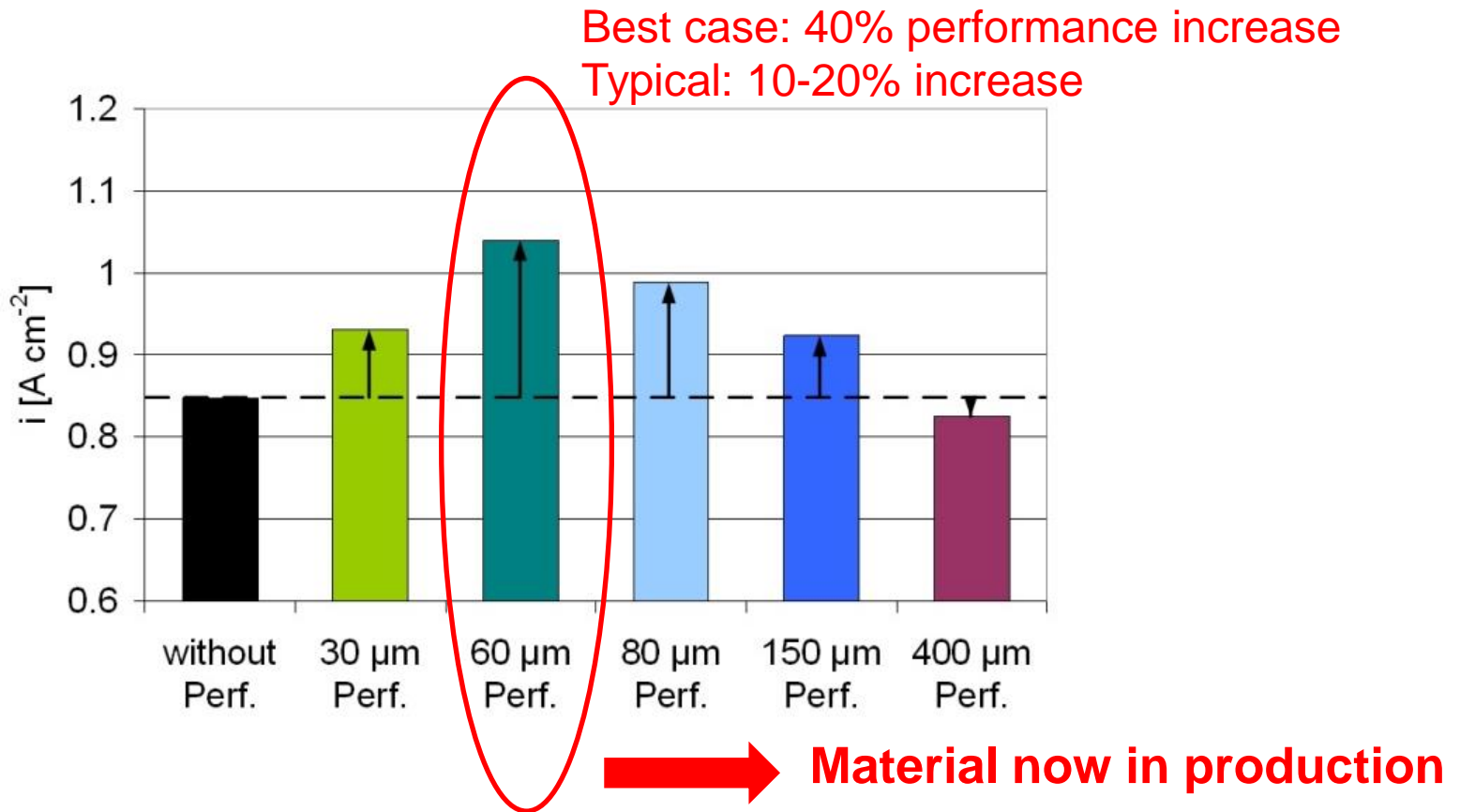
mechanic perforation



dynamic synchrotron radiography

# Attenuation Contrast

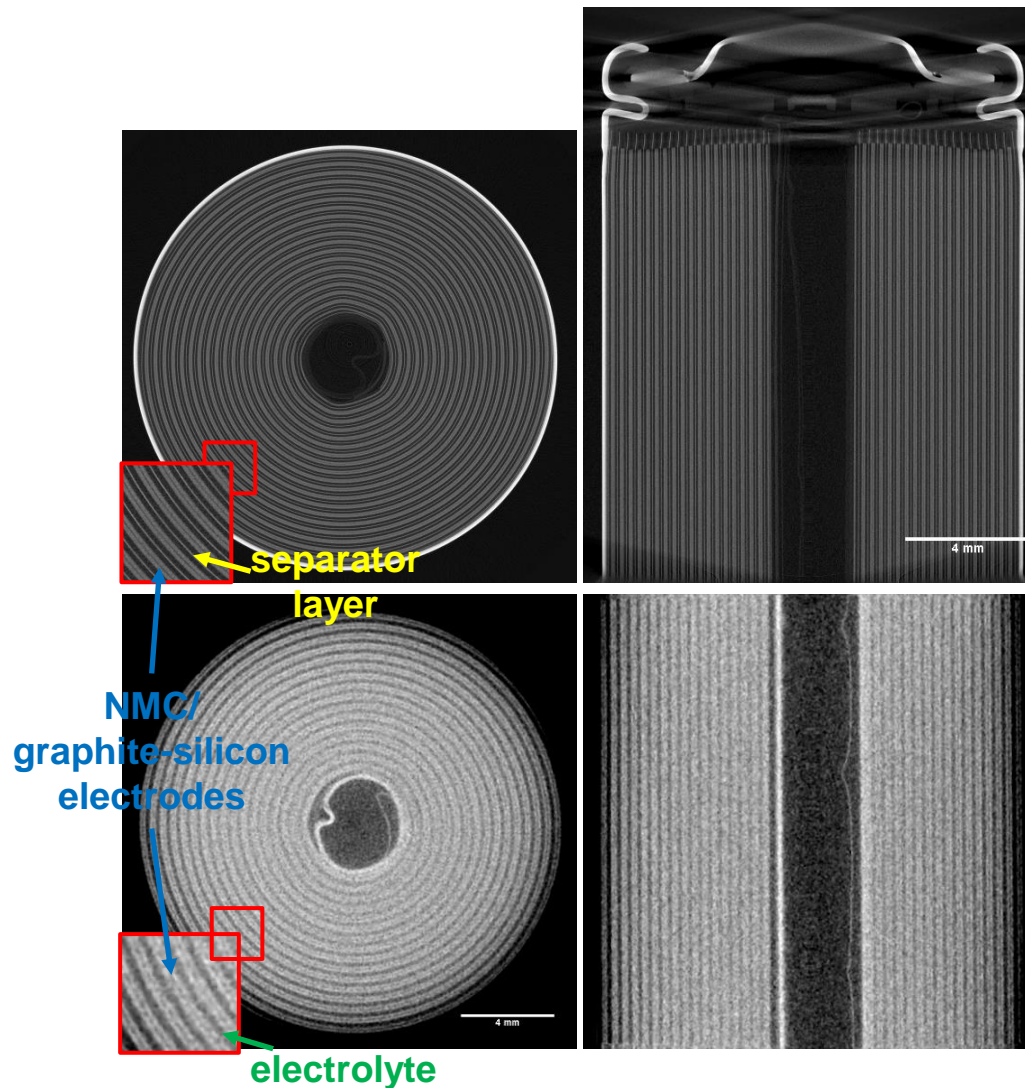
## Fuel Cells



J. Haußmann *et al*  
Journal of Power Sources 239  
(2013) 611



R. Ziesche  
W. Kockelmann  
P. Shearing

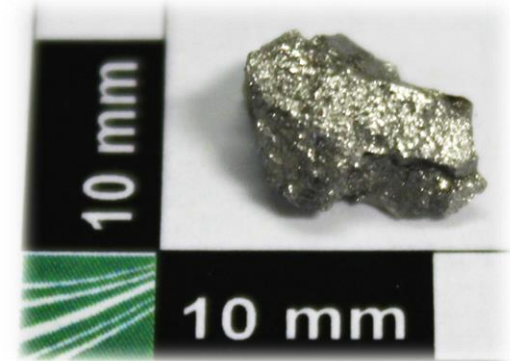


X-ray

Neutron

**MJ1:** Li-ion cell with NMC cathode and new graphite-silicon anode for a high capacity of 3500 mAh.

## 1. Introduction - Metal-Hydride-Composites (MHC)



|                      | Powder bed | MHC              |
|----------------------|------------|------------------|
| Second phase         | -          | 5 wt.% Graphite  |
| Thermal conductivity | ~1W/mK     | 14 W/mK (radial) |
| System load time     | >1h        | < 10 min         |
| Porosity             | ~ 70 vol.% | <b>30 vol.%</b>  |



**Challenge:**

**Volume expansion  
of the crystal lattice**



**Volume expansion  
and stability of MHCs?**

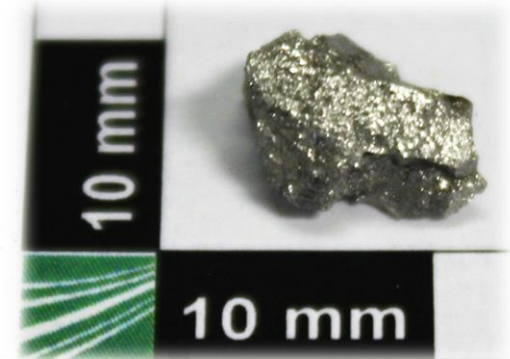
**Stress generation on  
container walls?**

F. Heubner, et al. Journal of Power Sources 397, 262-270

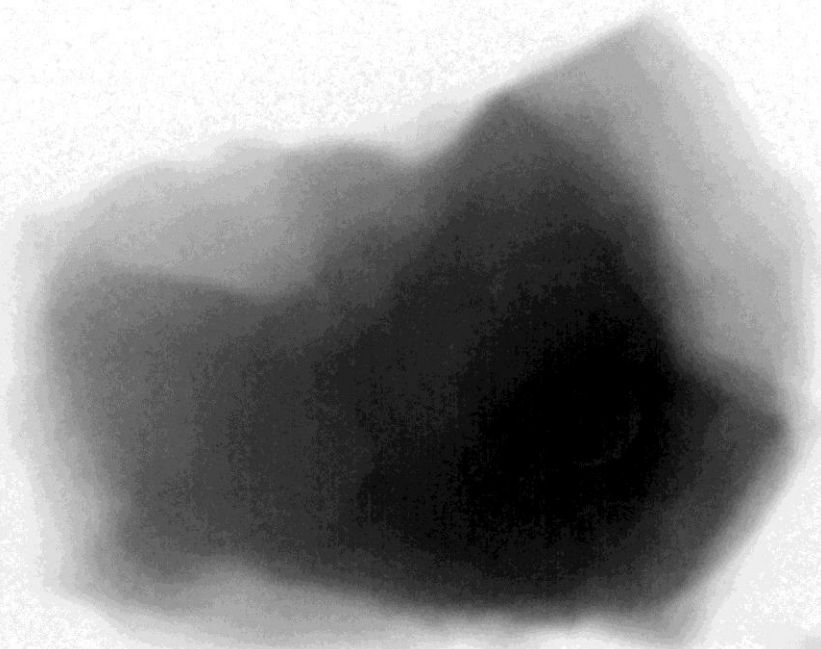


## 1. Introduction

Neutron imaging (radiograph series) of the hydrogenation of an  $AB_2$  alloy (Ti-Mn-based) at 30 bar  $H_2$ ; room temperature



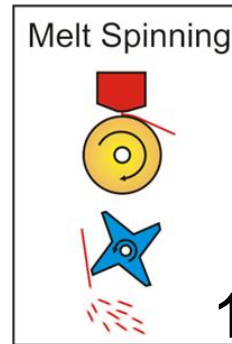
**Challenge:**  
Volume expansion  
of the crystal lattice



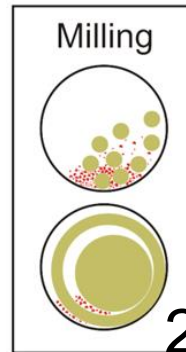
1 mm

## 2. Materials Processing

Granules



melt spinning (flakes)



milling (powder)

### Hydrallloy → Ti-Mn-V-Fe-Zr Alloy

|                     |               |
|---------------------|---------------|
| H-capacity          | ~1.8 wt.%     |
| reaction enthalpy   | ~ 27 kJ/mol   |
| working temperature | -10 ... 100°C |

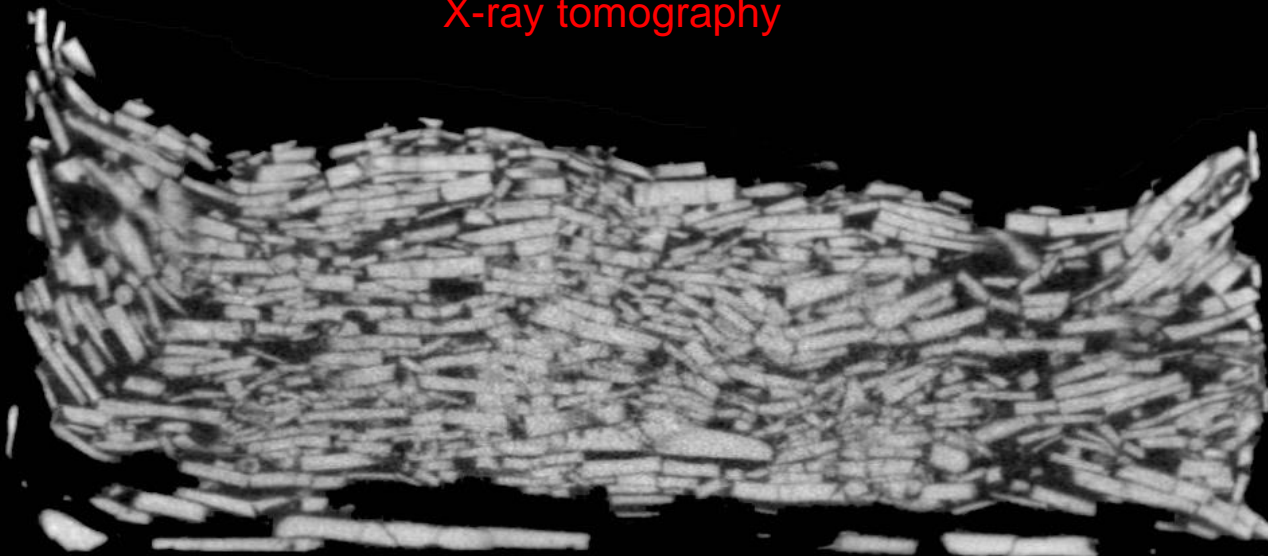
<sup>1</sup>Pohlmann et al. (2011), J of Alloys and Compounds 509, p. 625-628

<sup>2</sup>Pohlmann et al. (2013), J of Power Sources 231, p. 97-105



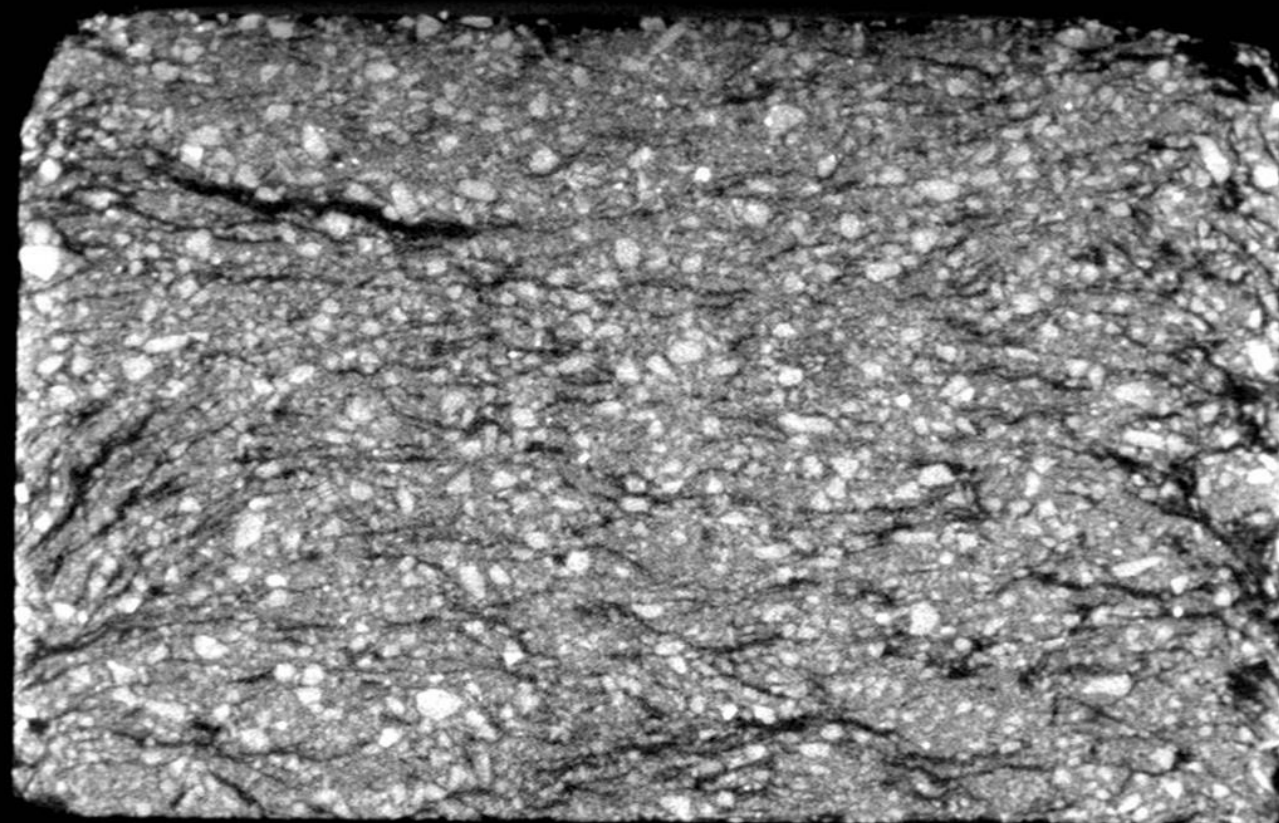
X-ray tomography

Flake MHC  
(radially aligned flakes)



Porosity:  
27-30 vol.%

Graphite:  
10 vol.%



Powder MHC

1 mm

---

## 3. Neutron Imaging

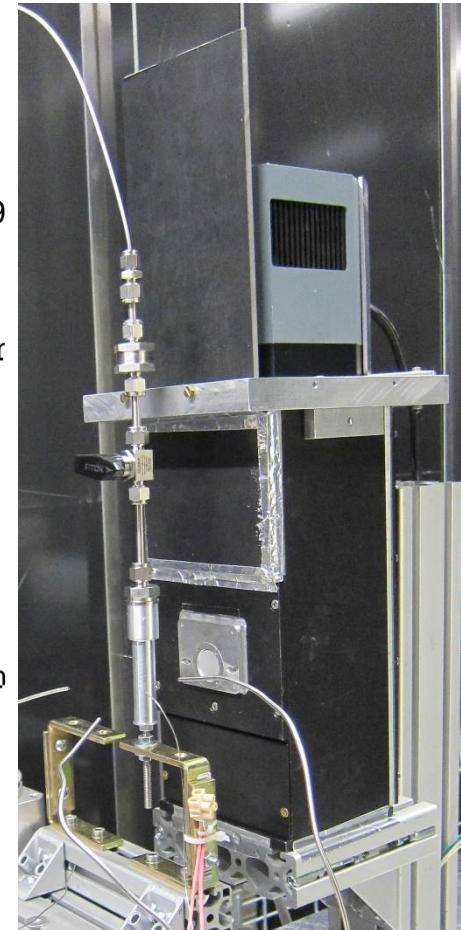
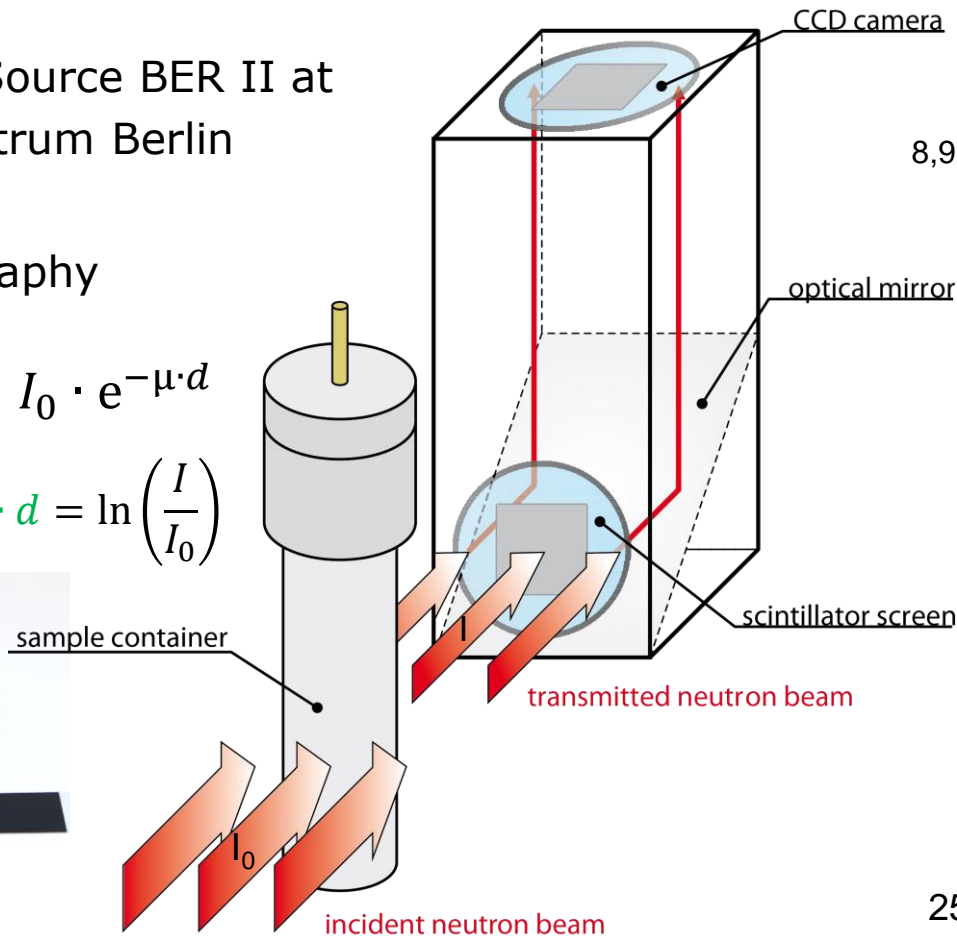
- Cold Neutron Source BER II at Helmholtz-Zentrum Berlin
- Beamline (V7): Radio-/Tomography

Lambert Beer:  $I = I_0 \cdot e^{-\mu \cdot d}$

Extinction:  $E = -\mu \cdot d = \ln\left(\frac{I}{I_0}\right)$



diameter=12mm  
height=6-7mm



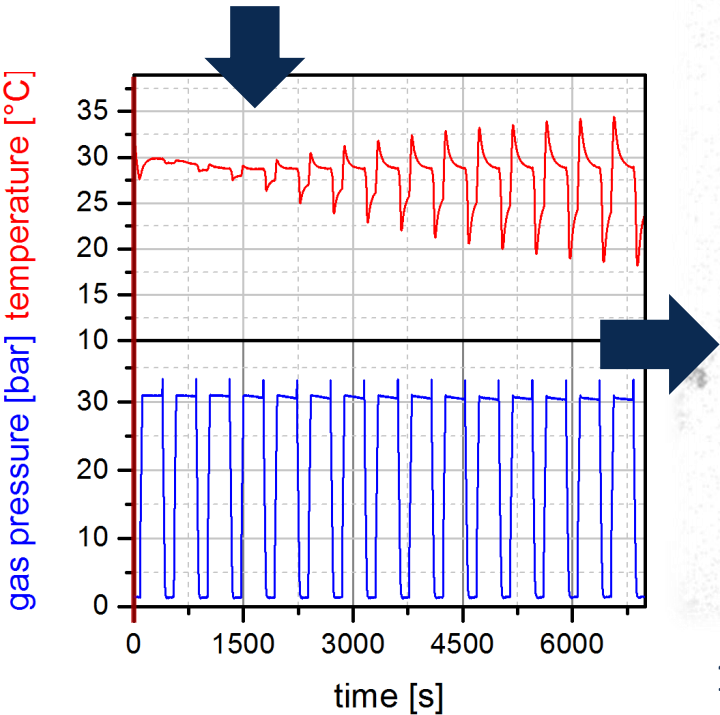
2560 x 2160 pixels (16 bit)  
Resolution per pixel: 5.4  $\mu\text{m}$

<sup>8</sup>Pohlmann et al. (2015), J of Power Sources 277, p. 360-369

<sup>9</sup>Herbrig et al. (2015), J of Power Sources 293, p. 109-118

## 3.1 Flake MHC

1. Activation  
2x heating up to 150°C
2. Cycling  
30/1.3 bar, RT



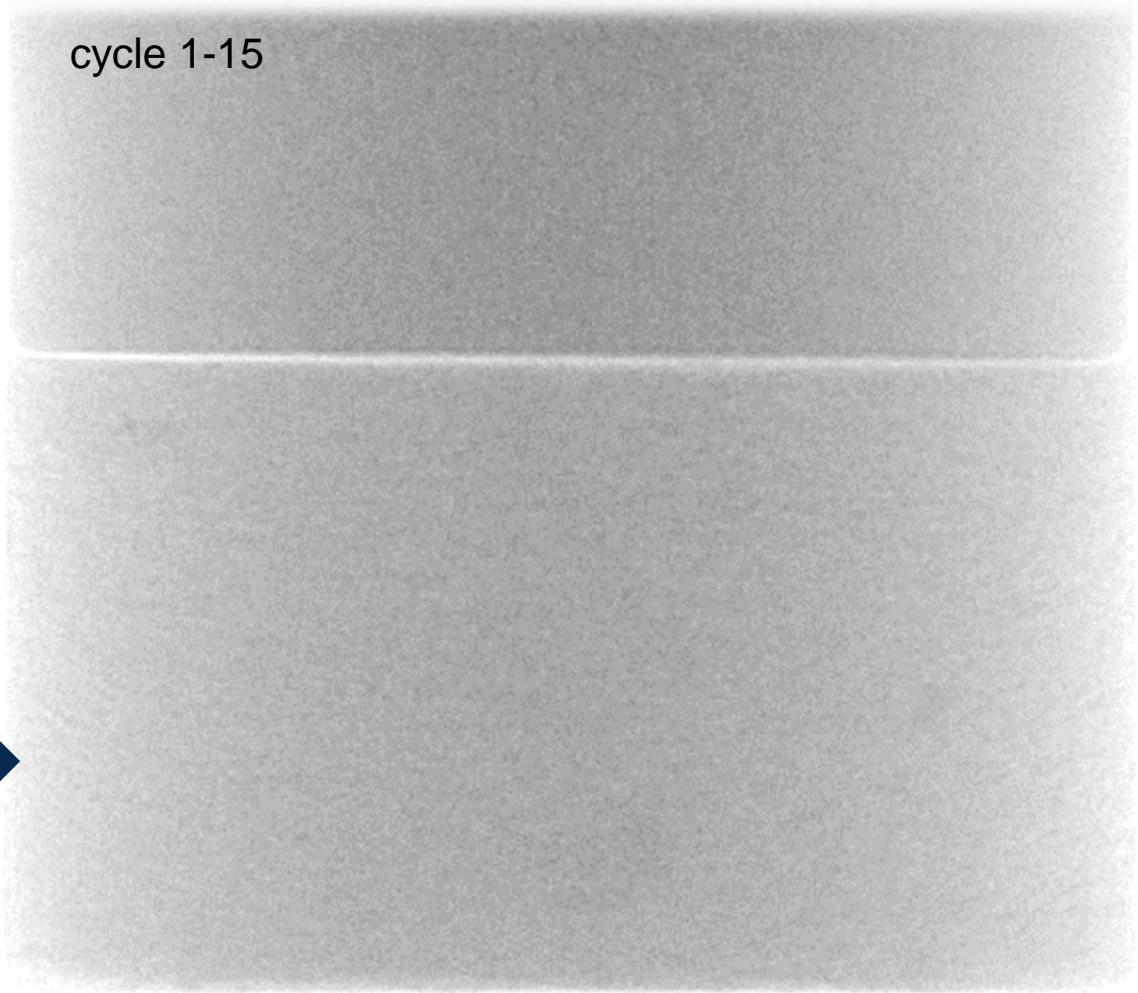
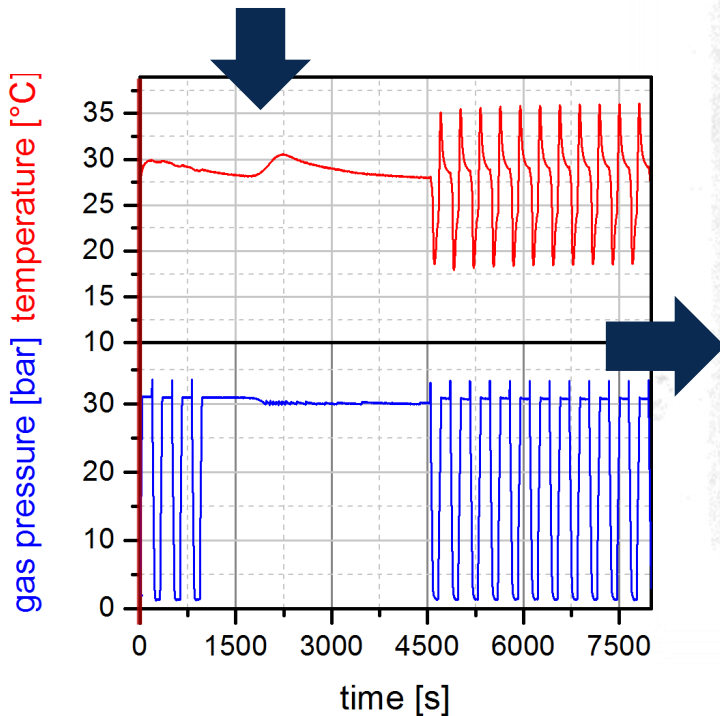
cycle 1-15





## 3.2 Powder MHC

1. Activation  
1x heating up to 150°C
2. Cycling  
30/1.3 bar, RT



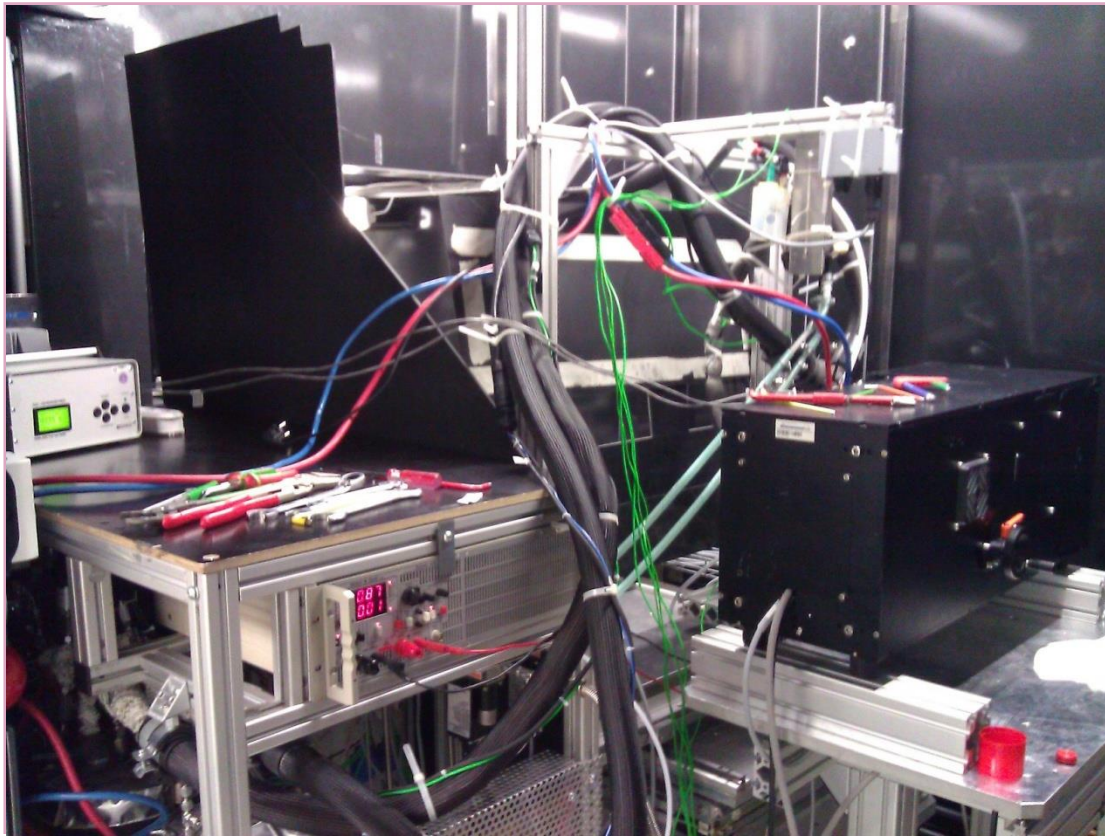
1 mm

# User operation





# User operation





# Thank you !

