#### Lecture 18 Helium II

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- Describe the properties of He II including:
  - Nature of He II
  - Two fluid model
  - Quantized vortices
  - Heat transfer
    - » Internal convection
    - » Mutual friction
    - » Heat Transfer limits
    - » Kapitza conductance
  - Fluid mechanics
  - Fountain Pumps and porous plugs
  - He II refrigerators
  - He II Heat Exchangers



### He II Superfluid Helium



- 2<sup>nd</sup> liquid phase of helium (hence He II)
- Phase transition is  $2^{nd}$  order (no latent heat) but there is a discontinuity in the specific heat ( $\lambda$  transition)
- T<sub>λmax</sub> = 2.2 K
- Has unique thermal and fluid properties
  - High effective thermal conductivity
  - Zero viscosity under certain conditions



#### **Recall the Helium Phase Diagram**







# **Applications of He II**



- Lower temperature results in:
  - Higher field s/c magnets (Tore Supra, LHC, high field labs in USA, Japan and Europe)
  - Lower BCS losses in SCRF (CEBAF, SNS, ILC, FRIB)
  - Lower background temperatures for IR astronomy (COBE, IRAS, Spitzer)
- Fundamental studies of turbulence
- Superfluid "wind tunnels" (very large Re#)



### **Applications of He II**







# What is He II ?



- A Bose Einstein Condensate
  - A fraction of atoms in He II have condensed to the quantum ground state
  - He II was the first BE condensate discovered
  - The only one that has significant industrial applications
- The properties of He II can be understood via the two fluid model



### **Two Fluid Model**





Relative Densities of Superfluid and Normal fluid components (From <u>Helium Cryogenics</u> – Van Sciver)



#### Quantized Vortices (or does He II at 1 K rotate in a bucket)



- At 1 K He II is almost entirely the superfluid component and thus has almost 0 viscosity. This would imply that He at 1 K in a spinning bucket wouldn't rotate but it does. What's the answer?
  - The vortices are quantized:

$$C = \int V_s \cdot dl = n \frac{h}{m}$$

- Solves rotating bucket problem
  - In the body of the fluid:  $\nabla^2 V_s = 0$
  - At the wall:  $\nabla^2 V_{s} \neq 0$
- This has been experimentally observed
- The quantized vortices in the superfluid component are an important part of heat transfer mechanism in He II



#### Direct Observation of Quantized Vortices via Electron Trapping



Fig. 4.26. Photographic reproduction of vortex line array in rotating He II (from Yarmchuk and Packard<sup>12</sup>): (a) through (1) indicate increasing angular frequency. EUROPEAN SPALLATION

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# Heat Transfer in He II



The basic mechanism is internal convection:



- No net mass flow
- Note that this is not conduction or classical convection but an entirely different heat transfer mechanism
- This can be extremely efficient (more than 1000x better than conduction through copper)



### Heat Transfer in He II



There are 2 heat transfer regimes:
V<sub>s</sub> < V<sub>sc</sub>

$$q = \frac{\left(\rho s d^2\right)T}{\beta \eta_n} \frac{dT}{dx}$$

•  $V_s > V_{sc}$ 

» Mutual Friction Regime (quantized vortices interact with the viscosity of the normal component

$$q = \left[ f^{-1}(P,T) \frac{dT}{dx} \right]^{\frac{1}{3}}$$

- As V  $_{sc}$  ~ d<sup>1/4</sup> (cgs units) the mutual friction regime is most applicable in engineering applications of He II



### **Heat Conductivity Function**



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# He II Heat Transfer Limits



- $\bullet$  In pressurized He II: T  $_h~$  must be less than T  $_\lambda~$
- Thus the peak heat flux q\* is:

$$q^*L^{\frac{1}{3}} = \left(\int_{T_b}^{T_\lambda} f^{-1}(T)dt\right)^{\frac{1}{3}}$$

• At 1.9 K and 1 bar :

 $q^{L^{1/3}} \sim 15 \text{ kW/m}^{5/3}$ 



#### Peak Heat Flux (q\*) in Pressurized He II







### Limits on He II Heat Transfer



- In saturated He II, the limit is given by the local saturation temperature & the degree of local subcooling
- In the ILC cavity He vessel this works out to about 1 W/cm<sup>2</sup> or ~ 30 W total through the connection tube
  - More heat than that would require a redesign
- Exceeding the heat transfer limits in either the saturated or pressurized case results in conversion to He I and boiling at the heated surface





# **Surface Heat Transfer**



- Heat transfer from a surface into He II is completely dominated by a fundamental inefficiency in moving energy from the surface to the fluid
- This effect exists but is not important in standard convection problems
  - Normally we assume  $T_w = T_{fw}$  but this is not true in the case of He II
- This surface heat transfer effect is described by Kapitza Conductance

• For q < 1 kW/m<sup>2</sup> 
$$q = h_K \Delta T_S$$

• For q > 1 kW/m<sup>2</sup> 
$$q = a(T_s^m - T_b^m)$$

h k, a and m are empirical and dependent on material, temperature and surface condition



### **Surface Heat Transfer**



- m ~ 3
- Kapitza conductance is <u>not</u> dependent on helium flow rate

Kapitza Conductance for Copper (<u>Helium Cryogenics</u>, Van Sciver)





# Forced Convection and He II



- If Kapitza Conductance is independent of flow rate does forced convection in He II make any sense?
  - Yes! Forced convection has the effect of reducing the maximum temperature in the He II and thus allowing more heat to be transferred before reaching the peak heat flux





# He II Fluid Dynamics



- Despite the presence of the superfluid component, in almost all engineering applications He II behaves as a classical fluid. This includes :
  - Pump performance
    - » Except cavitation in saturated He II
  - Pressure drop in tubes, valves, bellows and fittings
  - Flow metering techniques
- This is likely a result of the quantized vortices in the superfluid component being coupled via mutual friction to the normal fluid viscosity
- However, keep in mind that the unique heat transfer properties still exist as described.



# **He II Fluid Dynamics**



- He II does behave differently in cases of:
  - Film flow
  - Porous plugs
  - Hot wire anemometers
    - » Since Kapitza conductance is independent of mass flow rate, HWA will not work in He II
  - Two phase flow (liquid/vapor) due to the large density difference between liquid and vapor in the case of He II



# **Fountain Pumps**



- Unique to He II
- Allows pumping with no moving parts
- Superfluid component can pass through the porous plug while normal fluid component can't
- $\Delta P = \rho S \Delta T$





### **Fountain Pumps**





**Figure 10-21** The fountain effect pump developed for the SHOOT flight demonstration. It is an aluminasilica composite ceramic with a 0.4- $\mu$ m effective pore size. This pump demonstrated a flow rate of 30 g/s in flight (from Ref. [43]).



### **Porous Plugs for Phase Separation**



- There is no gravity driven stratification how do we separate vapor from liquid?
- The use of He II with porous plugs provides a solution
- Build the plug with pores large enough to admit both the normal fluid and superfluid components
- Evaporation at the vent end causes cooling. The superfluid component is driven back to the dewar and only vapor escapes
- If the plugs are properly sized the helium in the dewar will stay at the correct temperature



Fig.1 Schematic diagram of a He II vapor-liquid phase separator

Yu et al. J. Thermal Science Vol. 14, No. 1 (2005)

SPALLATION



#### **Typical He II Refrigeration System**



*He II (Superfluid Helium)* S. W. Van Sciver, in <u>Handbook of Cryogenic Engineering</u>,





### He II Heat Exchangers



- Must take into account the unique He II heat transfer properties
- Must allow for rapidly changing specific heat with temperature
- Handbook methods e.g.  $\epsilon$  NTU are not suitable



# Summary



- He II is a unique fluid that displays quantum behavior on a macroscopic scale
- He II has significant applications in large scale cryogenics for scientific research
- Despite its unique properties, the use of He II in industrial scale engineering applications is well understood and significant experience exists: Tore Supra, LHC, Jlab, NASA
- This lecture just hit the high points and many other He II topics remain including :
  - Film flow
  - Second sound
  - Detailed investigations of heat and mass transfer
- There is a large amount of information in the literature