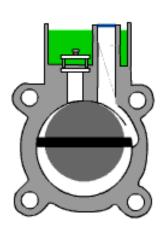
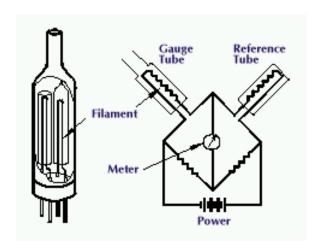
Vacuum equipment

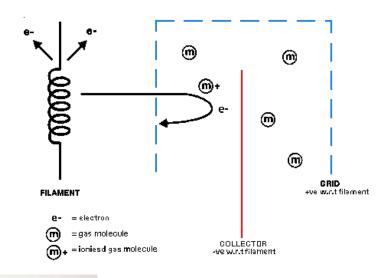


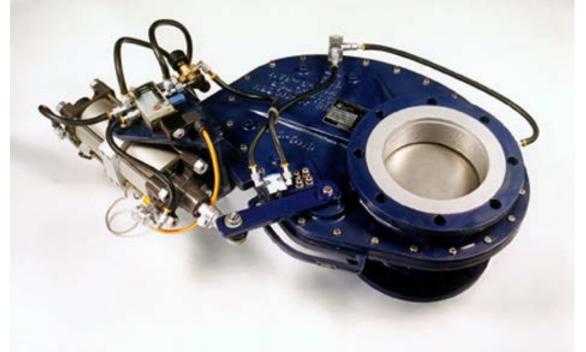




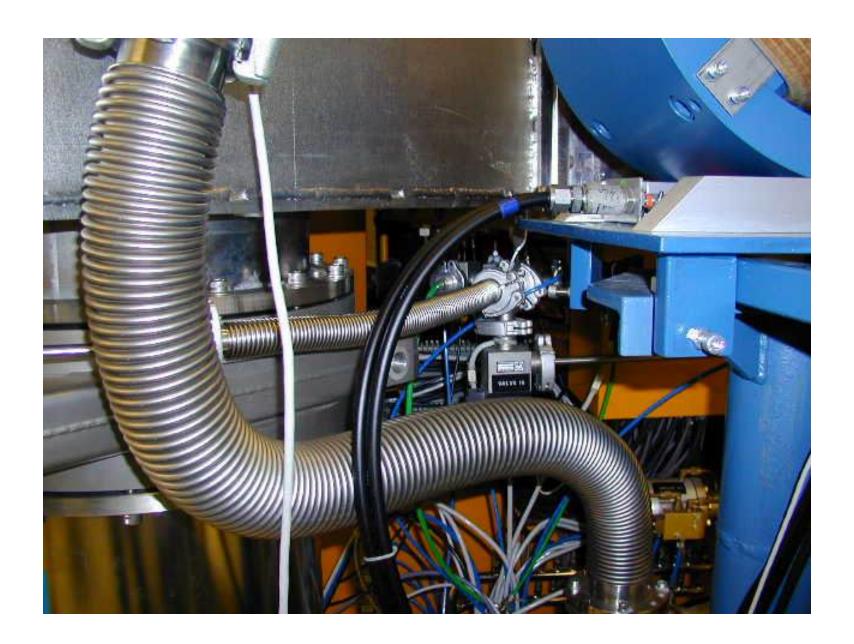








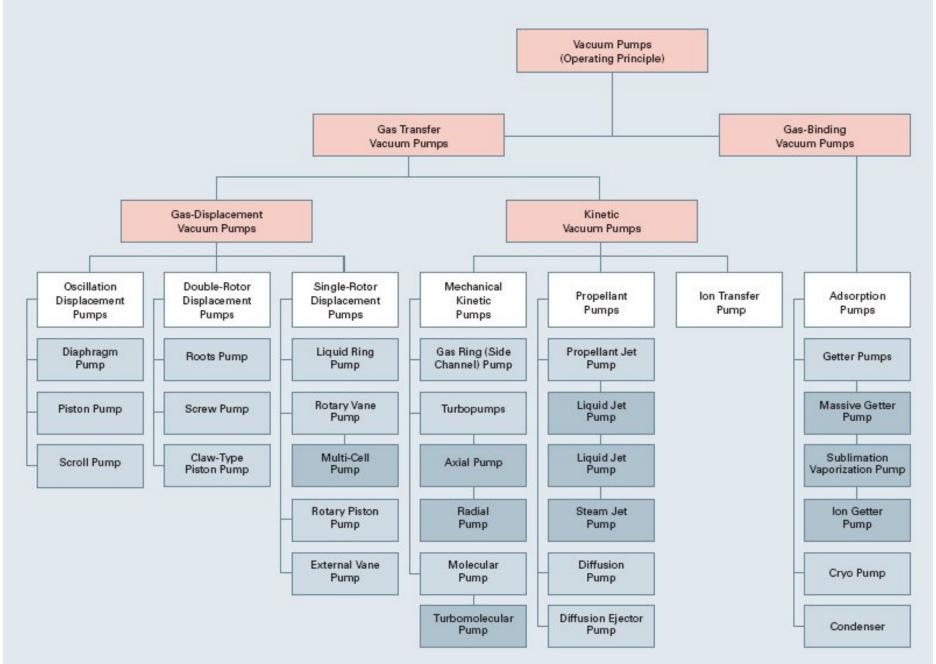






Pumps

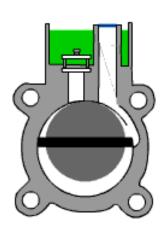
- Rotary vane pumps
- Rotary piston pumps
- Roots pumps
- Scroll pumps
- Turbo molecular pumps
- Diffusion pumps
- Getter pumps
 - Sublimation pumps
 - Non-evaporable getter pumps
- Ion pumps
 - Electrostatic pumps
 - Sputter-Ion pumps
- Cryogenic pumps



Courtesy of Pfeiffer

Rotary vane pumps

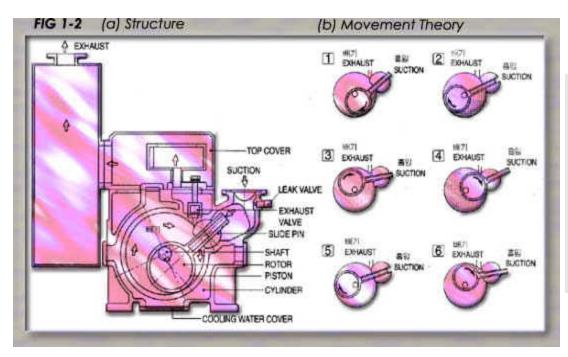
- Pressure range 1 to 10⁵ Pa
- 10 to 200 m^3/h
 - Rough pumping
 - Backing for diffusion and turbo pumps

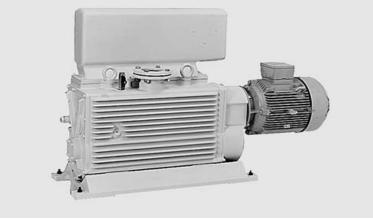




Rotary piston pumps

- Roughing pumps
- In combination with Roots
- 30 to $1500 \text{ m}^3/\text{h}$
- Rugged and simple

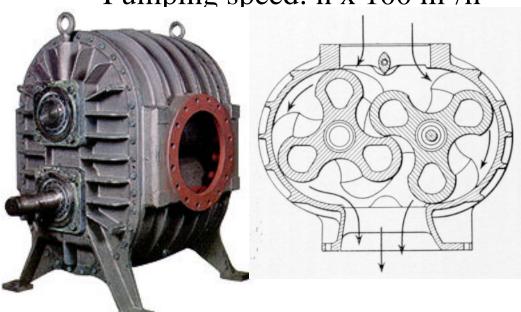




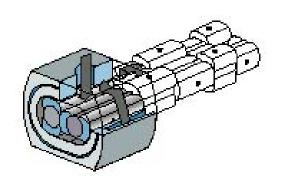
Roots pumps

- Positive displacement blower
- 3000 to 3500 rpm
- Two lobed rotors on parallel shafts

• Pumping speed: n x 100 m³/h







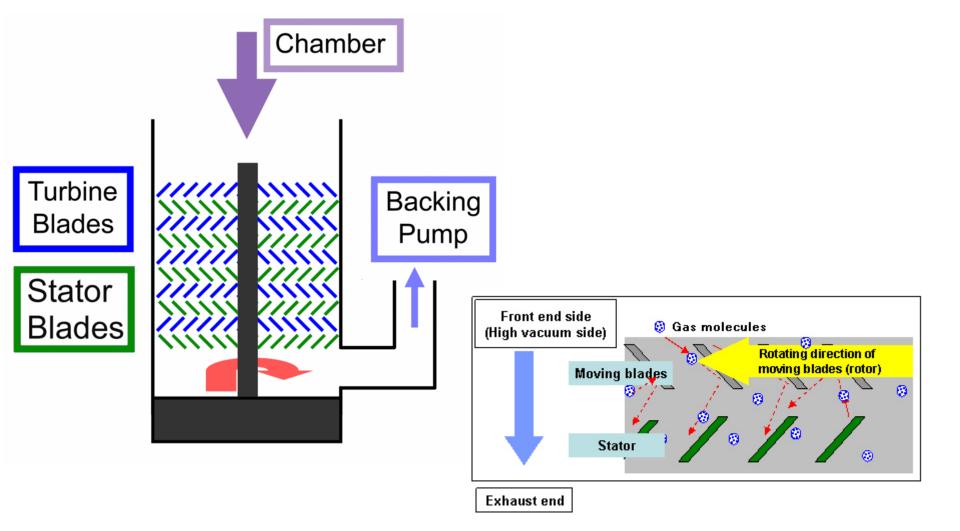
Scroll pump

- Oil free (dry)
- Tens of m^3/h
- Down to 10⁻² mbar



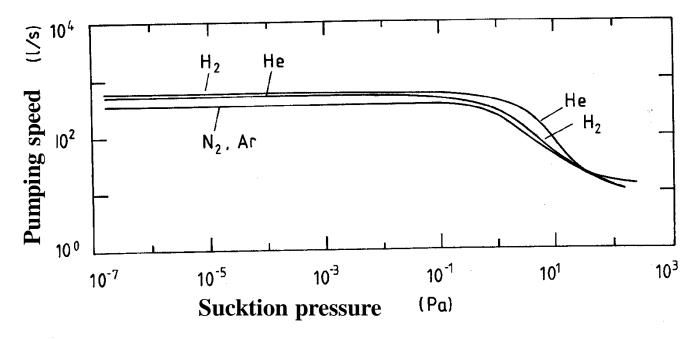
Turbo pumps

- High speed turbine
- Down to 10⁻⁸ Pa
- Vane speed higher than molecular speed
 - -100 500 m/s
 - Momentum transfer





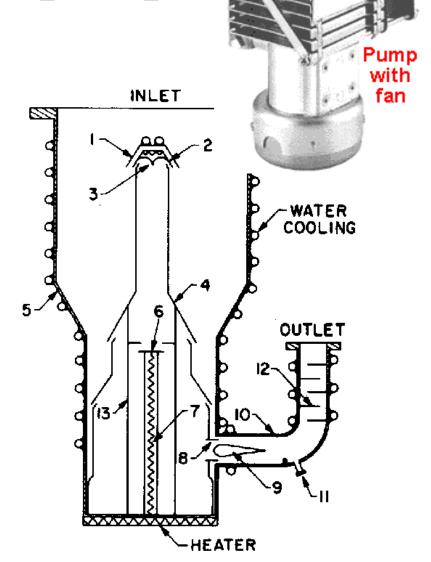


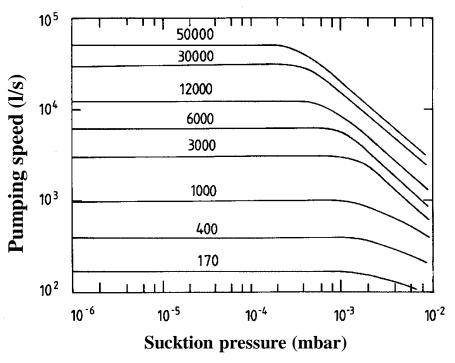


Pumping speed of a turbopump for different gases as a function of inlet pressure

Diffusion pumps

- Vapor jet
 - Momentum transfer on oil collision with the vapor stream
- Ultimate pressure 10⁻¹⁰ Pa





Pumping curves for a diffusion pump as a function of inlet pressure. Pumping speed for dry air has been marked for the curves

Getter pumps

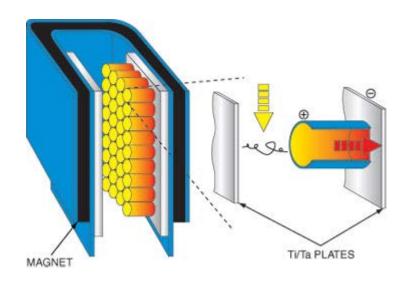
- Gas molecules collide and react with the surface
- Sublimation pump is a surface getter pump for active gases
 - Titanium: inexpensive, effective, easily sublimed
- Non-evaporable getter
 - Pump by surface adsorption followed by bulk diffusion
 - High temperatures



The first titanium sublimation getter pump has been assembled and is being tested in the PBFA-II Integrated Test Facility

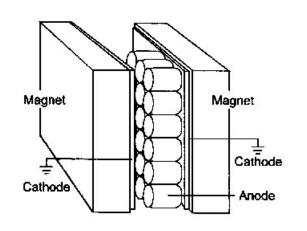
Ion pumps

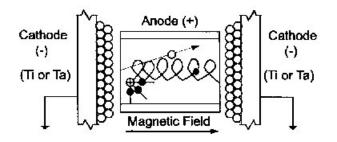
- Ions are pumped easily because they are more reactive with surfaces than neutral molecules
 - Bombarding electrons





- The pump is started by applying high voltage between the tube shaped anode and the cathode of the ion pump.
- Electrons are accelerated toward the positive anode and are forced to follow a spiral path in the tube shaped anode because of the magnetic field.
 - This has the effect of sweeping out more space and increasing the probability that an electron will collide with a gas molecule.
- The positive ions that are formed in the collisions strike the chemically active titanium cathode "getter" plate.
- The ions combine with the cathode material and eject more cathode material which ends up on the surface of the anode.
 - This constantly replenishes the film of chemically active cathode material on the anodes which combines with active gas molecules and effectively pumps them from the system. This process of removing **chemically active gasses** such as Nitrogen, Oxygen, and Hydrogen is called "gettering".





Inert gasses are handled a little differently. They are buried in the pump surfaces. This happens when they are ionized and hurled into the cathode. They penetrate a few layers and bury themselves in the cathode lattice structure. They can be reemitted when other ions strike the surface so they tend to collect where there is little of this "sputtering" going on.

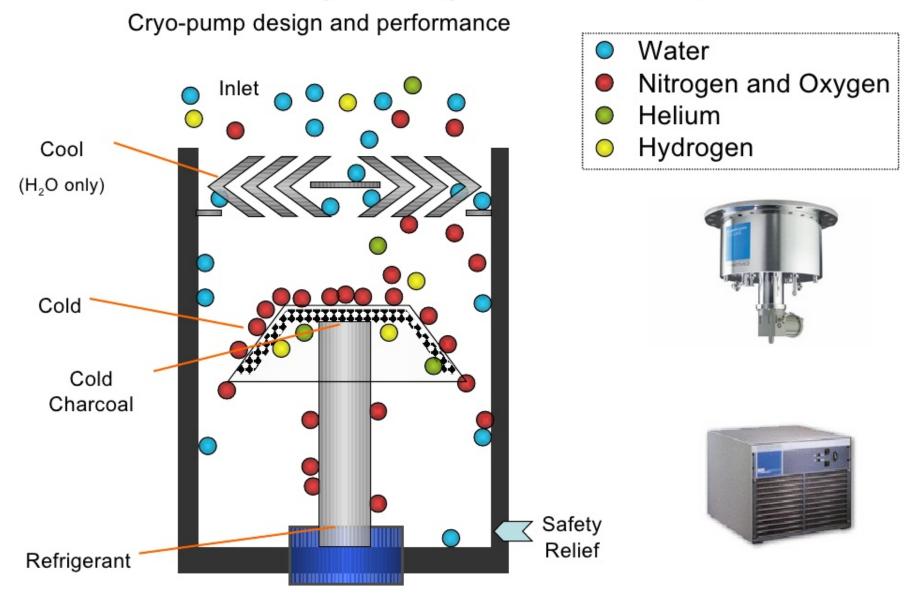
Cryogenic pumps

- Entrainment of molecules on a cooled surface by weak van der Waals or dispersion forces
- In principle, any gas can be pumped, provided that the temperature is low enough
- Cryocondensation, cryo sorption and cryotrapping
- Down to 10⁻⁹ mbar





How do we get the gas out of the space?

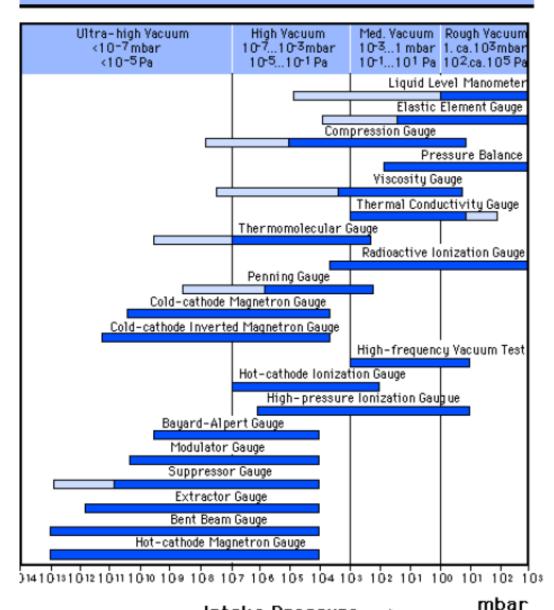


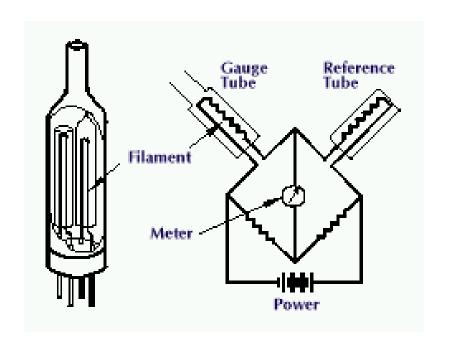
Cryopumps usually consist of two internal stages

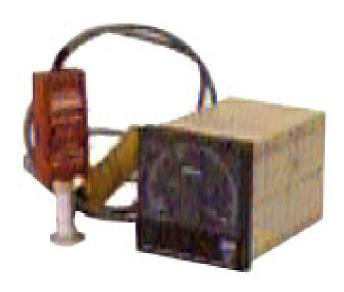
- The inlet array
 - 65 -100 K.
 - pumps or captures water vapor
- The second stage consists of a series of metal pumping surfaces
 - 10 20 K
 - pumps gases such as nitrogen and argon
- The metal pumping surfaces are partially covered with charcoal granules
 - Gases such as hydrogen and helium, which cannot be frozen at typical second-stage temperatures, are adsorbed by the charcoal granules and thereby removed from the vacuum chamber



Working Pressure Ranges of Vacuum Gauges

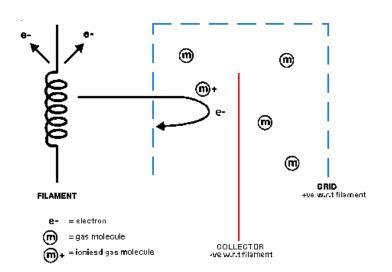






Pirani

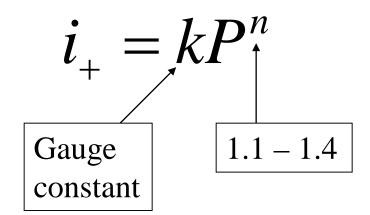
- •Thermal conductivity gauge
- •Heated wire forms one arm of a Wheatsone bridge
- •Down to 10⁻³ Pa with a compensating tube

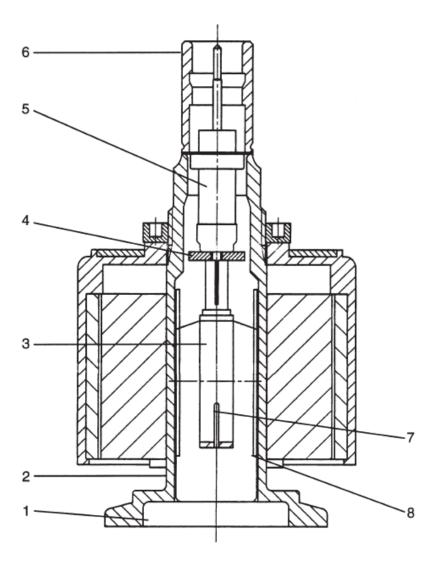




Penning

- •Cold cathode tube
- •Magnetic field guides electrons: more effective ionization of residual gas
- •Down to 10-8 Pa
- •Pressure by measuring current



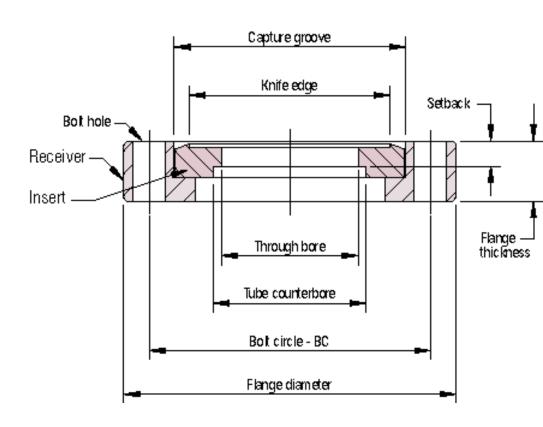


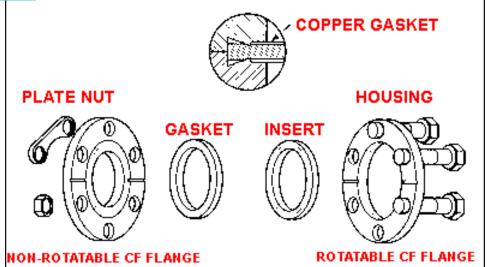
- 1 Small flange DN 25 KF; DN 40 KF
- 2 Housing
- 3 Ring anode with ignition pin
- 4 Ceramic washer

- 5 Current leadthrough
- 6 Connecting bush
- 7 Anode pin
- 3 Cathode plate

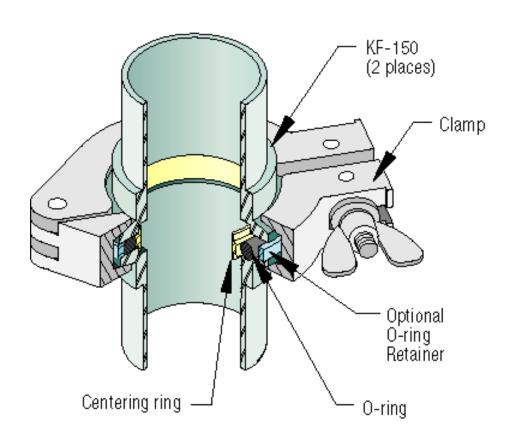








KF





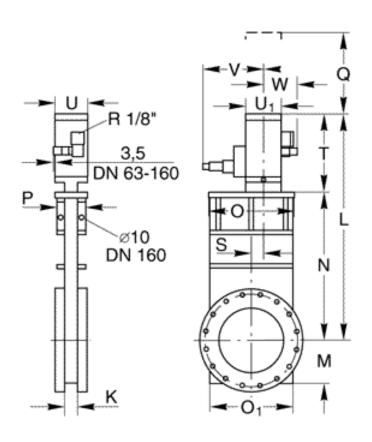
ISO-K







ISO-F



Valves



