

# **Lecture 12**

# **Air Separation**

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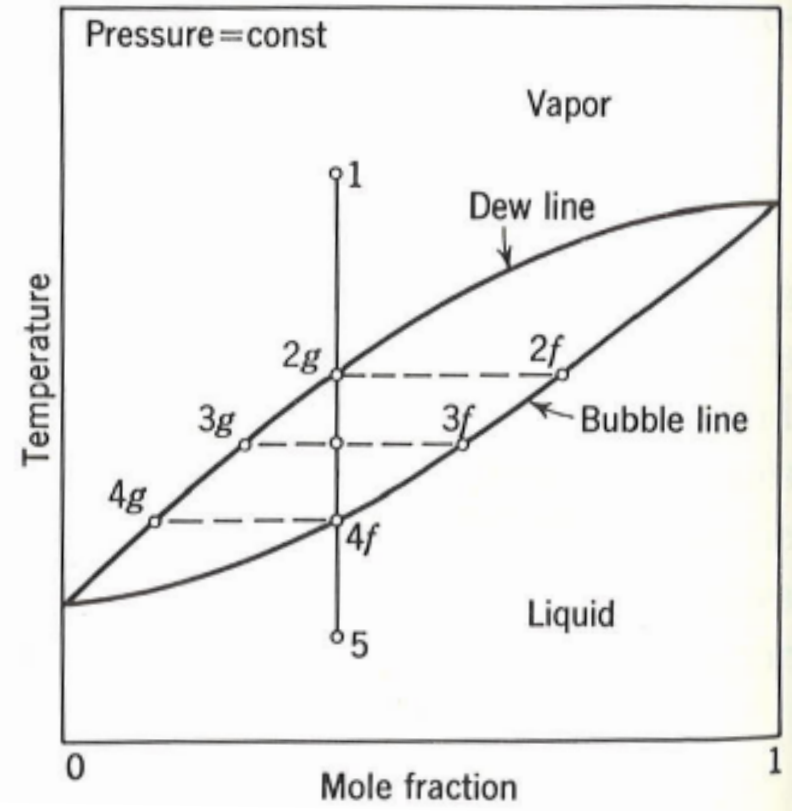
# Goals

- Describe the basics of the cryogenic separation of air into useful products
- Describe how He is recovered from Natural Gas

- While other techniques (adsorption, filtering by membranes, certain chemical reactions) exist, cryogenic air separation is the more efficient and mature technology for large scale air separation
- The separation of air into its constituent parts is a major industry and along with MRI systems is the bulk of the economic activity associated with cryogenics
- The technology associated with air separation is both mature and large scale – 10' s to 1000' s of tons per day per plant
- This industry involves the actual separation of the air and the distribution and storage of the resulting products
- Plants can be stand alone systems to separate air into its components for sale and distribution or can be incorporated into other industrial facilities ( chemical plants, steel mills, oil fields) to produce gases specifically for use at that site.

| Element  | Vol % in Dry Air | Boiling Point at 1 Atm (K) | Typical Uses  |
|----------|------------------|----------------------------|---|
| Nitrogen | 78.08            | 77                         | Cryogenics, Enhanced Oil & Gas Recovery, Chemical Plants, Other Purging & Pressurization Uses |
| Oxygen   | 20.95            | 90                         | Chemical Plants, Steel Mills, Medical Gases, Welding Gases                                    |
| Argon    | 0.93             | 87                         | Welding Gases, Inert Atmospheres, Incandescent & Florescent Lights                            |
| Neon     | 0.0018           | 27                         | Lighting and Signs  |
| Helium   | 0.0005           | 4.2                        | Cryogenics, Welding Gas, Fiber Optics Processing, Lifting (Balloons, Blimps)                  |
| Krypton  | 0.00011          | 119.8                      | Incandescent & Florescent Lights  |
| Xenon    | 0.000009         | 129                        | Incandescent & Strobe Lights  |

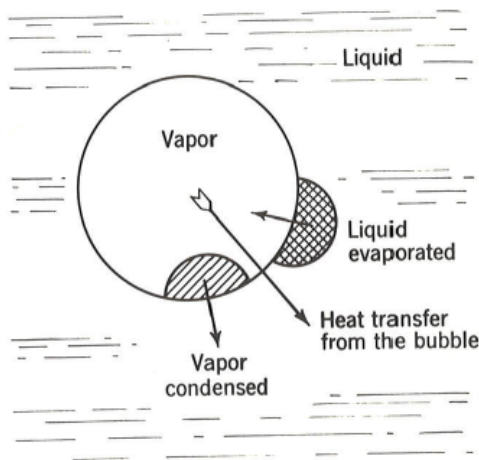
- All component gases (except He) are produced from air
  - He is recovered from certain natural gas fields where its concentration is 0.2 - 2 % or better
- In air separation plants, the working fluid is air which is of course a mixture not a pure fluid
- A basic principle of cryogenic air separation is Rectification
  - This is the cascading of evaporations and condensations done in counterflow



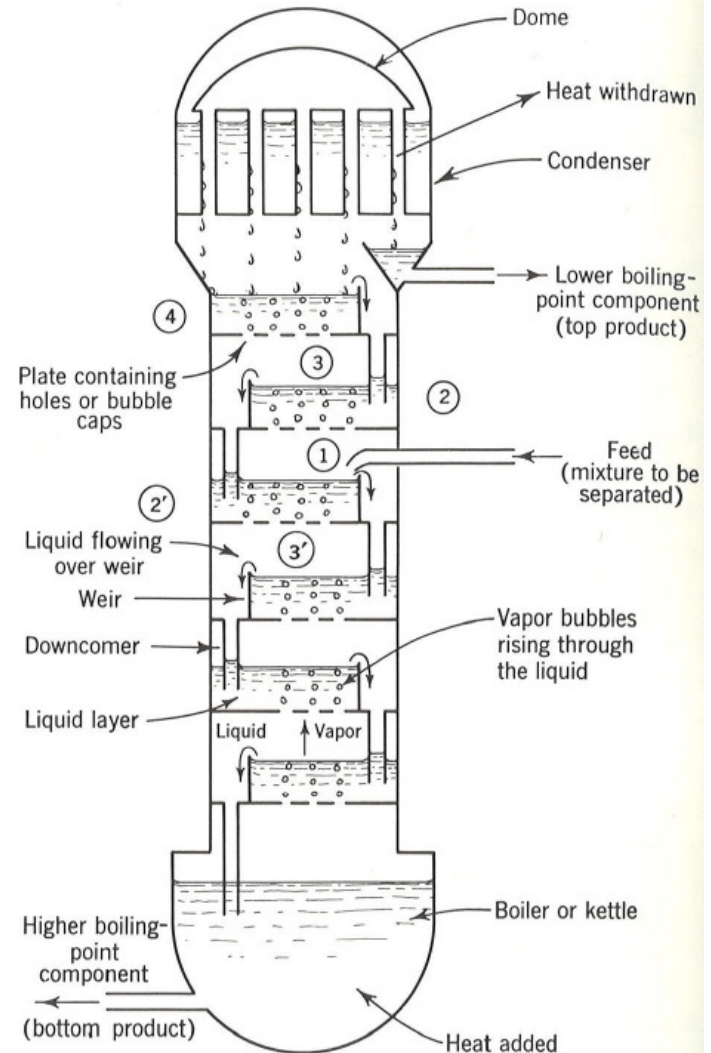
From Barron Cryogenic Systems

# Cryogenic Air Separation

- Consider an idealized 2 component system (e.g.  $N_2$  &  $O_2$ )
- As the bubbles move through the liquid, the higher boiling point component condenses out
- Thus, the bubbles become progressively richer in the lower boiling point component ( $N_2$ ) while the liquid flowing downwards becomes richer in the higher boiling point component ( $O_2$ )



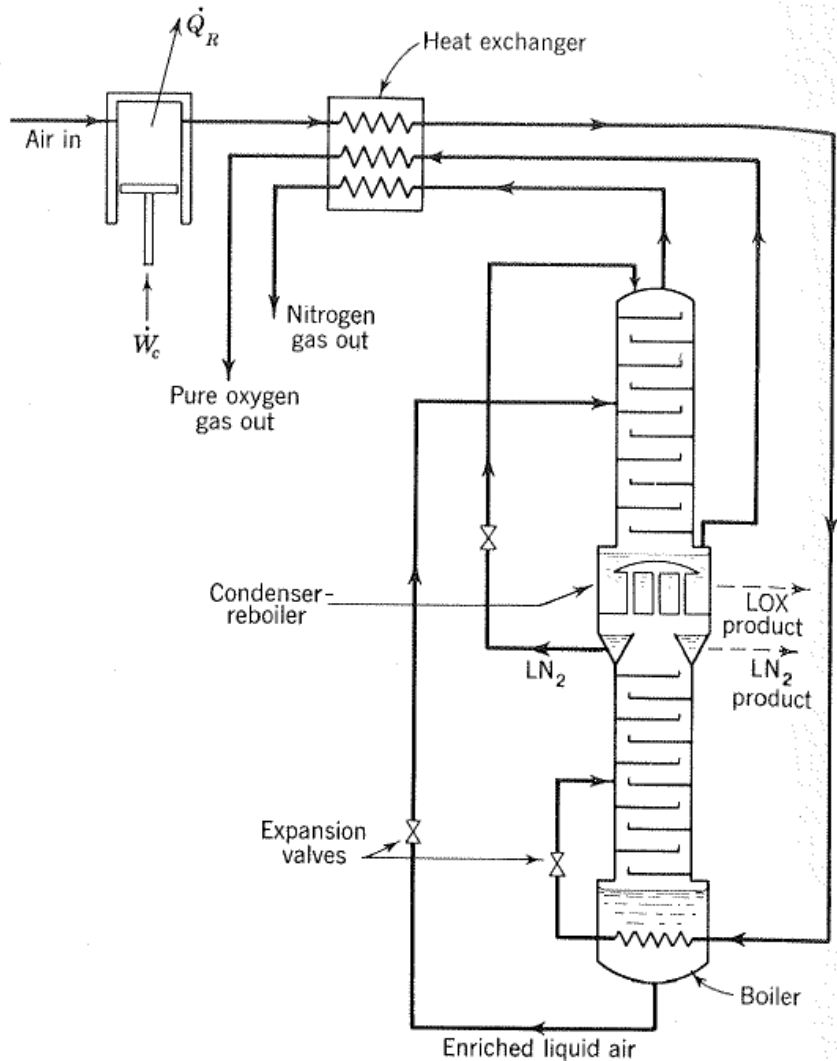
From Barron  
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Systems



# “Real World” Examples of Air Separation via Rectification

- Still Idealized – there is much proprietary detail
- However, note that basic cryogenic engineering techniques still apply:
  - Compression
  - Isenthalpic (JT) expansion
  - Isentropic expansion
  - Appropriate use of heat exchangers between flows of different temperatures
  - Clever use of different boiling points as a function of pressure
  - Everything is in cryostats and connected via transfer lines
- Which components are recovered and in what form depends on the customer specific economics and applications
- All examples come from Barron’s Cryogenic Systems

# Linde Double Column System

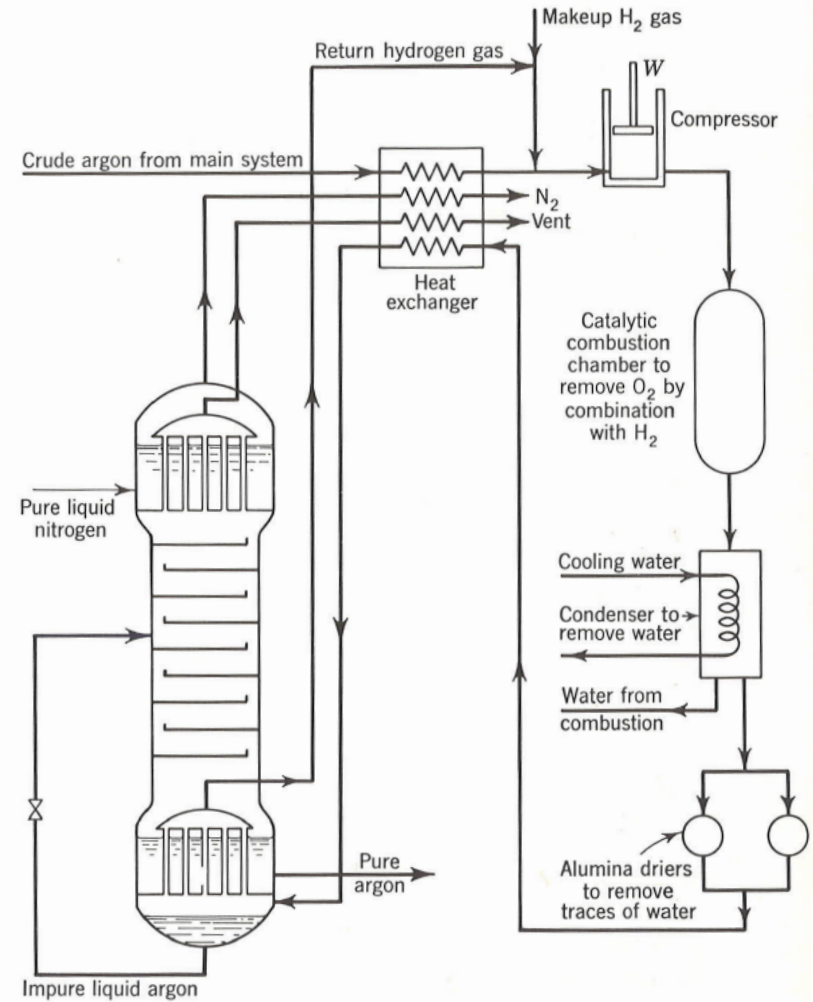
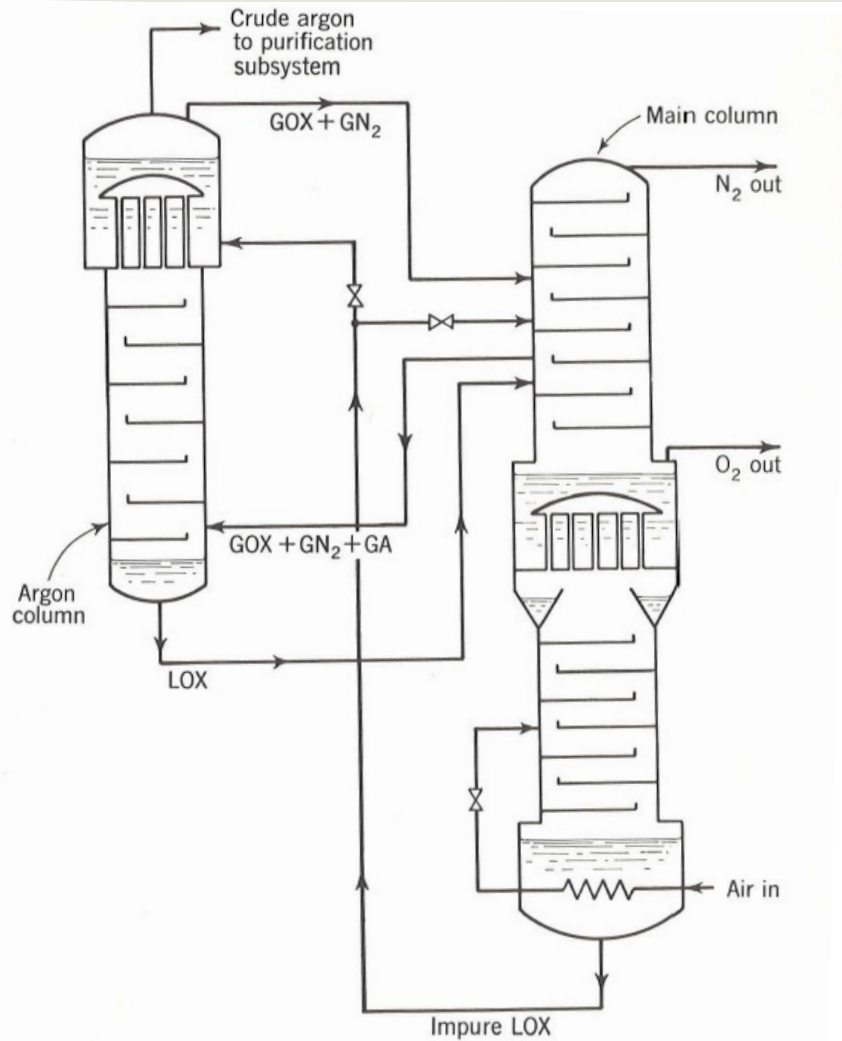


Note:

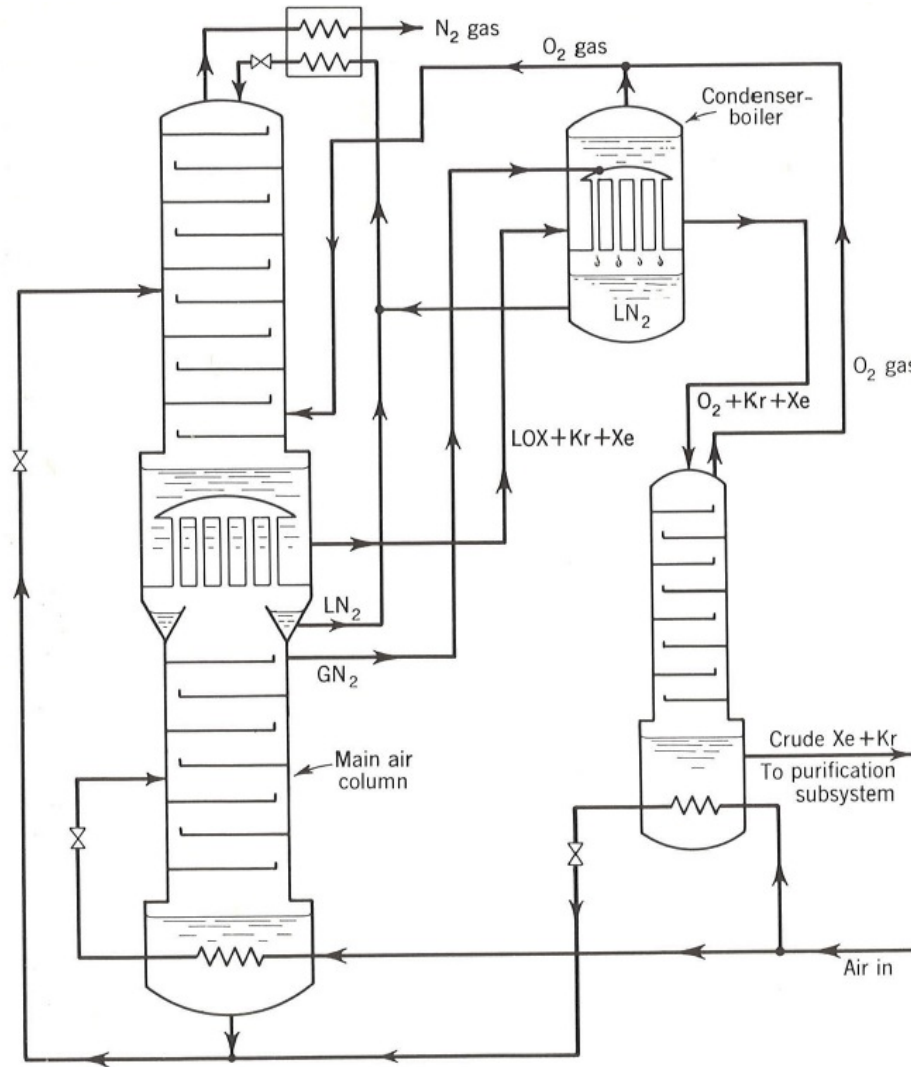
- 1) Use of JT expansion & HXs
- 2) Lower half of column operates at 5 Atm. while upper half operates 1 Atm.
- 3) The boiling point of  $\text{N}_2$  @ 5 ATM is higher than that of  $\text{O}_2$  at 1Atm
- 4) The warmer  $\text{N}_2$  vapor serves to boil off the  $\text{O}_2$  and is condensed by this process



# Argon Separation & Purification



# Xenon & Krypton Recovery





# A Large Scale Example



Linde Air  
Separation  
plant for  
N<sub>2</sub> for  
enhanced  
oil  
recovery

17,500 t/d  
N<sub>2</sub>

Built for  
Pemex  
( Mexico)

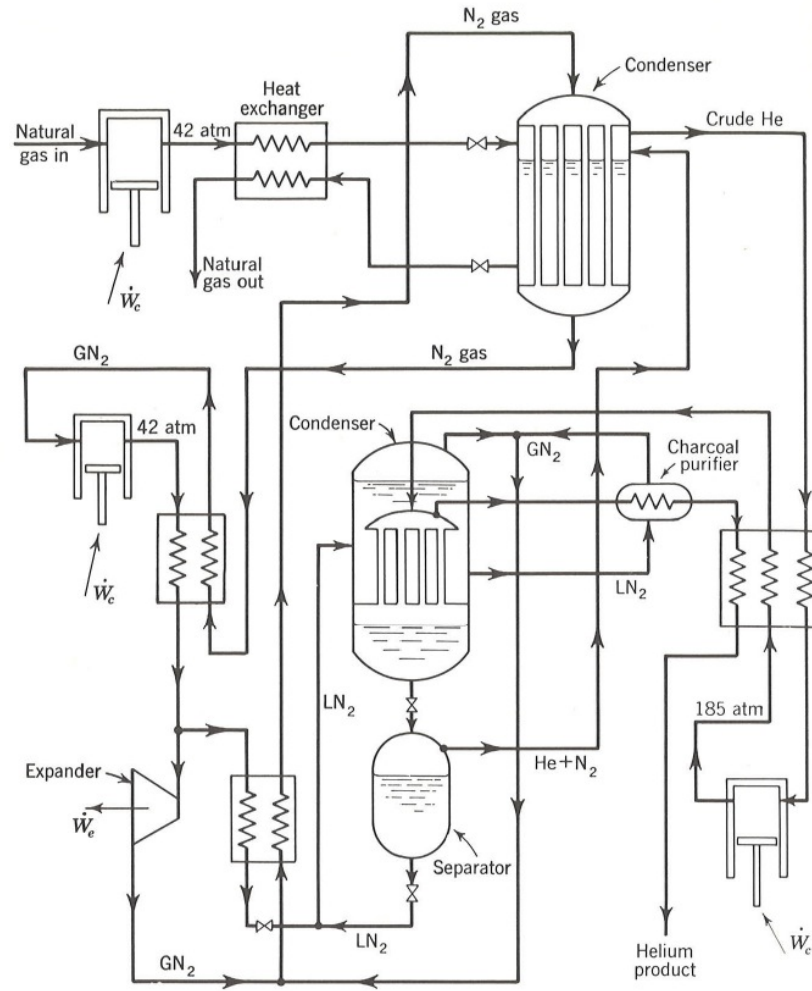
Courtesy  
Linde



# Helium Supply

- He was first discovered in the solar spectrum only later on Earth
- Almost all He on earth is a result of Alpha decay from radioactive elements in the Earth's crust
- Helium is recovered as a byproduct of some but not all Natural gas wells
- Helium was considered a strategic element and a US federal reserve was established in Oklahoma
- Today the supply of He gas is quite volatile
  - US Federal He Reserve is being sold off
  - New sources of He gas are being developed in the Algeria, Qatar, Australia and Russia among other places.
  - Rapid increase in natural gas exploration and production may open up new supplies

# Helium Recovery from Natural Gas (US Bureau Of Mines Process)



From Barron  
Cryogenic  
Systems

- World wide the estimate is that we have 100 years of He left in the ground
- Currently, the US is estimated to become a net importer of He gas in 10 – 15 years
- Prices of He gas have gone up
  - » MSU paid \$9.87 per 100 SCF in 2005 and paid \$18.59 per 100 SCF in 2012
- Some spot shortages have occurred for complicated reasons
- How will this impact cryogenics? (~ 28 % of all He is used in cryogenics)
  - Larger investment in both recovery and storage systems (with power backups)
  - Increased motivation for low loss or Zero Boil Off cryostats
  - Increased use of cryogen free systems using small cryocoolers
- $^3\text{He}$  has other complications – rarer, lack of production and use in Nuclear Security applications